



NATIONAL BOARD FOR TECHNICAL EDUCATION

BASIC SCIENCE (GENERAL)

CURRICULUM AND COURSE SPECIFICATIONS

FOR

NATIONAL DIPLOMA (ND) AND HIGHER NATIONAL DIPLOMA (HND) PROGRAMMES

MAY, 2025

PLOT B, BIDA ROAD, P.M.B. 2239 KADUNA – NIGERIA

FOREWARD

The reviewed **Basic Science Courses**, encompassing **Physics, Chemistry, and Biology**, designed for the National Diploma (ND) and Higher National Diploma (HND) Programmes represents the Board's recognition of the role of foundational scientific knowledge in nurturing competent and adaptable professionals for Nigeria's industries.

This review, the first since the curriculum's inception in 1999, was necessitated by the dynamic shifts in Technical and Vocational Education (TVE) and the imperative to meet the demands of the 21st century. This aims to ensure that our diplomates are not only equipped with specialized technical skills but also possess a robust scientific understanding that underpins innovation and problem-solving.

These reviewed Basic Science Courses is designed to instill learning, critical thinking and problem-solving abilities firmly grounded in scientific principles and promote an understanding of natural phenomena.

It is our profound hope that the effective implementation of this revised Basic Sciences Courses will empower our diplomates to apply their knowledge with confidence and contribute meaningfully to various sectors of the economy

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EXECUTIVE SECRETARY NBTE, KADUNA

Introduction

The Basic Sciences Courses (General) are a foundational component of various National Diploma (ND) and Higher National Diploma (HND) Programmes, have remained largely unchanged since their development in 1999. A comprehensive review of these Basic Science courses was carried out to address the evolving needs of Technical and Vocational Education (TVE).

This aims to instill knowledge and skills beyond technical specialization, fostering critical thinking and problem-solving abilities grounded in scientific principles, and promoting a understanding of natural phenomena.

Specifically, the reviewed Basic Sciences Courses (Physics, Chemistry and Biology) are designed to equip students with:

- Improved critical thinking and problem-solving skills, cultivated through scientific methodologies.
- A foundational knowledge of fundamental scientific concepts across Physics, Chemistry, and Biology.
- Essential skills like numeracy and scientific literacy.

These integral elements of the Basic Sciences courses will produce diplomates who are not only technically proficient but also scientifically literate and capable of applying their knowledge to contribute meaningfully to society and various industries.

Lecturer Qualifications

Basic Sciences courses should primarily be taught by qualified lecturers from the Science Laboratory Technology (SLT) Department. In instances where qualified lecturers are unavailable within the SLT Department, the Department may employ the services of a qualified lecturer from another department within the same institution or engage one on a part-time basis.

Teaching Staff Requirements

Lecturers teaching SLT courses should possess relevant qualifications and experience, with a minimum of a Bachelor's degree in a relevant field.

- **ND Courses:** Should be taught by lecturers of Lecturer III rank and above.
- **HND Courses:** Should be taught by lecturers of Lecturer II rank and above.

NATIONAL BOARD FOR TECHNICAL EDUCATION

CURRICULUM TABLE

S/N	COURSE CODE	COURSE TITLE	L	P	CH	CU
ND PHYSICS, CHEMISTRY AND BIOLOGY COURSES						
1	STP 101	GENERAL PHYSICS I	2	1	3	3
2	STC 101	GENERAL CHEMISTRY I	2	1	3	3
3	STB 101	GENERAL BIOLOGY I	2	1	3	3

S/N	COURSE CODE	COURSE TITLE	L	P	CH	CU
HND PHYSICS, CHEMISTRY AND BIOLOGY COURSES						
1	STP 301	GENERAL PHYSICS II	2	1	3	3
2	STC 301	GENERAL CHEMISTRY II	2	1	3	3
3	STB 301	GENERAL BIOLOGY II	2	1	3	4

NATIONAL BOARD FOR TECHNICAL EDUCATION

NATIONAL DIPLOMA

PROGRAMME: NATIONAL DIPLOMA		
COURSE TITLE: General Physics I	COURSE CODE: STP 101	CONTACT HOURS: 3
	CREDIT UNITS: 3	THEORETICAL: 2
YEAR: I SEMESTER: 0	PRE-REQUISITE:	PRACTICAL: 1
GOAL: The course is designed to acquaint students with the knowledge and skills of applications of Physics in everyday life.		
GENERAL OBJECTIVES: On completion of this course, the students should be able to: 1.0 Know the historical development of physics as a science 2.0 Know the concept of motion 3.0 Know the principle of operation of simple machines 4.0 Know the concept of energy 5.0 Know the concept of waves 6.0 Know the concept of electric and magnetic fields. 7.0 Understand the concept of thermodynamics		

PROGRAMME: NATIONAL DIPLOMA						
COURSE TITLE: General Physics I			COURSE CODE: STP 101		CONTACT HOURS: 3	
			CREDIT UNITS: 3		THEORETICAL: 2	
YEAR: I SEMESTER: 0			PRE-REQUISITE:		PRACTICAL: 1	
COURSE SPECIFICATION: THEORETICAL AND PRACTICAL						
GOAL: The course is designed to acquaint students with the knowledge and skills of applications of Physics in everyday life.						
GENERAL OBJECTIVE 1.0: Know the historical developments of Physics as a science						
THEORETICAL CONTENT				PRACTICAL CONTENT		
WEEK	SPECIFIC LEARNING OUTCOME	TEACHER'S ACTIVITIES	RESOURCES	SPECIFIC LEARNING OUTCOME	TEACHER'S ACTIVITIES	RESOURCES
1-2	1.1 Outline the development of scientific observations and knowledge as science e.g.: <ul style="list-style-type: none">• Making of fire• Free fall• Movement of stars• Law of flotation• Motion of projectile (warfare)• Burning glass 1.2 Outline the growth of Physics into different areas such as:	Explain the development of scientific observations and knowledge as science e.g.: <ul style="list-style-type: none">• Making of fire• Free fall• Movement of stars• Law of flotation• Motion of projectile (warfare)• Burning glass Explain the growth of Physics into different areas such as:	Textbooks Journals Whiteboard Marker Computer Internet Projector	Perform simple activities to demonstrate scientific principles: <ul style="list-style-type: none">• Falling from a height• Floating of objects in water.	Guide student to: Perform simple activities to demonstrate scientific principles: <ul style="list-style-type: none">• Falling from a height• Floating of objects in water.	Floating objects Free-falling objects

	<ul style="list-style-type: none"> • Optics • Mechanics • Electricity • Magnetisms • Heat etc. 	<ul style="list-style-type: none"> • Optics • Mechanics • Electricity • Magnetisms • Heat etc. 				
1.3	Outline the contributions of named scientists in the development of physics as a science. <ul style="list-style-type: none"> • Early man • Aristotle • Galileo • Archimedes • Isaac Newton • Einstein etc. 	Explain the contributions of named scientists in the development of physics as a science: <ul style="list-style-type: none"> • Early man • Aristotle • Galileo • Archimedes • Isaac Newton • Einstein etc. 				
1.4	Explain Physics as a subject	Explain Physics as a subject				
1.5	Distinguish between early observations mentioned in 1.1 above to modern applications of physics	Distinguish between early observations mentioned in 1.1 above to modern applications of physics				
1.6	Enumerate some	Explain some modern				

	modern applications of physics in their everyday life	applications of physics in their everyday life				
	1.7 List the industries and firms that require the services of physicists	Explain the industries and firms that require the services of physicists				
GENERAL OBJECTIVE 2.0: Know the concept of motion						
3-4	<p>2.1 Define Motion</p> <p>2.2 Explain the types of motion e.g.</p> <ul style="list-style-type: none"> • Translational • Circular • Rational • Oscillatory etc. <p>2.3 List the application of each type of motion mentioned in 2.2 above</p> <p>2.4 Explain the cause of motion</p> <p>2.5 Define the following terms giving practical examples of each one of</p>	<p>Explain motion</p> <p>Explain the types of motion e.g.:</p> <ul style="list-style-type: none"> • Translational • Circular • Rational • Oscillatory etc. <p>Explain the application of each type of motion mentioned in 2.2 above</p> <p>Explain the cause of motion</p> <p>Explain the following terms giving practical examples of each one of</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>	<p>Carryout experiments to illustrate the following:</p> <ul style="list-style-type: none"> • Circular motion • Rational motion • S.H.M <p>Demonstrate the methods of measurement of length time-using stop watch as in the application of simple pendulum and mass</p>	<p>Guide student to:</p> <p>Carryout experiments to illustrate the following:</p> <ul style="list-style-type: none"> • Circular motion • Rational motion • S.H.M <p>Demonstrate the methods of measurement of length time-using stop watch as in the application of simple pendulum and mass</p>	<p>Stop-watch</p> <p>Beam balances</p> <p>Vernier callipers</p> <p>Meter ruler</p> <p>Simple pendulum</p>

	them: <ul style="list-style-type: none"> • Speed • Acceleration • Velocity • Distance • Displacement 	them: <ul style="list-style-type: none"> • Speed • Acceleration • Velocity • Distance • Displacement 		Perform experiments in the measurement of mass using beam balances, length using vernier callipers and meter-ruler; time, using stop watch, simple pendulum.	Perform experiments in the measurement of mass using beam balances, length using vernier callipers and meter-ruler; time, using stop watch, simple pendulum.	
GENERAL OBJECTIVE 3.0: Know the principle of operation of simple machines						
5-6	3.1 Define a machine as a device which facilitates work 3.2 List simple machines such as: <ul style="list-style-type: none"> • Levers • Pulleys • Inclined planes • Hydraulic press etc. 	Explain machine as a device which facilitates work Explain simple machines such as: <ul style="list-style-type: none"> • Levers • Pulleys • Inclined planes • Hydraulic press etc. 	Textbooks Journals Whiteboard Marker Computer Internet Projector	Perform simple experiments to illustrate the principles of : <ul style="list-style-type: none"> • Operation of levers • Inclined planes • Pulleys • Hydraulic press 	Guide student to: Perform simple experiments to illustrate the principles of : <ul style="list-style-type: none"> • Operation of levers • Inclined planes • Pulleys • Hydraulic press 	Lever Pulleys Inclined planes Hydraulic press
	3.3 Cite instances of the applications of each type of machine in 3.2 above in their everyday life	Explain instances of the applications of each type of machine in 3.2 above in their everyday life				

	<p>3.4 Explain complex machines as a combination of simple machines e.g.:</p> <ul style="list-style-type: none"> • Bicycles • Sewing machines. etc. <p>3.5 List factors affecting motion for example air resistance, viscosity, solid friction as it affects machines</p> <p>3.6 State the advantages and disadvantages of friction</p>	<p>Explain complex machines as a combination of simple machines e.g.:</p> <ul style="list-style-type: none"> • Bicycles • Sewing machines. etc. <p>Explain factors affecting motion for example air resistance, viscosity, solid friction as it affects machines</p> <p>Explain the advantages and disadvantages of friction</p>				
General Objective 4.0: Know the concept of energy						
7-8	<p>4.1 Define work, energy, power and conversion of energy, stating their units</p> <p>4.2 State the law of conversion of energy</p> <p>4.3 List different types of energy:</p> <ul style="list-style-type: none"> • Mechanical energy 	<p>Explain work, energy, power and conversion of energy, stating their units.</p> <p>Explain the law of conversion of energy</p> <p>Explain the different types of energy:</p> <ul style="list-style-type: none"> • Mechanical energy 	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>	<p>Demonstrate experimentally the effects of heat energy e.g.:</p> <ul style="list-style-type: none"> • Ball and ring experiment • Melting of ice • Boiling of water to form steam. 	<p>Guide student to:</p> <p>Demonstrate experimentally the effects of heat energy e.g.:</p> <ul style="list-style-type: none"> • Ball and ring experiment • Melting of ice • Boiling of water to form steam 	<p>Ball</p> <p>Ring</p> <p>Ice</p> <p>Water</p> <p>Car battery</p> <p>Heating coils</p>

	<ul style="list-style-type: none"> • Chemical energy • Heat energy • Electrical energy • Sound energy • Light energy • Nuclear energy 	<ul style="list-style-type: none"> • Chemical energy • Heat energy • Electrical energy • Sound energy • Light energy • Nuclear energy 		<p>Perform simple experiments to illustrate. Conversion of energy from one form to another</p>	<p>Perform simple experiments to illustrate Conversion of energy from one form to another</p>	<p>Touch Light bulbs</p>
4.4	Explain simple practical problems involving work and energy	Explain simple practical problems involving work and energy		<p>Perform experiments to illustrate conduction, convection and radiation:</p> <ul style="list-style-type: none"> • Conduction of heat through a metal rod, • Convection in air and radiation through different materials. 	<p>Perform experiments to illustrate conduction, convection and radiation:</p> <ul style="list-style-type: none"> • Conduction of heat through a metal rod, • Convection in air and radiation through different materials. 	<p>Beating drum</p>
4.5	Define temperature stating the various units of measurement	Explain temperature stating the various units of measurement				<p>Metal rod</p>
4.6	Distinguish between heat and temperature	Explain the differences between heat and temperature				<p>Heating mantel</p>
4.7	<p>State the uses of the following types of thermometers:</p> <ul style="list-style-type: none"> • Mercury in glass • Alcohol in glass • Gas-Volume Thermometer • Clinical 	<p>State the uses of the following types of thermometers:</p> <ul style="list-style-type: none"> • Mercury in glass • Alcohol in glass • Gas-Volume Thermometer 				

	Thermometer etc. • Pyrometers etc.	• Clinical Thermometer etc. • Pyrometers etc.				
	4.8 Describe specific application of the above Thermometers in everyday life	Explain the specific application of the different types of thermometers in everyday life				
General Objective 5.0: Know the concept of waves						
9-10	5.1 Define waves 5.2 Explain the differences between mechanical and electro-magnetic waves 5.3 List examples of transverse and longitudinal waves 5.4 List examples of sound waves 5.5 Explain the differences between music and noise 5.6 Classify musical	Explain waves Explain and state the differences between mechanical and electro-magnetic waves Explain examples of transverse and longitudinal waves Explain examples of sound waves Explain the differences between music and noise Classify musical	Textbooks Journals Whiteboard Marker Computer Internet Projector	Illustrate experimentally the concept of waves using examples such as: <ul style="list-style-type: none"> • Ripples on water • Waves on string • Helical string Conduct some simple experiments to demonstrate properties of waves:	Guide student to: Illustrate experimentally the concept of waves using examples such as: <ul style="list-style-type: none"> • Ripples on water • Waves on string • Helical string Conduct some simple experiment to demonstrate properties of	Water Strings Beating drums Curved mirror Glass block Prism glass

	instruments: <ul style="list-style-type: none"> • String instrument • Wind • Percussion 	instruments: <ul style="list-style-type: none"> • String instrument • Wind • Percussion 		<ul style="list-style-type: none"> • Reflection • Refraction. 	waves: <ul style="list-style-type: none"> • Reflection • Refraction. 	
5.7	Explain the term Echo and state its application in everyday life, e.g. depth finding, radar, oil prospecting etc.	Explain the term Echo and state its application in everyday life, e.g. depth finding, radar, oil prospecting etc.		Measure angles of incidence and reflection	Measure angles of incidence and reflection	
5.8	List sources of light	Explain sources of light		Conduct simple experiment to demonstrate different sound produced by different instruments.	Conduct simple experiment to demonstrate different sound produced by different instruments.	
5.9	Explain the formation of shadows, eclipse, with diagram	Explain the formation of shadows, eclipse, with diagram				
5.10	Explain the phenomenon of reflection and refraction of light	Explain the phenomenon of reflection and refraction of light		Identify the different sounds produced by string, wind and percussion	Identify the different sounds produced by string, wind and percussion	
5.11	Analyse descriptively the application of reflection and refraction of light as in: <ul style="list-style-type: none"> • Shaving mirror 	Analyse descriptively the application of reflection and refraction of light as in: <ul style="list-style-type: none"> • Shaving mirror • Driving mirror 		Perform simple experiments on reflection and refraction:	Perform simple experiments on reflection and refraction:	

	<ul style="list-style-type: none"> • Driving mirror • Reflectors • Burning glass • Magnifying glass 	<ul style="list-style-type: none"> • Reflectors • Burning glass • Magnifying glass 		<ul style="list-style-type: none"> • Plane Mirror • Curved mirror experiment, • Refraction through a glass block, • Refraction through water 	<ul style="list-style-type: none"> • Plane mirror • Curved mirror experiment • Refraction through a glass block, • Refraction through water 	
5.12	Explain the phenomenon of apparent depth and mirage	Explain the phenomenon of apparent depth and mirage				
5.13	List types of lenses – diverging and converging	Explain types of lenses – diverging and converging		Perform experiment to illustrate dispersion of white light by prism and relate to its formation of rainbow.	Perform experiment to illustrate dispersion of white light by prism and relate to its formation of rainbow.	
5.14	Explain the power of a lens	Explain the power of lens				
5.15	Describe the working of some optical instruments and briefly describe their uses: <ul style="list-style-type: none"> • Microscope • Telescope • Camera • Eye • Projector 	Explain the working of some optical instruments and briefly describe their uses: <ul style="list-style-type: none"> • Microscope • Telescope • Camera • Eye • Projector 				

	5.16 Compare the eye and the camera	Compare the eye and the camera.				
	5.17 List the component of white light and rainbow	Explain the component of white light and rainbow				
	5.18 Explain mixing of colours as in coloured television	Explain mixing of colours as in coloured television				
GENERAL OBJECTIVE 6.0: Know the concept of electric and magnetic fields						
11-12	6.1 Explain and differentiate using diagrams of the following: <ul style="list-style-type: none"> • Magnetic field • Electric field • Gravitational field 	Explain experimentally the following: <ul style="list-style-type: none"> • Magnetic field • Electric field • Gravitational field 	Textbooks Journals Whiteboard Marker Computer Internet Projector	Perform simple experiments to illustrate various effects of: <ul style="list-style-type: none"> • Electric current, • Heating effects as in heating coil, • Electric bulbs • Sound effects as in bell, loud speaker, microphone 	Guide students to: Perform simple experiments to illustrate various effects of: <ul style="list-style-type: none"> • Electric current, • Heating effects as in heating coil • Electric bulbs • Sound effects as in bell, loud speaker, microphone 	Heating coil Electric bulbs Bells Loud speaker Microphone Torch bulb Wires Fuse Cable Plug
	6.2 Explain the use of a compass	Explain the use of a compass		Perform simple experiment to show the	Perform simple experiment to	
	6.3 Illustrate the electrical wiring of a house and equipment using series and parallel connection	Explain the electrical wiring of a house and equipment using series and parallel connection				

6.4 Explain the use of fuses and plugs in an electrical wiring	Explain the use of fuses and plugs in an electrical wiring		relationship in above using torch bulb, wires and relevant instruments	show the relationship in above using torch bulb and wires and relevant instruments	
6.5 Describe precautionary measures involved in electrical wiring and use of electrical appliances in homes, industries and offices	Explain precautionary measures involved in electrical wiring and use of electrical appliances in homes, industries and offices		Insert a fuse in an electric socket	Insert a fuse in an electric socket	
6.6 List common electric instrument and their uses e.g.: • Ammeters • Voltmeters • Watt meters • Ohmmeter • Multimeter	Explain common electric instrument and their uses e.g.: • Ammeters • Voltmeters • Watt meters • Ohmmeter • Multimeter		Fix a cable into a plug	Fix a cable into a plug	
6.7 Explain the uses of electromagnets in dynamos, motors transformers, electric bell, telephone, loudspeakers	Explain the uses of electromagnets in dynamos, motors transformers, electric bell, telephone, loudspeakers.				
6.8 Define Electric power	Explain Electric power				

	6.9 Explain generation and reception of radio and television signals	Explain generation and reception of radio and television signals				
GENERAL OBJECTIVE 7.0: Understand the concept of thermodynamics						
13-15	7.1 Define thermodynamics	Explain thermodynamics	Textbooks			
	7.2 Explain thermodynamic systems	Explain thermodynamic systems	Journals			
	7.3 Explain the laws of thermodynamics	Explain the laws of thermodynamics	Whiteboard			
	7.4 Explain practical examples of the applications of thermodynamics in daily lives	Explain practical examples of the applications of thermodynamics in daily lives	Computer			
			Internet			
			Maker			
			Projector			

EVALUATION

C/A:60%

EXAMS: 40%

TOTAL: 100%

PROGRAMME: NATIONAL DIPLOMA (ND)		
COURSE TITLE: General Chemistry I	COURSE CODE: STC 101	CONTACT HOURS: 3
	CREDIT UNITS: 3	THEORETICAL: 2
YEAR: I SEMESTER: 0	PRE-REQUISITE:	PRACTICAL: 1
GOAL: This course is designed to acquaint students with the basic principles Chemistry and their applications in everyday life and society		
GENERAL OBJECTIVES: On completion of this course, the students should be able to: <ul style="list-style-type: none"> 1.0 Understand the scope and concept of science. 2.0 Understand the concept of Chemistry. 3.0 Understand the basic concepts of matter 4.0 Understand the main constituents of air. 5.0 Appreciate the importance of water in our society 6.0 Know the general characteristics and properties of acids, bases and salts. 7.0 Understand the locally available metals, their sources, locations and economic importance. 8.0 Know the nature and importance of organic compounds. 9.0 Understand the concepts of Chemical reactions. 		

PROGRAMME: NATIONAL DIPLOMA CHEMISTRY						
COURSE TITLE: General Chemistry I			COURSE CODE: STC 101		CONTACT HOURS: 3	
			CREDIT UNITS: 3		THEORETICAL: 2	
YEAR: I SEMESTER: 0			PRE-REQUISITE:		PRACTICAL: 1	
COURSE SPECIFIFCATION: THEORETICAL AND PRACTICAL						
GOAL: This course is designed to acquaint students with the basic principles Chemistry and their applications in everyday life and society						
GENERAL OBJECTIVE 1.0: Understand the scope and concept of science.						
	THEORETICAL CONTENT				PRACTICAL CONTENT	
WEEK	SPECIFIC LEARNING OUTCOME	TEACHER’S ACTIVITIES	RESOURCES	SPECIFIC LEARNING OUTCOME	TEACHER’S ACTIVITIES	RESOURCES
1-2	1.1 Define science.	Define science.	Text books Journals			
	1.2 Explain the history and philosophy of science.	Explain the history and philosophy of science.	Whiteboard Marker Computer			
	1.3 Explain the relevance of science to man and society.	Explain the relevance of science to man and society.	Internet Projector			
	1.4 Explain the difference between Basic and Applied Sciences	Explain the difference between Basic and Applied Sciences				
	1.5 List different disciplines in Basic and Applied science.	Explain different disciplines in Basic and Applied science.				

GENERAL OBJECTIVE 2.0: Understand the concept of Chemistry.

3-4	<p>2.1 Explain chemistry as a branch of science.</p> <p>2.2 Give examples of applications of chemistry to man and society e.g.:</p> <ul style="list-style-type: none">• Production of salt from hydro-chloric acid and sodium hydroxide.• Processing of crude oil into various products like gasoline, kerosene, gas, etc.• Production of soap (saponification)• Production of cement.• Production of shoe polish from charcoal• Refining of vegetable oil. <p>2.3 List local chemical industries and their products.</p>	<p>Explain chemistry as a branch of science.</p> <p>Explain examples of applications of chemistry to man and society e.g.:</p> <ul style="list-style-type: none">• Production of salt from hydro-chloric acid and sodium hydroxide.• Processing of crude oil into various products like gasoline, kerosene, gas, etc.• Production of soap (saponification)• Production of cement• Production of shoe polish from charcoal• Refining of vegetable oil. <p>Explain local chemical industries and their products.</p>	<p>Text books Journals Whiteboard Marker Computer Internet Projector</p>			
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GENERAL OBJECTIVE 3.0: Understand the basic concepts of matter

5-6	3.1 Define the three states of matter (solid, liquid and gas).	Explain the three states of matter (solid, liquid and gas).	Text books Journals Whiteboard Marker Computer Internet Projector			
	3.2 Define an atom.	Explain an atom.				
	3.3 List the three main constituents of an atom.	Explain the three main constituents of an atom				
	3.4 Define an element, a molecule, a mixture and a compound.	Explain an element, a molecule, a mixture and a compound				
	3.5 Explain the various physical methods of separating a mixture	Explain the various physical methods of separating a mixture				
	3.6 Explain the relationship between 3.2 and 3.4 above.	Explain the relationship between 3.2 and 3.4 above.				
	3.7 Give the symbols and valences of some common elements e.g. O ₂ , N ₂ H ₂ etc.	Explain the symbols and valences of some common elements e.g. O ₂ , N ₂ H ₂ etc.				

	<p>3.8 Determine the effect of heat and chemicals on different compound e.g.:</p> <ul style="list-style-type: none"> Lead nitrate Sodium sulphate(aq) + Ball(aq) NQ2, S2, O3(aq) + Ball2(aq) 	<p>Explain the effect of heat and chemicals on different compound e.g.:</p> <ul style="list-style-type: none"> Lead nitrate Sodium sulphate(aq) + Ball(aq) NQ2, S2, O3(aq) + Ball2(aq) 				
GENERAL OBJECTIVE 4.0: Understand the main constituents of air.						
7	<p>4.1 Outline the main constituents of air O₂, CO₂, N₂, rare gases and H₂O.</p> <p>4.2 State the proportion by volume of each constituent in the air.</p> <p>4.3 Define air pollution</p> <p>4.4 List the causes of air pollution.</p> <p>4.5 State the consequences of air pollution.</p>	<p>Explain constituents of air O₂, CO₂, N₂, rare gases and H₂O.</p> <p>Explain the proportion by volume of each constituent in the air.</p> <p>Explain air pollution</p> <p>Explain the causes of air pollution</p> <p>Explain the consequences of air pollution.</p>	<p>Text books</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p> <p>Slides</p>			

	4.6 Describe methods of checking air pollution.	Explain methods of checking air pollution.				
GENERAL OBJECTIVE 5.0: Appreciate the importance of water in our society						
8-9	5.1 Define water as a universal solvent	Explain water as a universal solvent	Textbooks Journals Whiteboard Marker Computer Internet Projector Slides	Carry out the electrolysis of water using Hoffman Voltmeter to identify the constituents of water	Guide student to: Carry out the electrolysis of water using Hoffman Voltmeter to identify the constituents of water	Hoffman Voltmeter
	5.2 Explain the constituents of water.	Explain the constituents of water.				Distiller
	5.3 List the properties of water.	Explain the properties of water.				
	5.4 List the sources of water.	Explain the sources of water		Carry out distillation of water to produce distilled water which can be used for routine experiments in laboratories and charging of batteries.	Carry out distillation of water to produce distilled water which can be used for routine experiments in laboratories and charging of batteries.	
	5.5 Differentiate between hard and soft water.	Explain the difference between hard and soft water.				
	5.6 Describe methods of softening hard water.	Explain methods of softening hard water				
	5.7 Describe methods of purifying drinking water.	Explain methods of purifying drinking water.				
	5.8 Describe water cycle	Explain water cycle				
	5.9 List the causes of water	Explain the causes of				

	pollution.	water pollution.				
	5.10 List methods of checking water pollution in our society	Explain methods of checking water pollution in our society				
GENERAL OBJECTIVE 6.0: Know the general characteristics and properties of acids, bases and salts						
10-11	<p>6.1 Define acid, base and salts.</p> <p>6.2 List examples of common acids, bases, and salts.</p> <p>6.3 List the properties of acids, base and salts</p> <p>6.4 Differentiate between concentrated and diluted acids.</p> <p>6.5 Give examples of concentrated and diluted acids.</p> <p>6.6 List uses of acids, bases, and salts mentioned in 6.2 above.</p>	<p>Explain acid, base and salts.</p> <p>Explain examples of common acids, bases, and salts.</p> <p>Explain the properties of acids, base and salts</p> <p>Explain the differences between concentrated and diluted acids.</p> <p>Explain examples of concentrated and diluted acids.</p> <p>Explain uses of acids, bases, and salts mentioned in 6.2 above.</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p> <p>Slides</p>	<p>Carry out simple test for acids, bases and salts.</p> <p>Use indicators to distinguish between acids, bases and salts.</p> <p>Carry out acid base titration.</p> <p>Obtain salt such as sodium chloride from sea water by evaporation process.</p>	<p>Guide students to:</p> <p>Carry out simple test for acids, bases and salts.</p> <p>Use indicators to distinguish between acids, bases and salts.</p> <p>Carry out acid base titration.</p> <p>Obtain salt such as sodium chloride from sea water by evaporation process.</p>	<p>Acid</p> <p>Bases</p> <p>Bunsen Burner</p> <p>Gas</p> <p>Burette</p> <p>Pipette</p> <p>Evaporating dish</p> <p>Volumetric Flask</p> <p>Litmus paper</p> <p>Wire gauze</p> <p>Tripod Stand</p>

						Retord Stand Clamp Sea water
GENERAL OBJECTIVE 7.0: Understand the locally available metals, their sources, locations and economic importance.						
12-13	<p>7.1 Define metals</p> <p>7.2 Describe the properties of metals</p> <p>7.3 List the locally available metals e.g.:</p> <ul style="list-style-type: none"> • Iron and Steel • Tin • Aluminium • Uranium etc. <p>7.3 List the places where the metals mentioned in 7.1 above are located.</p> <p>7.4 Outline the method of purification of metals listed in 7.1 above.</p> <p>7.5 Explain the uses some common metals with specific examples.</p>	<p>Explain metals</p> <p>Explain the properties of metals</p> <p>Explain the locally available metals e.g.:</p> <ul style="list-style-type: none"> • Iron and Steel • Tin • Aluminium • Uranium etc. <p>Explain the places where the metals mentioned in 7.1 above are located.</p> <p>Explain the methods of purification of metals listed in 7.1 above.</p> <p>Explain the uses some common metals with specific examples.</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p> <p>Slides</p>			

GENERAL OBJECTIVE 8.0: Know the nature and importance of organic compounds.

14	8.1 Explain organic compound	Explain organic compound	Textbooks Journals Whiteboard Marker Computer Internet Projector Slides.	Carry out an experiment to produce soap from palm oil and potash rich ashes.	Guide student to: Carry out an experiment to produce soap from palm oil and potash rich ashes.	Millet Distillation Apparatus Soap Bunsen Burner Tripod Stand Wire Gauze Pyrex glass Spatula Stirrer Balance Oil Strong base Water bath
	8.2 Define hydrocarbons.	Explain hydrocarbons.				
	8.3 Explain the class of organic compound through the functional groups.	Explain the class of organic compound through the functional groups.		Carryout an experiment to produce ethanol through the fermentation of locally available cereals.	Carryout an experiment to produce ethanol through the fermentation of locally available cereals.	
	8.4 Outline the production process of petroleum and its products from crude oil e.g.: • Petrol • Kerosene • Plastic.	Explain the production process of petroleum and its products from crude oil e.g.: • Petrol • Kerosene • Plastic.				
	8.5 Explain the production of soap through saponification process.	Explain the production of soap through saponification process.				
	8.6 Differentiate between a soap and a detergent.	Explain the differences between a soap and a detergent.				
	8.7 Define alcohol	Explain alcohol				

	8.8 Outline sources of alcohol	Explain sources of alcohol				
	8.9 List different classes of alcohol	List different classes of alcohol				
	8.10 Outline the methods of production of alcohol from local raw materials e.g.: • Corn • Cocoyam • Banana • Sugar-cane, etc.	Explain the methods of production of alcohol from local raw materials e.g.: • Corn • Cocoyam • Banana • Sugar-cane, etc.				
	8.11 State the various uses of alcohol and ethanol.	Explain the various uses of alcohol and ethanol.				
GENERAL OBJECTIVE 9.0: Understand the concepts of chemical Reactions						
15	9.1 Define chemical reaction	Explain chemical reaction	Textbooks Journals Whiteboard			
	9.2 List different types of chemical reaction	Explain different types of chemical reaction	Marker Computer Internet			
	9.3 Explain Catalysts used in chemical reactions	Explain Catalysts used in chemical reactions	Projector Slides.			

	9.4 Give examples of different types of reaction mentioned in 9.2 above	Give examples of different types of reaction mentioned in 9.2 above				
	9.5 Define rate of chemical reaction	Explain rate of chemical reaction				
	9.6 Explain factors affecting rates of chemical reactions	Explain factors affecting rates of chemical reactions				
	9.7 Outline some chemical reaction that has beneficial industrial applications	Outline some chemical reaction that has beneficial industrial applications				

EVALUATION

C/A:60%

EXAMS: 40%

TOTAL: 100%

PROGRAMME: NATIONAL DIPLOMA		
COURSE TITLE: General Biology I	COURSE CODE: STB 101	CONTACT HOURS: 2
	CREDIT UNIT: 3	THEORETICAL: 2
YEAR: I SEMESTER: 0	PRE-REQUISITE:	PRACTICAL: 1
GOAL: This module is designed to introduce the students to basic biological sciences and to make them understand the environment in which they live		
GENERAL OBJECTIVES: On completion of this course, the student should be able to: <ul style="list-style-type: none"> 1.0 Understand the concept of living things 2.0 Know the structure of cells 3.0 Understand the principal features of different groups of living things 4.0 Know the methods by which plants and animals feed themselves 5.0 Understand the methods by which organisms generate energy from their food 6.0 Know the methods by which substances are moved in the bodies of plants and animals 7.0 Understand how plants and animals remove their waste products 8.0 Understand how living things move 9.0 Know how living things increase in size and number 10.0 Understand the interaction between an organism and its environment 11.0 Know man's interference with his environment 12.0 Understand the fundamentals of inheritance 		

PROGRAMME: NATIONAL DIPLOMA						
COURSE TITLE: General Biology I			COURSE CODE: STB 101		CONTACT HOURS: 3	
			CREDIT UNIT: 3		THEORETICAL: 2	
YEAR: I SEMESTER: 0			PRE-REQUISITE:		PRACTICAL: 1	
COURSE SPECIFICATION: THEORETICAL AND PRACTICAL						
GOAL: This module is designed to introduce the students to basic biological sciences and to make them understand the environment in which they live						
GENERAL OBJECTIVE: 1.0 Understand the concept of living things						
THEORETICAL CONTENT				PRACTICAL CONTENT		
WEEK	SPECIFIC LEARNING OUTCOME	TEACHER’S ACTIVITIES	RESOURCES	SPECIFIC LEARNING OUTCOME	TEACHER’S ACTIVITIES	RESOURCES
1-2	1.1 Explain Living Things	Explain living things	Textbooks			
	1.2 Define Biology	Explain Biology	Journals			
	1.3 Explain living things and non-living things	Explain living and non-living things in the immediate environment	Whiteboard			
	1.4 Explain characteristics of living things (e.g., MR NIGER D or MARRIAGE’D)	Explain the characteristics of living organisms	Marker			
	1.5 Explain the differences between a plant, an animal and a protist e.g.: Mango tree plant,	Explain the differences between a plant, an animal and a protist e.g.: Mango tree plant,	Computer			
			Internet			
			Projector			

	mycobacterium tuberculosis for protist and man for animal)	mycobacterium tuberculosis for protist and man for animal)				
GENERAL OBJECTIVE 2.0: Know the structure of cells						
3-4	2.1 Discuss History and Discovery of Cell	Discuss History and Discovery of Cell	Text books Journals Whiteboard Marker Computer Internet Projector	Observes some cells under the microscope (e.g. a plant cell, and animal cell)	Guide students to: Observes some cells under the microscope (e.g. a plant cell, and animal cell)	Microscope Magnifying glass Prepared slide
	2.2 Explain a Cell theory	Explain a cell theory				
	2.3 Define Cell	Explain Cell				
	2.4 Different types of Cell (Eukaryotic & Procaryotic)	Distinguish between prokaryotic and eukaryotic cells		Draw and label the cells observed in above practical	Draw and label the cells observed in above practical	
	2.5 Differentiate between an animal cell and a plant cell	Differentiate between plant and animal cell				
	2.6 List the components/Cell Organelles (Structure)	Explain the components/Cell Organelles (Structure)				
	2.7 Explain the function of cell organelles (Structure)	Explain the functions of the different cell organelles				
	2.8 Explain cell Division	Explain cell Division				

2.9	Explain the processes of cell division	Explain the processes of cell division			
2.10	Differentiate between Mitosis and Meiosis cell division	Differentiate between Mitosis and Meiosis cell division			
2.11	Explain Mitotic cell division and Meiotic Cell Division Diagrammatically	Explain Mitotic cell division and Meiotic Cell Division Diagrammatically			
2.12	Explain the differences between mitosis and meiosis cell division	Explain the differences between mitosis and meiosis cell division			
2.13	List the importance of cell division	Explain the importance of cell division			
GENERAL OBJECTIVE 3.0: Understand the principal features of different groups of living things					

5-6	<p>3.1 Explain Major divisions of the Plant Kingdom:</p> <ul style="list-style-type: none"> • Thallophyta • Bryophyta • Pteridophyta • Spermatophyta <p>(indicate the features of a subgroup called angiosperm and gymnosperm)</p> <p>3.2 State the characteristic features of each division in 3.1 above</p> <p>3.3 List the differences between algae and fungi</p> <p>3.4 List two examples of each group in 3.3 above</p> <p>3.5 Explain the features of Flowering Plant</p>	<p>Explain the four major divisions under the plant kingdom:</p> <ul style="list-style-type: none"> • Thallophyta • Bryophyta • Pteridophyta • Spermatophyta <p>(indicate the features of a subgroup called angiosperm and gymnosperm)</p> <p>Explain the characteristic features of each division in 3.1 above</p> <p>Explain the differences between algae and fungi</p> <p>Explain two examples of each group in 3.3 above</p> <p>Explain the external structures of a flowering plant</p>	<p>Text books</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>			
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3.6 Explain types of flower i.e., insect pollinated flower and wind pollinated flower based on the structural differences	Explain types of flower i.e., insect pollinated flower and wind pollinated flower based on the structural differences				
3.7 Describe the external structures of a flowering plant	Describe the external structures of a flowering plant				
3.8 Explain Animal Kingdom	Explain Animal Kingdom				
3.9 Explain the two (2) group of the animal kingdom i.e., the invertebrate animal and vertebrate animal each with examples and features differentiating them.	Explain the two (2) group of the animal kingdom i.e., the invertebrate animal and vertebrate animal each with examples and features differentiating them.				
3.10 List the different phyla under animal kingdom	Explain the different phyla under animal kingdom				
3.11 Explain the distinguishing	Explain the distinguishing				

	characteristics of each phylum mentioned in 3.10	characteristics of each phylum mentioned in 3.6				
	3.12 State two common examples of animals belonging to each group	Explain two common examples of animals belonging to each group				
	3.13 Describe the external structure of a mammal	Explain the external structure of a mammal				
GENERAL OBJECTIVE 4.0: Know the methods by which plants and animals feed themselves (Nutrition)						
7	4.1 Explain Plant and Animal Nutrition	Explain food Plant and Animal Nutrition	Text books Journals Whiteboard Marker Computer Internet Projector	Draw and label the digestive system in man	Guide students to: Draw and label the digestive system in man	Diagrams
	4.2 Distinguish between the food of plants and that of animals	Distinguish between the food plants and that of animals				
	4.3 Distinguish between autotrophic and heterotrophic modes of feeding	Distinguish between autotrophic and heterotrophic modes of feeding				
	4.4 Explain with examples the different types of heterotrophies e.g. parasitism and saprophytism.	Explain with examples the different types of heterotrophies e.g. parasitism and saprophytism.				

	<p>4.5 Explain the different types of autotrophy e.g.</p> <ul style="list-style-type: none"> • Photosynthesis, • Chemosynthesis. <p>4.6 Explain Animal Nutrition</p> <p>4.7 Explain Balanced diet and its composition</p> <p>4.8 List the stages involved in feeding animals, e.g.:</p> <ul style="list-style-type: none"> • Ingestion • Digestion • Assimilation and • Egestion. 	<p>Explain the different types of autotrophy e.g.</p> <ul style="list-style-type: none"> • Photosynthesis, • Chemosynthesis. <p>Explain Animal Nutrition</p> <p>Explain Balanced diet and its composition</p> <p>Explain the stages involved in feeding animals e.g.:</p> <ul style="list-style-type: none"> • Ingestion • Digestion • Assimilation and • Egestion. 				
GENERAL OBJECTIVE 5.0: Understand the methods by which organisms generate energy from their food						
8	<p>5.1 Define respiration</p> <p>5.2 Distinguish between respiration and breathing</p> <p>5.3 Differentiate between aerobic and anaerobic respiration.</p>	<p>Explain respiration</p> <p>Explain the differences between respiration and breathing</p> <p>Explain the differences between aerobic and anaerobic respiration</p>	<p>Text books</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>			

	<p>5.4 Explain sites of respiration in cells (e.g. mitochondrion and free cytoplasm)</p> <p>5.5 State the economic importance of fermentation (Production of alcoholic beverages, lactic acid, vinegar, bread, garri and condiment production.)</p> <p>5.6 List the respiratory organs in animals (e.g. plasma membrane in protozoa, tracheal system in arthropods, gills in fish and lungs in terrestrial animals)</p>	<p>Explain sites of respiration in cells (e.g. mitochondrion and free cytoplasm)</p> <p>Explain the economic importance of fermentation Production of alcoholic beverages, lactic acid, vinegar, bread, garri and condiment production.)</p> <p>Explain the respiratory organs in animals (e.g. plasma membrane in protozoa, tracheal system in arthropods, gills in fish and lungs in terrestrial animals)</p>				
GENERAL OBJECTIVE 6.0: Know the methods by which substances are moved in the bodies of plants and animals						
9	<p>6.1 Explain the following:</p> <ul style="list-style-type: none"> • Osmosis • Diffusion • Plasmolysis <p>6.2 Define transpiration</p>	<p>Explain the following:</p> <ul style="list-style-type: none"> • Osmosis • Diffusion • Plasmolysis <p>Explain transpiration</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>	<p>Demonstrate osmosis and diffusion by simple experiments.</p>	<p>Guide student to :</p> <p>Demonstrate osmosis and diffusion by simple experiments.</p>	<p>Perfume</p> <p>Potato</p> <p>1000ml beaker</p> <p>salt</p>

	6.3 Explain the importance of transpiration in living organisms	Explain the importance of transpiration in living organisms		Draw and label the structure of a mammalian heart.	Draw and label the structure of a mammalian heart.	
	6.4 Explain the importance of translocation	Explain the importance of translocation				
	6.5 Explain Vascular System of Plant	Explain the vascular system of the plant				
	6.6 Differentiate between Vascular system of Dicot and Monocot Plants	Differentiate between Vascular system of Dicot and Monocot Plants				
	6.7 Explain the Structure and Function of blood in higher animals	Explain the structure and function of blood in higher animals				
	6.8 Describe clotting of blood	Describe clotting of blood				
GENERAL OBJECTIVE 7.0: Understand how plants and animals remove their waste products						
10	7.1 Define Excretion and give examples	Explain Excretion and give examples	Textbooks Journals Whiteboard			
	7.2 Outline the excretory organs and excretory products in man	Explain excretory organs and excretory products in man	Marker Computer Internet Projector			

	7.3 List the structures involved in excretion and the excretory products in plants	Explain the structures involved in excretion and the excretory products in plants				
	7.4 Explain the processes of osmo-regulation in plants and various types of animals	Explain the processes of osmo-regulation in plants and various types of animals				
GENERAL OBJECTIVE 8.0: Understand how living things move						
11	8.1 Explain the organs of locomotion in animals (e.g. pseudopodia: cilia, chaetae, jointed limbs etc.)	Explain the organs of locomotion in animals (e.g. pseudopodia: cilia, chaetae, jointed limbs etc.)	Textbooks Journals Whiteboard Marker Computer Internet Projector			
	8.2 Explain general plan of the Skeletal System in mammals (axial skeleton and appendicular skeleton)	Explain the general plan of the skeleton in mammals (axial skeleton and appendicular skeleton)				
	8.3 Explain types of skeleton i.e., chitinous exoskeleton in insects and endoskeleton in mammals.	Explain types of skeleton i.e., chitinous exoskeleton in insects and endoskeleton in mammals.				

	8.4 State the functions of the skeleton in higher animals	Explain the functions of the skeleton in higher animals				
	8.5 List the different types of movements in plants (e.g. tropism's taxisms, nastic movements e.t.c)	Explain the different types of movements in plants (e.g. tropism's taxisms, nastic movements e.t.c)				
GENERAL OBJECTIVE 9.0: Know how living things increase in size and number						
12	9.1 Explain Reproduction in living things	Explain reproduction in living things	Textbooks Journals Whiteboard Marker Computer Internet Projector			
	9.2 Distinguish between sexual, asexual and vegetative propagation	Explain the difference between sexual and asexual reproduction				
	9.3 List some examples of asexual reproduction in plants and animals (e.g. binary fission in bacteria, budding in yeast, fragmentation in spirogyra, grafting etc.)	Explain some examples of asexual reproduction in plants and animals (e.g. binary fission in bacteria, budding in yeast, fragmentation in spirogyra, grafting etc.)				
	9.4 List the advantages of asexual and sexual	Explain the advantages of asexual and sexual				

	<p>reproduction in plants and animals</p> <p>9.5 Discuss the modern advances in plants cultures and in invitro and in vivo fertilization (e.g. test tube babies and artificial insemination respectively).</p> <p>9.6 Describe methods of measuring growth in living things</p>	<p>reproduction in plants and animals</p> <p>Explain the modern advances in plants cultures and in invitro and in vivo fertilization (e.g. test tube babies and artificial insemination respectively).</p> <p>Explain methods of measuring growth in living things</p>				
GENERAL OBJECTIVE 10.0: Understand the interaction between an organism and its environment						
13	<p>10.1 Define the following terms</p> <ul style="list-style-type: none"> • Ecology (autecology and synecology) • Habitat (terrestrial and aquatic [lotic and lentic]) • Community • Population <p>10.2 Explain the interaction of living things as</p>	<p>Explain the following terms:</p> <ul style="list-style-type: none"> • Ecology (autecology and synecology) • Habitat (terrestrial and aquatic [lotic and lentic]) • Community • population <p>Explain the interaction of living things as shown</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>			

	<p>shown by predation, parasitism, saprophytism, commensalism, scavenging</p>	<p>by predation, parasitism, saprophytism, commensalism</p>				
	<p>10.3 List examples of interactions in 10.2 above</p>	<p>Explain examples of interactions in 10.2 above</p>				
	<p>10.4 Explain the food relationship in the community (e.g. food chain, food web, pyramid of numbers, pyramid of energy etc.)</p>	<p>Explain the food relationship in the community (e.g. food chain, food web, pyramid of numbers etc.)</p>				
	<p>10.5 List useful and harmful microbes giving</p>	<p>Explain useful and harmful microbes giving</p>				
	<p>10.6 State the causes and control of the following diseases</p> <ul style="list-style-type: none"> • Malaria • Guinea worm • Bilharzia • River blindness • Cholera • Meningitis 	<p>Explain the causes and control of the following diseases</p> <ul style="list-style-type: none"> • Malaria • Guinea worm • Bilharzia • River blindness • Cholera • Meningitis 				

	<ul style="list-style-type: none"> • Typhoid fever <p>10.7 State the common sources and types of environmental pollution.</p> <p>10.8 Explain the types of environmental pollutants (biodegradable and non-biodegradable pollutants)</p> <p>10.9 Describe the effects of environmental pollution on living organisms</p> <p>10.10 List the control measures of environmental pollution described in 10.9 above</p>	<ul style="list-style-type: none"> • Typhoid fever <p>Explain the common sources of environmental pollution.</p> <p>Explain the types of environmental pollutants (biodegradable and non-biodegradable pollutants)</p> <p>Explain the effects of environmental pollution on living organisms</p> <p>Explain the control measures of environmental pollution described in 10.9 above</p>				
GENERAL OBJECTIVE 11.0: Understand man's interference with his environment						
14	<p>11.1 Explain Types of soil</p> <p>11.2 Explain the composition and characteristics of a fertile soil</p>	<p>Explain Types of soil</p> <p>Explain the composition and characteristics of a fertile soil</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>			

	11.3 List the causes of soil erosion	Explain the causes of soil erosion				
	11.4 Give three methods of prevention of soil erosion	Explain three methods of prevention of soil erosion				
	11.5 Describe man's activities (Farming, Mining, Deforestation, overgrazing that accentuate disasters e.g. desert encroachment, flooding etc.	Explain man's activities (Farming, Mining, Deforestation, overgrazing that accentuate disasters e.g. desert encroachment, flooding etc.				
	11.6 Enumerate different methods by which the deleterious effects can be minimized	Explain different methods by which the deleterious effects can be minimized				
	11.7 Explain the need for the creation of dams and lakes	Explain the need for the creation of dams and lakes				
	11.8 List the disadvantages of man-made lakes	Explain the disadvantages of man-made lakes				

GENERAL OBJECTIVE 12.0: Understand the fundamentals of inheritance

15	12.1 Explain the role of chromosomes and genes found in the nucleus of the cell	Explain the role of chromosomes and genes found in the nucleus of the cell	Textbooks Journals Whiteboard Marker Computer			
	12.2 Explain the role of gametes in cross fertilization	Explain the role of gametes in cross fertilization	Internet Projector			
	12.3 Define hybridization, phenotypes and genotypes	Explain hybridization, phenotypes and genotypes				
	12.4 Explain the ABO blood groupings	Explain the ABO blood groupings				
	12.5 Describe how sex is determined in man	Explain how sex is determined in man				
	12.6 Explain sex-linked characters (e.g. haemophilia)	Explain sex-linked characters (e.g. haemophilia)				

EVALUATION:

C.A: 60%

EXAMINATION: 40%

TOTAL: 100%

NATIONAL BOARD FOR TECHNICAL EDUCATION

HIGHER NATIONAL DIPLOMA

PROGRAMME: HIGHER NATIONAL DIPLOMA		
COURSE TITLE: General Physics II	COURSE CODE: STP 301	CONTACT HOURS: 3
	CREDIT UNIT: 3	THEORETICAL: 2
YEAR: I SEMESTER: 0	PRE-REQUISITE:	PRACTICAL: 1
GOAL: This course is designed to acquaint students with the knowledge and skills of core concepts in classical physics.		
GENERAL OBJECTIVES: On completion of this course, the student should be able to: 1.0 Understand units and dimensions 2.0 Understand the difference between vectors and scalars. 3.0 Know what is meant by rectilinear motion. 4.0 Understand the meaning of work, energy and power. 5.0 Understand friction as it occurs both in liquids and solids 6.0 Know the concept of elasticity 7.0 Know the general concept of static electricity 8.0 Know the nature of current electricity 9.0 Know the concept of heat and temperature 10.0 Understand reflection and refraction 11.0 Understand wave propagation		

PROGRAMME: HIGHER NATIONAL DIPLOMA						
COURSE TITLE: General Physics II		COURSE CODE: STP 301		CONTACT HOURS: 3		
		CREDIT UNIT: 3		THEORETICAL: 2		
YEAR: I	SEMESTER: 0	PRE-REQUISITE:		PRACTICAL: 1		
COURSE SPECIFICATION: THEORETICAL AND PRACTICAL						
GOAL: This course is designed to acquaint students with the knowledge of core concepts in classical physics.						
GENERAL OBJECTIVE 1.0: Understand units and dimensions						
THEORETICAL CONTENT				PRACTICAL CONTENT		
WEEK	SPECIFIC LEARNING OUTCOME	TEACHER'S ACTIVITIES	RESOURCES	SPECIFIC LEARNING OUTCOME	TEACHER'S ACTIVITIES	RESOURCES
1-2	1.1 Define units and dimensions	Explain units and dimensions	Textbooks Journals Whiteboard			
	1.2 Differentiate between fundamental and derived units	Differentiate between fundamental and derived units	Marker Computer Internet Projector			
	1.3 Give examples of both derived and fundamental units	Explain examples of both derived and fundamental units				
	1.4 Explain how to check the correctness of equations like: $V^2 = u^2 + 2as$; $V = U + at$; $T = 2\pi\sqrt{L/G}$ etc. using dimensions	Explain how to check the correctness of equations like $v^2 = u^2 + 2as$; $V = U + at$; $T = 2\pi\sqrt{L/G}$ etc. using dimensions				

GENERAL OBJECTIVE 2.0: Understand the difference between vectors and scalars.						
3	2.1 Define vectors and scalars	Explain vectors and scalars	Textbooks Journals Whiteboard			
	2.2 Explain examples of both scalars and vectors quantities	Explain examples of both scalars and vectors quantities	Marker Computer Internet Projector			
	2.3 Explain how to solve simple problems involving scalars and vectors	Explain how to solve simple problems involving scalars and vectors				
General Objective 3.0: Know what is meant by rectilinear motion.						
4-5	3.1 Explain velocity and speed	Explain velocity and speed	Textbooks Journals Whiteboard	Perform simple experiments to measure velocity and speed Solve problems using equations derived in 3.2 above Solve simple problems on motions	Guide student to: Perform simple experiments to measure velocity and speed Solve problems using equations derived in 3.2 above Solve simple problems on motions	Speedometer
	3.2 Differentiate between velocity and speed	Explain between velocity and speed	Marker Computer Internet Projector			
	3.3 Explain the equations of uniform motion in a straight line by graphical method	Explain the equations of uniform motion in a straight line by graphical method				
	3.4 Explain how to solve problems using equations derived in 3.3 above	Explain how to solve problems using equations derived in 3.3 above				

	<p>3.5 State Newton's law of motion</p> <p>3.6 Derive the relation between forces, mass and acceleration. i.e. $F=ma$ where F=force a= acceleration=mass</p> <p>3.7 Define force, mass and weight</p> <p>3.8 Explain how to solve simple problems using equation derived in 3.6 above.</p>	<p>Explain Newton's law of motion</p> <p>Derive the relation between forces, mass and acceleration. i.e. $F=ma$ where F=force a= acceleration=mass</p> <p>Explain force, mass and weight</p> <p>Explain how to solve simple problems using equation derived in 3.6 above.</p>				
GENERAL OBJECTIVE 4.0: Understand the meaning of work, energy and power						
6	<p>4.1 Define work, energy and power</p> <p>4.2 List types of energy</p> <p>4.3 Explain the law of conservation of energy and give examples</p>	<p>Explain work, energy and power</p> <p>Explain types of energy</p> <p>Explain the law of conservation of energy and give examples</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>			

	<p>4.4 Explain the conservation of kinetic energy and potential energy</p> <p>4.5 Explain how to solve simple problems on work, energy and power</p>	<p>Explain the conservation of kinetic energy and potential energy</p> <p>Explain how to solve simple problems on work, energy and power</p>				
GENERAL OBJECTIVE 5.0: Understand friction as it occurs both in liquids and solids						
7	<p>5.1 Define friction</p> <p>5.2 Explain types of friction</p> <p>5.3 Differentiate between static and dynamic friction</p> <p>5.4 Explain the applications of friction to human existence</p> <p>5.5 Explain the effects of friction to human existence</p>	<p>Explain friction</p> <p>Explain types of friction</p> <p>Explain the difference between static and dynamic friction</p> <p>Explain the applications of friction to human existence</p> <p>Explain the effects of friction to human existence</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>			

	5.6 Describe methods of reducing friction	Explain simple methods of reducing friction				
GENERAL OBJECTIVE 6.0: Know the concept of elasticity						
8-9	6.1 Define the following: <ul style="list-style-type: none"> Elasticity, Stress, Strain Elastic limit 	Explain the following: <ul style="list-style-type: none"> Elasticity, Stress, Strain Elastic limit 	Textbooks Journals Whiteboard Marker Computer Internet Projector	Conduct simple experiments on the following: <ul style="list-style-type: none"> Elasticity, Stress Strain Elastic limit 	Guide students to: Conduct simple experiments on the following: <ul style="list-style-type: none"> Elasticity, Stress Strain Elastic limit 	Elastic band
	6.2 State the relationship between load and extension	Explain the relationship between load and extension				
	6.3 Estimate elastic energy from work done on elasticity body	Estimate elastic energy from work done on elasticity body		Demonstrate the application of elasticity in day-to-day life.	Demonstrate the application of elasticity in day-to-day life.	
	6.4 Give practical examples of application of elasticity in day-to-day life.	Explain the practical application of elasticity in day-to-day life.				
GENERAL OBJECTIVE 7.0: Know the general concept of static electricity and magnetisms						
10-11	7.1 Define static electricity	Explain static electricity	Text books Journals Whiteboard	Use the goldleaf electroscope to determine the nature of charges on different bodies.	Guide student to : Use the goldleaf electroscope to determine the nature of charges on different bodies	Goldleaf electroscope
	7.2 Explain the processes of charging by friction, and conduction	Explain the processes of charging by friction, and conduction	Marker Computer Internet			Rubber balloon Glass rod

	<p>7.3 Explain the working principles of van de Greaf generator</p> <p>7.4 Define Magnets</p> <p>7.5 Explain the classifications of magnets</p> <p>7.6 Describe how magnets are made</p> <p>7.7 Explain the application of magnets in daily life</p>	<p>Explain the working principles of van de Greaf generator</p> <p>Explain Magnets</p> <p>Explain the classifications of magnets</p> <p>Explain how magnets are made</p> <p>Explain the application of magnets in daily life</p>	Projector	<p>Perform some laboratory activities to generate some static electricity such as:</p> <ul style="list-style-type: none"> Rubber balloon and glass rod experiments Metal sphere experiment etc. 	<p>Perform some laboratory activities to generate some static electricity such as:</p> <ul style="list-style-type: none"> Rubber balloon and glass rod experiments Metal sphere experiment etc. 	Metal sphere
GENERAL OBJECTIVE 8.0: Know the nature of current electricity						
12	<p>8.1 Define electric current</p> <p>8.2 Explain the units of electric current</p> <p>8.3 State and express Ohm's Law mathematically.</p> <p>8.4 Explain the relationship between voltage (V),</p>	<p>Explain electric current</p> <p>Explain the units of electric current</p> <p>State and express Ohm's Law mathematically.</p> <p>Explain the relationship between voltage (V), current (I)</p>	<p>Text books</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>	<p>Conduct simple experiments in electricity such as:</p> <ul style="list-style-type: none"> Light bulb experiment Resistor experiment etc. 	<p>Guide student to:</p> <p>Conduct simple experiments in electricity such as:</p> <ul style="list-style-type: none"> Light bulb experiment Resistor experiment etc. 	<p>Light bulb</p> <p>Resistor</p> <p>Circuit</p> <p>Power Source</p>

	<p>current (I) and resistance (R).</p> <ul style="list-style-type: none"> • $V=IR$ where • V=Voltage • I=Current • R=Resistance 	<p>and resistance (R).</p> <ul style="list-style-type: none"> • $V=IR$ where • V=Voltage • I=Current • R=Resistance 				
	8.5 Illustrate diagrammatically electrical circuit symbols.	Illustrate the circuit symbols with their names in circuit arrangement				
	8.6 Explain operation of simple electrical measuring instruments e.g. Voltmeter, Ammeter etc.	Explain operation of simple electrical measuring instruments e.g. Voltmeter, Ammeter etc.				
	8.7 Explain how to solve simple problems involving resistors in series and parallel	Explain how to solve simple problems involving resistors in series and parallel				
	8.8 Define electrical energy and power	Explain electrical energy and power				

	8.9 Explain how to solve problems involving electrical energy	Explain how to solve problems involving electrical energy				
	8.10 Differentiate between primary and secondary cells	Differentiate between primary and secondary cells				
	8.11 List example of both primary and secondary cells	List example of both primary and secondary cells				
	8.12 Solve simple problem involving primary cells in series and parallel circuits	Solve simple problem involving primary cells in series and parallel circuits				
GENERAL OBJECTIVE 9.0: Know the concept of heat and temperature						
13	9.1 Define temperature in relation to heat and energy 9.2 Explain the properties of thermometric liquids.	Explain temperature in relation to heat and energy. Explain the properties of thermometric liquids.	Text books Journals Whiteboard Marker Computer Internet Projector	Conduct experiments in thermometric liquids and thermometers such as: <ul style="list-style-type: none"> Thermal expansion experiment Calibration of thermometers Thermometer comparison experiment 	Guide student to: Conduct experiments in thermometric liquids and thermometers such as: <ul style="list-style-type: none"> Thermal expansion experiment 	Thermometer Thermometric liquids Water bath Bunsen Burner Gas

9.3 Explain temperature scales i.e. Kelvin and Celsius.	Explain temperature scales i.e. Kelvin and Celsius.			<ul style="list-style-type: none"> • Calibration of thermometers • Thermometer comparison experiment 	
9.4 List the various types of thermometers.	Explain the various types of thermometers.				
9.5 Explain the working principles of the thermometers listed in 9.4 above.	Explain the working principles of the thermometers listed in 9.4 above.				
9.6 Give various examples of the applications of the thermometers listed in 9.4 above.	Explain various examples of the applications of thermometers listed in 9.4 above.				
9.7 Explain the modes of heat transfer <ul style="list-style-type: none"> • Conduction • Convection • Radiation 	Explain the modes of heat transfer <ul style="list-style-type: none"> • Conduction • Convection • Radiation 				
9.8 State the effect of heat on liquids and solids	Explain the effect of heat on liquids and solids				

	9.9 Explain some applications of expansion in real life situations.	Explain some applications of expansion in real life situations.				
	9.10 Define expansivity and solve simple problems using the related equation	Explain expansivity and solve simple problems using the related equation				
	9.11 Define heat capacity, specific heat capacity and specific latent heats.	Explain heat capacity, specific heat capacity and specific latent heats.				
	9.12 Explain how to solve problems involving heat capacity, specific heat capacity, latent heat of vaporisation	Explain how to solve problems involving heat capacity, specific heat capacity, latent heat of vaporisation				
GENERAL OBJECTIVE 10.0: Understand reflection and refraction						
14	10.1 Define reflection and refraction	Explain reflection and refraction	Text books Journals Whiteboard			
	10.2 Explain how reflection and refraction take place on plane and curved surfaces.	Explain how reflection and refraction take place on plane and curved surfaces.	Marker Computer Internet Projector			

	10.3 State the laws of refraction and reflection.	Explain the laws of refraction and reflection.				
	10.4 Explain apparent depth and total internal reflection	Explain apparent depth and total internal reflection				
	10.5 Explain the formation of images by plane mirror, spherical mirrors and lenses	Explain the formation of images by plane mirror, spherical mirrors and lenses				
GENERAL OBJECTIVE 11.0: Understand wave propagation						
15	11.1 Define wave propagation 11.2 Explain types of waves 11.3 Define the following: <ul style="list-style-type: none"> • Displacement, • Amplitude • frequency, • Period, • Wavelength, • Velocity • Phase of a wave 	Explain wave propagation. Explain types of waves Explain the following: <ul style="list-style-type: none"> • Displacement, • Amplitude • frequency, • Period, • Wavelength, • Velocity • Phase of a wave 	Text books Journals Whiteboard Marker Computer Internet Projector	Conduct simple experiments on water wave such as: <ul style="list-style-type: none"> • Wave generation, • Reflection and Refraction of waves. 	Guide student to: Conduct simple experiments on waves such as: <ul style="list-style-type: none"> • Wave generation, • Reflection and Refraction of waves 	Ruler Water Shallow Basin Dropper Colours Measuring Tape

	<p>11.4 Explain how to solve simple problems involving</p> $v = f\lambda$ <p>Where v= velocity f= frequency λ= wavelength</p> <p>11.5 Give practical applications of waves in daily life.</p>	<p>Explain how to solve simple problems involving</p> $v = f\lambda$ <p>Where v= velocity f= frequency λ= wavelength</p> <p>Explain practical applications of waves in daily life</p>				
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EVALUATION

C/A:60%

EXAMS: 40%

TOTAL: 100%

PROGRAMME: HIGHER NATIONAL DIPLOMA (HND)		
COURSE TITLE: General Chemistry II	COURSE CODE: STC 301	CONTACT HOURS: 3
	CREDIT UNITS: 3	THEORETICAL: 2
YEAR: I SEMESTER: 0	PRE-REQUISITE:	PRACTICAL: 1
GOAL: This course is designed to equip student with knowledge of fundamental chemical phenomena and their relevance to everyday life and industry.		
GENERAL OBJECTIVES: On completion of this course, the students should be able to: <ol style="list-style-type: none"> 1.0 Understand the basic structure of the atom. 2.0 Understand the different types of chemicals bonds (their formation and properties). 3.0 Understand the periodic table. 4.0 Understand the concepts of solutions and solubility. 5.0 Understand the concepts of electrolysis 6.0 Understand the laws governing chemical combination. 7.0 Understand the change of matter and behaviour of gas under certain conditions. 8.0 Understand the rate of chemical reaction and factors affecting rate of reaction. 9.0 Appreciate the usefulness of some non-metals like H₂, Cl₂ and their compounds. 10.0 Appreciate the importance of some metals like Cu, Al, Na etc. in our society. 11.0 Appreciate the importance of carbon compound 12.0 Conduct laboratory work based on the above topics. 		

PROGRAMME: HIGHER NATIONAL DIPLOMA (HND)						
COURSE TITLE: General Chemistry II			COURSE CODE: STC 301		CONTACT HOURS: 3	
			CREDIT UNITS: 3		THEORETICAL: 2	
YEAR: I SEMESTER: 0			PRE-REQUISITE:		PRACTICAL: 1	
COURSE SPECIFICATION: THEORETICAL AND PRACTICAL						
GOAL: This course is designed to equip student with knowledge of fundamental chemical phenomena and their relevance to everyday life and industry.						
GENERAL OBJECTIVE 1.0: Understand the basic structure of the atom.						
	THEORETICAL CONTENT			PRACTICAL CONTENT		
WEEK	SPECIFIC LEARNING OUTCOME	TEACHER’S ACTIVITIES	RESOURCES	SPECIFIC LEARNING OUTCOME	TEACHER’S ACTIVITIES	RESOURCES
1	1.1 Define atom. 1.2 Explain the 3 basic constituents of the atom: <ul style="list-style-type: none">• Electrons,• Protons• Neutrons. 1.3 Explain the arrangement of electrons in the atoms	Explain an atom Explain the 3 basic constituents of the atom: <ul style="list-style-type: none">• Electrons,• Protons• Neutrons Explain the arrangement of electrons in the atoms	Text books Journals Whiteboard Marker Charts Computer Internet Projector			

	1.4 State the properties of the electrons, protons and neutrons.	Explain the properties of the electrons, protons and neutrons.				
	1.5 Define the relative atomic mass and relative molecular mass	Explain the relative atomic mass and relative molecular mass.				
	1.6 Define isotopes and give examples of some elements that exhibit isotopy e.g. chlorine.	Explain isotopes and give examples of some elements that exhibit isotopy e.g. chlorine.				
	1.7 Describe how to calculate the relative atomic mass of some isotopic elements.	Explain how to calculate the relative atomic mass of some isotopic elements.				
	1.8 Explain the arrangement of electrons in the atoms.	Explain the arrangement of electrons in the atoms.				
GENERAL OBJECTIVE 2.0: Understand the different types of chemicals bonds (their formation and properties).						
2	2.1 Define chemical bond	Explain chemical bond	Text books Journals Whiteboard			

	<p>2.2 Outline the 3 types of chemical bonds, viz:</p> <ul style="list-style-type: none"> • Covalent bond • Dative bond • Electrovalent bond. 	<p>Explain the 3 types of chemical bonds, viz:</p> <ul style="list-style-type: none"> • Covalent bond • Dative bond • Electrovalent bond 	<p>Marker Charts Computer Internet Projector</p>			
	2.3 Define valency.	Explain valency.				
	2.4 List the valences of some common elements like H ₂ , O ₂ , Na and K	Explain the valences of some common elements like H ₂ , O ₂ , Na and K				
	2.5 Outline the formation of each bond type.	Outline the formation of each bond type.				
	2.6 Outline the properties of each bond type.	Outline the properties of each bond type.				
GENERAL OBJECTIVE 3.0: Understand the periodic table.						
3	3.1 Define periodic table	Define periodic table	Text books Journals Whiteboard			
	3.2 State the importance of periodic table	State the importance of periodic table	Marker Charts Computer			

	<p>3.3 Describe the periodic table.</p> <p>3.4 Explain the relative positions of elements on the table.</p> <p>3.5 Discuss the two main division of the periodic table.</p> <p>3.6 State the general properties of each group of elements.</p>	<p>Explain periodic table.</p> <p>Explain relative positions of elements on the table.</p> <p>Explain the two main division of the periodic table.</p> <p>Explain the general properties of each group of elements</p>	<p>Internet Projector Periodic table</p>			
GENERAL OBJECTIVE 4.0: Understand the concepts of solutions and solubility						
4	<p>4.1 Define a solution.</p> <p>4.2 Differentiate between types of solutions:</p> <ul style="list-style-type: none"> • Saturated • Unsaturated • Super –saturated <p>4.3 Define the term solubility.</p>	<p>Explain a solution.</p> <p>Explain the difference types of solutions:</p> <ul style="list-style-type: none"> • Saturated, • Unsaturated • Super – saturated <p>Explain the term solubility</p>	<p>Text books Journals Whiteboard Marker Charts Computer Internet Projector</p>	<p>Conduct some simple experiments of water quality (physical and chemical).</p>	<p>Guide student to Conduct some simple experiments of water quality (physical and chemical).</p>	<p>Water pH meter</p>

	4.4 Explain the solubility of some compounds like KNO_3 , CuSO_4 , etc. in water.	Explain the solubility of some compounds like KNO_3 , CuSO_4 , etc in water.				
	4.5 Explain the significance of the use of solubility curves.	Explain the significance of the use of solubility curves.				
	4.6 Describe how to solve simple problems based on 4.4 and 4.5 above.	Explain how to solve simple problems based on 4.4 and 4.5 above.				
GENERAL OBJECTIVE 5.0: Understand the concepts of electrolysis						
6	5.1 Define electrolysis.	Explain electrolysis.	Text books Journals			
	5.2 Define electrolytes and non-electrolytes.	Explain electrolytes and non-electrolytes.	Whiteboard Marker Charts			
	5.3 List examples of electrolyte and non-electrolyte	Explain examples of electrolyte and non-electrolyte	Computer Internet Projector			
	5.4 Explain Faraday's law of electrolysis.	Explain Faraday's law of electrolysis.				
	5.5 Explain the uses of	Explain the uses of				

	<p>electrolysis in some electrolytes</p> <p>5.6 Explain the uses of electrolytes.</p> <p>5.7 Derive mathematically the relation between current ionic charge and mass of substances liberated electrodes.</p> <p>5.8 Describe how to solve some problems based on 5.7 above.</p> <p>5.9 List some applications of electrolysis in industry.</p>	<p>electrolysis in some electrolytes</p> <p>Explain the uses of electrolytes.</p> <p>Explain mathematically the relation between current ionic charge and mass of substances liberated electrodes.</p> <p>Explain how to solve some problems based on 5.7 above.</p> <p>Explain some applications of electrolysis in industry</p>				
GENERAL OBJECTIVE 6.0: Understand the laws governing chemical combination.						
7	<p>6.1 State the laws of:</p> <ul style="list-style-type: none"> • Conservation of matter • Gay lussac's • Avogadros 	<p>Explain the laws of:</p> <ul style="list-style-type: none"> • Conservation of matter • Gay lussac's • Avogadros 	<p>Text books</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Charts</p> <p>Computer</p>			

	6.2 Describe how to solve simple problems involving formulae, chemical composition and mole concept.	Explain how to solve simple problems involving formulae, chemical composition and mole concept.	Internet Projector			
GENERAL OBJECTIVE 7.0: Understand the change of matter and behaviour of gas under certain conditions.						
8	7.1 Explain the kinetic theory of matter in relation to: <ul style="list-style-type: none"> • The nature of solids, liquids and gases. • Diffusion of gasses • Melting and boiling points. 	Explain the kinetic theory of matter in relation to: <ul style="list-style-type: none"> • The nature of solids, liquids and gases. • Diffusion of gasses • Melting and boiling points. 	Text books Journals Whiteboard Marker Charts Computer Internet Projector			
	7.2 State the following gas laws qualitatively and mathematically – Boyles, Charle's, Graham's and Dalton's.	Explain the following gas laws qualitatively and mathematically – Boyles, Charle's, Graham's and Dalton's.				
	7.3 Derive the relationship between the vapour	Explain the relationship between				

	density of a gas and its relative molar mass. 7.4 Describe how to solve simple problems based on 7.2 above.	the vapour density of a gas and its relative molar mass. Explain how to solve simple problems based on 7.2 above.				
13.0 GENERAL OBJECTIVE 8.0: Understand the rate of chemical reaction and factors affecting rate of reaction.						
9	8.1 Differentiate between reversible and irreversible reactions. 8.2 List examples of reversible and irreversible reactions e.g. N_2O_4 , 2NO_2 . 8.3 Explain the term equilibrium. 8.4 State Le-chatelier's principle. 8.5 Outline the factors which can change the rate of a reaction viz:	Explain the difference between reversible and irreversible reactions Explain examples of reversible and irreversible reactions e.g. N_2O_4 , 2NO_2 . Explain the term equilibrium. Explain Le-chatelier's principle. Explain the factors which can change the rate of a reaction	Text books Journals Whiteboard Marker Charts Computer Internet Projector			

	<p>temperature, concentration, surface area, catalyst and pressure.</p> <p>8.6 Explain the manufacture of ammonia from N_2 and H_2 (Habers process) to illustrate the effects of changes in the factors mentioned in 8.5 above</p>	<p>viz: temperature, concentration, surface area, catalyst and pressure.</p> <p>Explain the manufacture of ammonia from N_2 and H_2 (Habers process) to illustrate the effects of changes in the factors mentioned in 8.5 above</p>				
GENERAL OBJECTIVE 9.0: Appreciate the usefulness of some non-metals like H_2 , Cl_2 and their compounds.						
10	<p>9.1 Describe the laboratory preparation of the following non-metals:</p> <ul style="list-style-type: none"> Cl_2 H_2 O_2 <p>9.2 Describe the industrial preparation of non-metals mentioned in 10.1 above.</p>	<p>Explain the laboratory preparation of the following non-metals:</p> <ul style="list-style-type: none"> Cl_2 H_2 O_2 <p>Explain the industrial preparation of non-metals mentioned in 10.1 above.</p>	<p>Text books</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Charts</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>			

	9.3 List the physical and chemical. Properties of the following non-metals: Cl ₂ , O ₂ , C, S, N ₂ .	Explain the physical and chemical. Properties of the following non-metals: Cl ₂ , O ₂ , C, S, N ₂ .				
	9.4 List the uses of the non-metals and their compounds.	Explain the uses of the non-metals and their compounds.				
GENERAL OBJECTIVE 10.0: Appreciate the importance of some metals like Cu, Al, Na etc. in our society.						
11	10.1 Describe the methods of extracting metals from their ores.	Explain the methods of extracting metals from their ores.	Text books Journals Whiteboard Marker Charts Computer Internet Projector			
	10.2 List the properties of metals in 11.1 above.	Explain the properties of metals in 11.1 above				
	10.3 List the uses of the metals and their compounds.	Explain the uses of the metals and their compounds.				
General Objective 11.0: Appreciate the importance of carbon compounds in our society.						
12-13	11.1 Explain the tetrahedral arrangement of carbon bonds.	Explain the tetrahedral arrangement of carbon bonds	Text books Journals Whiteboard Marker Charts	Illustrate the tetrahedral arrangement of carbon bonds.	Guide students to: Illustrate the tetrahedral arrangement of carbon bonds.	Building sticks

11.2 Explain the characteristics functional group of each class of organic compound.	Explain the characteristics functional group of each class of organic compound.	Computer Internet Projector			
11.3 State the general formula of each class of compounds.	Explain the general formula of each class of compounds				
11.4 Explain the naming of members of each series.	Explain the naming of members of each series.				
11.5 Describe the general method for preparing carbon compound	Explain the general method for preparing each of the compound.				
11.6 State the physical properties of the compounds.	Explain the physical properties of the compounds.				
11.7 Describe the reaction of each class of compound.	Explain the reaction of each class of compound.				

	11.8 Outline the usefulness of the compounds.	Explain the usefulness of the compounds.				
GENERAL OBJECTIVE 12.0: Conduct laboratory work based on the above topics.						
14-15	12.1 Explain how to clean laboratory apparatus and glass wares	Explain how to clean laboratory apparatus and glass wares	Text books Journals Whiteboard Marker Charts Computer Internet Projector	Clean laboratory equipment's and apparatus	Guide student to: Clean laboratory equipment's and apparatus.	PPE Laboratory equipment
	12.2 Explain the effect of heat and reagent on substances.	Explain the effect of heat and reagent on substances.		Dry the apparatus either in hot oven or in air.	Dry the apparatus either in hot oven or in air.	Oven Sample reagents
	12.3. Explain how to prepare simple laboratory solutions	Explain how to prepare simple laboratory solutions		Observe the effects of heat and reagents on substances.	Observe the effects of heat and reagents on substances.	

EVALUATION

C/A:60%

EXAMS: 40%

TOTAL: 100%

PROGRAMME: HIGHER NATIONAL DIPLOMA (HND)		
COURSE TITLE: General Biology II	COURSE CODE: STB 301	CONTACT HOURS: 4
	CREDIT UNITS: 3	THEORETICAL: 2
YEAR: I SEMESTER: 0	PRE-REQUISITE:	PRACTICAL: 1
GOAL: This course is designed to equip the students in (non-science programmes) with the knowledge of the fundamental aspects of biological science.		
GENERAL OBJECTIVES: On completion of this course, the students should be able to: <ol style="list-style-type: none"> 1.0 Understand cell structure 2.0 Understand the types of plant and animal tissues 3.0 Appreciate the classification of living organisms. 4.0 Know the principal features of different groups in the plant kingdom 5.0 Understand the distinguishing characteristics of the different groups of animal kingdom 6.0 Understand the structure and life history of the following plants and animals:- <ol style="list-style-type: none"> i) An Alga ii) An angiosperm iii) A protozoon iv) A vertebrate 7.0 Understand the internal structure of a flowering plant 8.0 Understand the general arrangement of the internal organs in a mammal. 9.0 Understand methods of nutrition. 10.0 Understand the transport of substances in plants and animals. 11.0 Know the process and significance of respiration. 12.0 Know the excretory organs and products. 13.0 Understand the general plan of the skeleton in a mammal. 14.0 Know the reproductive structures in plants and animals. 15.0 Understand the principal sense organs and their functions. 16.0 Understand the relationship of organisms to their environment. 17.0 Understand the constituents and characteristics of soil. 18.0 Know the methods of conserving soil fertility. 		

19.0 Know the development and method of cultivation of maize.

20.0 Understand how character is inherited.

NATIONAL BOARD FOR TECHNICAL EDUCATION

PROGRAMME: HIGHER NATIONAL DIPLOMA						
COURSE TITLE: General Biology II			COURSE CODE: STB 301		CONTACT HOURS: 4	
			CREDIT UNITS: 3		THEORETICAL: 2	
YEAR: I SEMESTER: 0			PRE-REQUISITE:		PRACTICAL: 1	
COURSE SPECIFICATION: THEORETICAL AND PRACTICAL						
GOAL: This course is designed to equip the students in (non-science programmes) with the knowledge of the fundamental aspects of biological science.						
GENERAL OBJECTIVE 1.0: Understand cell structure						
	THEORETICAL CONTENT				PRACTICAL CONTENT	
Week	Specific Learning Outcome	Teacher’s Activities	Resources	Specific Learning Outcome	Teacher’s Activities	Resources
	1.1 Explain Cell theory and its discovery	Explain cell theory and its discovery	Textbooks Journals Whiteboard Marker Computer Internet Projector	Prepare cell slides using onion	Guide student to: Prepare cell slides using onion	Microscope Onion
	1.2 Mention some scientist associated with cell discovery	Explain some scientist associated with cell discovery		Identify an animal cell using a microscope	Identify an animal cell using a microscope	Knife Slides
	1.3 Describe the structure of a plant cell as seen under a light microscope	Explain the structure of a plant cell as seen under a light microscope		Draw and label a plant cell and an animal cell	Draw and label a plant cell and an animal cell	Cover slip Epithelium cell from the cheek
	1.4 Explain the structure of an animal cell.	Explain the structure of an animal cell.				
	1.5 State the differences between a plant cell and animal cell	Explain the differences between a plant cell and animal cell				

GENERAL OBJECTIVE 2.0: Understand the different types of plant and animal tissues

2.1 Defines a tissue.	Explain a tissue	Textbooks Journals	Draw and label the tissues	Guide students to	Permanent slides
2.2 Name 4 different types of plant tissues.	Explain 4 different types of plant tissues	Whiteboard Marker Computer Internet Projector	given permanent slide	Draw and label the tissues given permanent slide	Microscope Magnifying lens
2.3 State the functions of each of the tissue.	Explain the functions of each of the tissue				
2.4 Distinguish between the different animal tissues.	Distinguish between the different animal tissues.				
2.5 Give the names of the different animal tissues	Explain the names of the different animal tissues				
2.6 List the functions of 2.5 above.	Explain the functions of 2.5 above.				

GENERAL OBJECTIVE 3.0: Appreciate the necessity for classification of living organisms.

3.1 Explain Taxonomy	Explain taxonomy	Textbooks Journals			
3.2 Explain the rules of nomenclature	Explain the rules of nomenclature	Whiteboard Marker Computer Internet Projector			
3.3 Explain the necessity for classification of living organisms.	Explain the necessity for classification of living organisms.				
3.4 Explain binomial	Explain binomial				

	nomenclature: Features of kingdom, phylum/division, subphylum/subdivision, class, sub class, superclass, order, sub order, genus and species	nomenclature: Features of kingdom, phylum/division, subphylum/subdivision, class, sub class, superclass, order, sub order, genus and species				
GENERAL OBJECTIVE 4.0: Know the classification and principal features of different groups in the plant kingdom						
	4.1 List the major groups in the plant kingdom.	Explain the major groups in the plant kingdom.	Textbooks Journals Whiteboard	Observe, draw and label plant specimens.	Guide student to: Observe, draw and label plant specimens.	Plant Specimen
	4.2 Describe the general characteristic of the major groups of plants.	Explain the general characteristic of the major groups of plants.	Marker Computer Internet Projector			Writing materials
	4.3 Classify some common examples of plants	Explain some common examples of plants				
	4.4 State three reasons for the classifications of plants.	Explain three reasons for the classifications of plants.				
GENERAL OBJECTIVE 5.0: Understand the classification and distinguishing characteristics of the different groups in the animal kingdom						
	5.1 Discuss the different phyla in the animal kingdom	Explain the different phyla in the animal kingdom	Textbooks Journals Whiteboard Marker			
	5.2 Explain the distinguishing characteristics of major	Explain the distinguishing characteristics of major groups of animals.	Computer Internet Projector			

	groups of animals.					
	5.3 Classify some examples of animals	Explain some examples of animals				
GENERAL OBJECTIVE 6.0: Understand the structure and life history of the following plants and animals:-i) An Alga, ii) An angiosperm, iii) A protozoon, iv) A vertebrate						
	6.1 Explain the structures of spirogyra	Explain the structures of spirogyra	Textbooks Journals Whiteboard Marker Computer Internet Projector			
	6.2 Describe the methods of reproduction in spirogyra	Explain the methods of reproduction in spirogyra				
	6.3 Outline the differences between a dicot and a monocot plant in external features	Explain the differences between a dicot and a monocot plant in external features				
	6.4 Explain the methods of reproduction in amoeba.	Explain the methods of reproduction in amoeba.				
GENERAL OBJECTIVE 7.0: Understand the internal structure of a flowering plant						
	7.1 Explain the internal structure of: • Root (Dicot and Monocot) • Stem (Dicot and Monocot)	Explain the internal structure of : • Root • Stem in a flowering plant	Textbooks Journals Whiteboard Marker Computer Internet Projector			
	7.2 List types of tissues in the	Explain types of tissues in				

	cross section of a stem	the cross section of a stem				
GENERAL OBJECTIVE 8.0: Understand the general arrangement of the internal organs in a mammal.						
	8.1 List Internal organs of a Mammal	Explain the internal organs	Textbooks Journals Whiteboard			
	8.2 State the functions of the different organs in 8.1 above	Explain the functions of the different organs in 8.1 above	Marker Computer Internet Projector			
GENERAL OBJECTIVE 9.0: Understand the different methods of nutrition.						
	9.1 Explain Nutrition in plant and Animal	Explain Nutrition in plant and Animal	Textbooks Journals Whiteboard			
	9.2 Explain the different methods of nutrition	Explain the different methods of nutrition	Marker Computer Internet Projector			
	9.3 List the macro and micro elements necessary in plants	Explain the macro and micro elements necessary in plants				
	9.4 Define photosynthesis	Explain photosynthesis				
	9.5 Explain the conditions necessary for photosynthesis	Explain the conditions necessary for photosynthesis				
	9.6 Describe the process of photosynthesis	Explain the process of photosynthesis				

	<p>9.7 Define digestion</p> <p>9.8 Explain types of Foods:</p> <ul style="list-style-type: none"> • Energy giving food • Body building food • Health preserving food • Balanced diet <p>9.9 State the functions of the liver in body metabolism</p>	<p>Explain digestion</p> <p>Explain the food substances</p> <ul style="list-style-type: none"> • Energy giving food • Body building food • Health preserving food • Balanced diet <p>Explain the functions of the liver in body metabolism</p>				
GENERAL OBJECTIVE 10.0: Understand the transport of substances in plants and animals.						
	<p>10.1 Define the following:</p> <ul style="list-style-type: none"> • Osmosis • Diffusion <p>10.2 Outline the significance of these processes in plants</p> <p>10.3 Explain transpiration</p> <p>10.4 Explain the structure of blood</p> <p>10.5 List the functions of blood</p>	<p>Explain, the following:</p> <ul style="list-style-type: none"> • Osmosis • Diffusion <p>Explain the significance of these processes in plants</p> <p>Explain transpiration</p> <p>Explain the structure of blood</p> <p>Explain the structure of blood</p>	<p>Textbooks</p> <p>Journals</p> <p>Whiteboard</p> <p>Marker</p> <p>Computer</p> <p>Internet</p> <p>Projector</p>	<p>Conduct an experiment to demonstrate osmosis in potato cells</p> <p>Conduct an experiment to demonstrate diffusion</p> <p>Compare osmosis and diffusion</p>	<p>Guide student to:</p> <p>Conduct an experiment to demonstrate osmosis in potato cells</p> <p>Conduct an experiment to demonstrate</p> <p>Compare osmosis and diffusion</p>	<p>Potato</p> <p>Salt</p> <p>1000ml beakers</p>

GENERAL OBJECTIVE 11.0: Understand the process and significance of respiration.

11.1 List the respiratory organs	Explain the respiratory organs	Textbooks Journals Whiteboard			
11.2 Outline the process of respiration and its significance	Explain the process of respiration and its significance	Marker Computer Internet Projector			
11.3 Explain the significance of 11.2 above	Explain the significance of 11.2 above				
11.4 Differentiate between aerobic and anaerobic respiration	Differentiate between aerobic and anaerobic respiration				
11.5 Explain the economic importance of fermentation	Explain the economic importance of fermentation				

GENERAL OBJECTIVE 12.0: Understand the excretory organs and products

12.1 Define Excretion	Explain excretion	Textbooks Journals Whiteboard			
12.2 Describe the excretory organs in a mammal	Explain the excretory organs in a mammal	Marker Computer Internet Projector			
12.3 List the excretory organs and excretory products in man	Explain the excretory organs and excretory products in man				
12.4 State the excretory products in a plant	Explain the excretory products in a plant				

GENERAL OBJECTIVE 13.0: Understand the general plan of the skeleton in a mammal

13.1.Explain the general Plan of Skeleton	Explain the general plan of the skeleton in a mammal	Textbooks Journals Whiteboard	Draw and label the structure of a vertebra e.g. a. Atlas b. Axis	Guide students to: Draw and label the structure of a vertebra e.g. a. Atlas b. Axis	
13.2 Explain types of skeleton in animals (exoskeleton and endoskeleton)	Explain types of skeleton in animals (exoskeleton and endoskeleton)	Marker Computer Internet Projector			
13.3 State the functions of the skeleton	Explain the functions of the skeleton				
13.4 Describe the general plan of the skeleton in a mammal	Explain the general plan of the skeleton in a mammal				
13.5 Describe the skeleton of the fore and hind limbs of a mammal	Explain the skeleton of the fore and hind limbs of a mammal				

GENERAL OBJECTIVE 14.0: Know the reproductive structures in plants and animals.

14.1 Explain Structure of a flower	Explain the structure of a flower	Textbooks Journals Whiteboard	Draw and Label the structure of a flower	Guide students to: Draw and Label the structure of a flower	Drawing materials
14.2 Differentiate insect and wind pollinated flower	Differentiate insect and wind pollinated flower	Marker Computer Internet Projector			
14.3 Differentiate between a staminate and pistilate	Differentiate between a staminate and pistilate				

flower	flower				
14.4 List the parts of a flower and their functions	Explain the parts of a flower and their functions				
14.5 Distinguish between pollination (self-pollinated flower and cross pollinated flower) and fertilization.	Explain between pollination and fertilization.				
14.6 Distinguish the process of fertilization in a flowering plant	Explain the process of fertilization in a flowering plant				
14.7 Distinguish between a grain, drupe and berry	Explain differences between grain, drupe and berry				
14.8 List the different methods of dispersal of fruits and seeds	Explain the different methods of dispersal of fruits and seeds				
14.9 Explain the structure of a seed and its germination of seed	Explain the structure of a seed and its germination of seed				
14.10 Explain dicotyledonous seed and	Explain dicotyledonous seed and				

	monocotyledonous seeds	monocotyledonous seeds				
	14.11 Explain germination of seeds (epigeal germination and hypogeal germination)	Explain germination of seeds (epigeal germination and hypogeal germination)				
	14.12 Explain factors affecting germination	Explain factors affecting germination				
GENERAL OBJECTIVE 15.0: Understand the principal sense organs and their functions.						
	15.1 Explain coordination	Explain coordination	Textbooks			
	15.2 Name the principal sense organs	Explain the principal sense organs	Journals			
	15.3 State the functions of 15.2 above	Explain the functions of 15.2 above	Whiteboard			
	15.4 Define hormones	Explain hormones	Marker			
	15.5 List types of Hormones	Explain types of Hormones	Computer			
	15.6 Enumerate the functions of hormones	Explain the functions of hormones	Internet			
	15.7 Explain the role of auxins in tropisms	Explain the role of auxins in tropisms	Projector			

General Objective 16.0: Understand the relationship of organisms to their environment.

16.1 Define the following ecological terms; <ul style="list-style-type: none">• Ecosystem• Community• Population• Habitat	Explain the following ecological terms; <ul style="list-style-type: none">• Ecosystem• Community• Population• Habitat	Textbooks Journals Whiteboard Marker Computer Internet Projector			
16.2 List the biotic and abiotic in a habitat	Explain the biotic and abiotic in a habitat				
16.3 Explain the following associates: <ul style="list-style-type: none">• Symbiosis• Parasitism• Commensalism• Saprophytism	Explain the following associates: <ul style="list-style-type: none">• Symbiosis• Parasitism• Commensalism• Saprophytism				
16.4 Describe the methods of studying an ecological habitat (e.g., quadrat)	Explain the methods of studying an ecological habitat				
16.5 Explain the causes and control of diseases such as malaria, bilharzia, cholera, river blindness and polio	Explain the causes and control of diseases such as malaria, bilharzia, cholera, river blindness and polio				

	16.6 Explain five pollutants of air and water	Explain five pollutants of air and water				
	16.7 Describe the methods of controlling pollution	Explain the methods of controlling pollution				
GENERAL OBJECTIVE 17.0: Know the constituents and characteristics of soil.						
	17.1 Explain soil Study	Explain soil Study	Textbooks Journals Whiteboard Marker Computer Internet Projector	Examine the texture of soil sample in the laboratory.	Guide student to: Examine the texture of soil sample in the laboratory.	Soil Sample pH meter Beaker
	17.2 List the constituents of soil	Explain the constituents of soil				
	17.3 Differentiate between different types of soil, e.g. sandy, clay, loamy.	Explain the differences between types of soil, e.g. sandy, clay, loamy.				
	17.4 Explain the characteristics of a good soil most favourable for crop cultivation	Explain the characteristics of a good soil most favourable for crop cultivation				
	17.5 Explain how soil organisms increase soil fertility	Explain how soil organisms increase soil fertility				
GENERAL OBJECTIVE 18.0: Understand the methods of conserving soil fertility.						
	18.1 Explain Soil conservation	Explain Soil conservation	Textbooks Journals Whiteboard Marker			
	18.2 List the ways in which soils lose fertility	Explain the ways in which soils lose fertility				

	18.3 Define Soil Erosion	Explain Soil Erosion	Computer Internet Projector			
	8.4 List types of soil erosion.	Explain types of soil erosion.				
	18.5 State the measures to check soil erosion	Explain the measures to check soil erosion				
	18.6 Outline the various ways to conserve soil fertility	Explain the various ways to conserve soil fertility				
GENERAL OBJECTIVE 19.0: Understand the development and method of cultivation of maize.						
	19.1 Discuss the type of soil preferred by maize	Explain the type of soil preferred by maize	Textbooks Journals Whiteboard			
	19.2 Explain the method of cultivation	Explain the method of cultivation	Marker Computer Internet Projector			
	19.3 Discuss major diseases which can affect maize	Explain major diseases which can affect maize				
	19.4 List how the diseases can be checked	Explain how the diseases can be checked				
GENERAL OBJECTIVE 20.0: Understand how character is inherited						
	20.1 Explain the following terms: <ul style="list-style-type: none"> Genetics, Heredity and 	Explain the following terms: <ul style="list-style-type: none"> Genetics, Heredity and 	Textbooks Journals Whiteboard Marker			

	<ul style="list-style-type: none"> Inheritance 	<ul style="list-style-type: none"> Inheritance 	Computer Internet Projector			
	20.2 Differentiate between Hereditary and Genetics	Explain the differences between Hereditary and Genetics				
	20.3 State Mendel's law	Explain Mendel's law				
	20.4 Explain mono-hybrid and dehybrid inheritance.	Explain mono-hybrid and dehybrid inheritance.				
	20.5 Define linkage and sex-linked characters	Explain linkage and sex-linked characters				
	20.6 Explain how sex-linked characters are inherited.	Explain how sex-linked characters are inherited.				
	20.7 Explain how to solve problems with 20.1	Explain how to solve problems with 20.1				

EVALUATION

C/A:60%

EXAMS: 40%

TOTAL: 100%

PRACTICAL MANUAL BASIC SCIENCE

GENERAL PHYSICS I

STP 101

1. Perform simple activities to demonstrate scientific principles:
 - Falling from a height
 - Floating of objects in water.
2. Carryout experiments to illustrate the following:
 - Circular motion
 - Rational motion
 - S.H.M
3. Demonstrate the methods of measurement of length time-using stop watch as in the application of simple pendulum and mass
4. Perform experiments in the measurement of mass using beam balances, replace length using vernier callipers and meter-ruler; time, using stop watch, simple pendulum.
5. Perform simple experiments to illustrate the principles of :
 - Operation of levers
 - Inclined planes
 - Pulleys
 - Hydraulic press
6. Demonstrate experimentally the effects of heat energy e.g.:
 - Ball and ring experiment
 - Melting of ice
 - Boiling of water to form steam.
7. Perform simple experiments to illustrate. Conversion of energy from one form to another

	<p>8. Perform experiments to illustrate conduction, convection and radiation:</p> <ul style="list-style-type: none"> • Conduction of heat through a metal rod, • Convection in air and radiation through different materials. <p>9. Illustrate experimentally the concept of waves using examples such as:</p> <ul style="list-style-type: none"> • Ripples on water • Waves on string • Helical string <p>10. Conduct some simple experiments to demonstrate properties of waves:</p> <ul style="list-style-type: none"> • Reflection • Refraction. <p>11. Measure angles of incidence and reflection</p> <p>12. Conduct simple experiment to demonstrate different sound produced by different instruments.</p> <p>13. Identify the different sounds produced by string, wind and percussion</p> <p>14. Perform simple experiments on reflection and refraction:</p> <ul style="list-style-type: none"> • Curved mirror experiment, • Refraction through a glass block, • Refraction through water <p>15. Perform experiment to illustrate dispersion of white light by prism and relate to its formation of rainbow.</p>
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	<p>16. Perform simple experiments to illustrate various effects of:</p> <ul style="list-style-type: none"> • Electric current, • Heating effects as in heating coil, • Electric bulbs • Sound effects as in bell, loud speaker, microphone <p>17. Perform simple experiment to show the relationship in above using torch bulb, wires and relevant instruments</p> <p>18. Insert a fuse in an electric current</p> <p>19. Fit a cable into a plug</p>
GENERAL PHYSICS II STP 301	<p>1. Perform simple experiments to measure velocity and speed</p> <p>2. Solve problems using equations derived in 3.2 above</p> <p>3. Solve simple problems on motions</p> <p>4. Conduct simple experiments on the following :</p> <ul style="list-style-type: none"> • Elasticity, • Stress • Strain • Elastic limit <p>5. Demonstrate the application of elasticity in day-to-day life.</p> <p>6. Use the goldleaf electroscope to determine the nature of charges on different bodies.</p>

	<p>7. Perform some laboratory activities to generate some static electricity such as:</p> <ul style="list-style-type: none"> • Rubber balloon and glass rod experiments • Metal sphere experiment etc. <p>8. Conduct simple experiments in electricity such as:</p> <ul style="list-style-type: none"> • Light bulb experiment • Resistor experiment etc. <p>9. Conduct experiments in thermometric liquids and thermometers such as:</p> <ul style="list-style-type: none"> • Thermal expansion experiment • Calibration of thermometers • Thermometer comparison experiment <p>10. Conduct simple experiments on water wave such as:</p> <ul style="list-style-type: none"> • Wave generation, • Reflection and Refraction of waves.
GENERAL CHEMISTRY I STC 101	<p>1. Carry out the electrolysis of water using Hoffman Voltmeter to identify the constituents of water</p> <p>2. Carry out distillation of water to produce distilled water which can be used for routine experiments in laboratories and charging of batteries.</p> <p>3. Carry out simple test for acids, bases and salts.</p> <p>4. Use indicators to distinguish between acids, bases and salts.</p> <p>5. Carry out acid base titration.</p>

	6. Obtain salt such as sodium chloride from sea water by evaporation process. 7. Carry out an experiment to produce soap from palm oil and potash rich ashes. 8. Carryout an experiment to produce ethanol through the fermentation of locally available cereals.
GENERAL CHEMISTRY II STC 301	1. Conduct some simple experiments of water quality (physical and chemical). 2. Conduct some experiments on acids, bases and salts (for example acid-base neutralisation, preparation of salts e.g. sodium chloride and copper sulphate) 3. Illustrate the tetrahedral arrangement of carbon bonds. 4. Clean laboratory equipment's and apparatus. 5. Dry the apparatus either in hot oven or in air. 6. Observe the effects of heat and reagents on substances.
GENERAL BIOLOGY I STB 101	1. Observes some cells under the microscope (e.g. a plant cell, and animal cell) 2. Draw and label the cells observed in above practical 3. Draw and label the digestive system in man 4. Demonstrate osmosis and diffusion by simple experiments. 5. Draw and label the structure of a mammalian heart.
GENERAL BIOLOGY II STB 301	1. Prepare cell slides using onion

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| | <ol style="list-style-type: none">2. Identify an animal cell using a microscope3. Draw and label a plant cell and an animal cell4. Draw and label the tissues given permanent slide5. Observe, draw and label plant specimens.6. Conduct an experiment to demonstrate osmosis in potato cells7. Conduct an experiment to demonstrate diffusion8. Compare osmosis and diffusion9. Draw and label the structure of a vertebra e.g. a. Atlas b. Axis10. Draw and Label the structure of a flower11. Examine the texture of soil sample in the laboratory.12. Determine the pH of the soil sample |
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LIST OF EQUIPMENT FOR BASIC SCIENCE

S/N	EQUIPMENT	QUANTITY
1	1000ml beakers	
2	Acid	Assorted
3	Balance	2
4	Ball	Assorted
5	Bases	
6	Beaker	
7	Beam balances	
8	Beating drum	
9	Bells	
10	Building sticks	
11	Bunsen Burner	
12	Burette	
13	Cable	Assorted
14	Car battery	Assorted
15	Circuit	
16	Clamp	
17	Colours	Assorted
18	Copper sulphate	
19	Cover slip	
20	Curved mirror	
21	Distillation Apparatus	
22	Distiller	
23	Drawing materials	Assorted
24	Dropper	

25	Elastic band	
26	Electric bulbs	Assorted
27	Epithelium cell from the cheek	
28	Evaporating dish	
29	Floating objects	Assorted
30	Free-falling objects	Assorted
31	Fuse	Assorted
32	Gas	
33	Glass block	
34	Glass rod	
35	Goldleaf electroscope	
36	Heating coils	
37	Heating mantel	
38	Hoffman Voltmeter	
39	Hydraulic press	
40	Inclined planes	
41	Indicator	
42	Knife	Assorted
44	Lever	
45	Light bulb	Assorted
46	Litmus paper	
47	Loud speaker	1
48	Magnifying glass	
49	Magnifying lens	
50	Measuring Tape	Assorted
51	Metal rod	
52	Metal sphere	
53	Meter ruler	

54	Microphone	1
55	Microscope	
56	Millet	Assorted
57	Oil	Assorted
58	Onion	Assorted
59	Oven	2
60	Perfume	Assorted
61	Permanent slides	
62	pH meter	
63	Pipette	
64	Plant Specimen	Assorted
65	Plug	
66	Potato	Assorted
67	PPE	Assorted
68	Prepared slide	
69	Prism glass	
70	Pulleys	
71	Pyrex glass	
72	Resistor	
73	Retort Stand	
74	Ring	
75	Rubber balloon	Assorted
76	Ruler	Assorted
77	Salt	Assorted
78	Sample reagents	Assorted
79	Sea water	Assorted
80	Shallow Basin	1
81	Simple pendulum	Assorted

82	Slides	Assorted
83	Soap	Assorted
84	Sodium chloride	
85	Soil Sample	Assorted
86	Spatula	
87	Speedometer	
88	Stirrer	
89	Stop-watch	
90	Strings	
91	Thermometer	
92	Thermometric liquids	
93	Touch Light bulbs	
94	Tripod Stand	
95	Tripod Stand Wire Gauze	
96	Vernier callipers	
97	Volumetric Flask	
98	Water bath	
99	Wire gauze	
100	Wires	Assorted
101	Writing materials	Assorted