

CURRICULUM AND COURSE SPECIFICATIONS FOR

NATIONAL DIPLOMA

IN

VETERINARY LABORATORY TECHNOLOGY
PLOT B, BIDA ROAD, P M B 2239, KADUNA, NIGERIA

2018

GENERAL INFORMATION

1.0 TITLE OF THE PROGRAMME

The programme is National Diploma in Veterinary Laboratory Technology

2.0 PROGRAMME GOAL

The goal of the National Diploma (ND) in **Veterinary Laboratory Technology** programme is to produce diplomatescapable of providing clinical and laboratory support to veterinarians and registered Veterinary Para-Professionals.

3.0 PROGRAMME OBJECTIVES

On completion of the programme, the Diplomates should be able to:

- i. Assist anesthetists, calculate drug dosages, and assist in invasive medical procedures.
- ii. Carry out sample collection and processing under the supervision of a veterinarian.
- iii. train diplomates in the field of veterinary laboratory technology who can effectively utilize their training to establish and manage their own veterinary laboratories.
- iv. Assist in management of veterinary and diagnostic laboratories, livestock farms and associated production processes.
- v. Assist in provision of animal care services in veterinary hospitals/clinics, research institutions, farms and other livestock facilities.

4.0 ENTRY REQUIREMENTS INTO THE PROGRAMME

NATIONAL DIPLOMA (ND)

The academic requirements for admission into the ND programmein Veterinary Laboratory Technology are:

I. Five (5) credit passes in GCE "O" level or Senior Secondary Certificate (SSCE), NECO NABTEB at not more than two sittings. The five subjects must include Mathematics, English language and any three (3) others chosen from; Physics, Chemistry and Biology/Health Science/Agricultural Science/Animal Husbandry.

II. Admission will be through the Joint Admission and Matriculation Board (JAMB) having met condition I above and scored up to the requisite cut-off point

5.0 PROGRAMME DURATION:

The ND programme runs for two academic sessions of two semesters each (four semesters).

6.0 CURRICULUM

- 6.1 The curriculum is structured into four semesters of classroom, studio, laboratories and other field activities. Each semester of educational based activities shall be for a duration of 17 weeks distributed as follows:
- (i.) 15 contact weeks of teaching, i.e. theory, practical, exercises, quizzes, tests, etc.; and
- (ii.) 2 weeks for examination and registration.
- 6.2 The curriculum of the programme consists of four main components viz:
 - i) General Studies Courses
 - ii) Foundation Courses
 - iii) Professional Courses
 - iv) Supervised Industrial Work Experience Scheme (SIWES)

The General Studies component includes courses in General Sciences, English Language/Communication, Entrepreneurship Development and Computer Studies.

The General Education Courses shall account for not more than 10-15% of the total contact hours for the programme.

Foundation Courses:Research Methods, Law courses etc. The number of hours for the foundation courses shall be between 10-15% of the total contact hours for the programme.

Professional Courses are specialised core courses, which give the student the theory and practical skills he/she needs to practice in his/her field of specialisation at the technical level.

Those specialised core courses account for between 60-70% of the total contact hours of the programme.

7.0 PROJECT

Every ND student is required to successfully complete a project in the field of **Veterinary Laboratory Technology**during his/her final year on the programme. Adequate project supervision/assessment should be done by qualified lecturers.

8.0 EVALUATION

For the purpose of awarding the National Diploma (ND) in **Veterinary Laboratory Technology**Certificate, evaluation of the student's work shall include: examinations, project and course work (tests, quizzes, practical etc.). The weighting of each of the components of the evaluation system shall be prescribed by the National Board for Technical Education and the institutions running the programme.

9.0 CONDITIONS FOR THE AWARD OF THE NATIONAL DIPLOMA

9.1 The National Board for Technical Education (NBTE) shall accredit the ND programme in **Veterinary Laboratory Technology** before the award of the diploma certificate. Details about the process of accrediting programmes for the award of National Diploma shall be in line with guidelines from the Executive Secretary, (NBTE), Plot B, Bida Road, P.M.B 2239, Kaduna Nigeria or www.nbte.gov.ng.

- 9.2 The institution will award the National Diploma to candidates who successfully completed the programme after passing the prescribed coursework, examination, and diploma project. Such candidates should have completed between 72-80 semester credit units as prescribed in the programme.
- 9.3 The National Diploma (ND) **Veterinary Laboratory Technology**shall be awarded to only persons who have satisfied all conditions for the award of the certificate as laid down by the National Board for Technical Education (NBTE) and the institution in which the programme is offered.

10.0 CLASSIFICATION/GRADING SYSTEM OF DIPLOMAS:

National Diploma (ND) shall be based on a total scale of 4 points classified into the following categories:

 Distinction
 CGPA of 3.50 and above

 Upper Credit
 CGPA of 3.00 – 3.49

 Lower Credit
 CGPA of 2.50 – 2.99

 Pass
 CGPA of 2.00 – 2.49

MARKED RANGE	LETTER GRADE	WEIGHTING
75% and Above	A	4.00
70% - 74%	AB	3.50
65% - 69%	В	3.25
60% - 64%	ВС	3.00
55% - 59%	С	2.75
50% - 54%	CD	2.50

45% - 49%	D	2.25
40% - 44%	Е	2.00
Below 40%	F	0.00

11.0 ACCREDITATION OF PROGRAMMES

This programme shall be accredited by the National Board for Technical Education (NBTE) in conjunction with Veterinary Council of Nigeria (VCN).

12.0 GUIDANCE NOTES FOR TEACHERS OF THE PROGRAMME

- 12.1 The curriculum is drawn in course units. This is in keeping with the provision of the National Policy on Education, which stresses the need to introduce semester credit unit to enable a student who wishes to transfer the units already completed in an institution of similar standard from which he is transferring.
- 12.2 In designing the units, the principles of the modular system by product has been adopted, thus making each of the professional modules, when completed, provide the student with technical operative skills, which can be used for employment purposes.
- 12.3 As the success of the credit unit system depends on the articulation of programmes in the institutions and industry, the curriculum content has been written in behavioural objectives, so that it is clear to all, the expected performance of the student who successfully complete the programme.
- 12.4 The teaching of the theory and practical work should, as much as possible, be integrated. Practical exercises especially those in professional courses and laboratory work should, as much as possible, be integrated to a ratio of 70:30.

CURRICULUM TABLES FOR NATIONAL DIPLOMA IN VETERINARY LABORATORY TECHNOLOGY

NDI SEMESTER ONE

COURSE CODE	COURSE TITLE	L	P	CU/CH	TOTAL HOURS/ SEM
VLT 111	Introduction to Veterinary Laboratory Technology	1	0	1	15

VLT 112	Laboratory Animal Studies	1	2	3	45
VLT 113	Veterinary Medical Terminology	1	0	1	15
VLT 114	Animal Handling and WelfareI	2	2	4	60
VLB 111	Biology (Morphology and Physiology of Living Organisms)	2	3	5	75
VLC 111	Inorganic Chemistry	1	2	3	45
VLP 111	Mechanics and Properties of Matter	2	3	5	75
GNS 101	Use of English		0	2	30
GNS 111	Citizenship Education I		0	2	30
		12	9	21	390

NDI SEMESTER TWO

COURSE CODE	COURSE TITLE	L	P	CU/CH	TOTAL HOURS /SEM
VLT 121	Introduction to Animal Diseases	1	1	2	30
VLT 122	Veterinary Medicinal Products & Drug Use	1	1	2	30
VLT 123	Animal Anatomy and Physiology	2	2	4	60
VLT 124	Veterinary Laboratory Procedures I (Instrumentation)	1	2	3	45
VLT 125	Animal Handling and Welfare II	1	2	3	45
VBC121	Introductory Biochemistry	1	2	3	45
EeD 126	Introduction to Entrepreneurship	1	2	3	45
GNS 102	Communication in English	2	0	2	30
GNS 121	Citizenship Education II	2	0	2	30
VLC 121	Organic Chemistry	1	2	3	45
VLP 121	Optics, Waves and Sound, Electromagnetic Radiation			3	45
					435

ND II SEMESTER ONE

COURSE CODE	COURSE TITLE	L	P	CU/CH	TOTAL HOURS /SEM
VLT 211	Introductory Veterinary Pathology	2	2	4	60
VLT 212	Introduction to Veterinary Microbiology			4	60
VLT 213	Introduction to Veterinary Surgery		2	4	60
VLT 214	Veterinary Nursing	1	2	3	45

VLT 215	Veterinary Laboratory Procedures II	2	2	4	60
	(Care and Maintenance of Equipment)				
VLT 216	Oral/Interpersonal Communication	1	0	1	15
VLT 217	Rural Sociology	1	0	1	15
VLT 218	Research Methodology (Bio Statistics)	2	0	2	30
					345

ND II SEMESTER TWO

COURSE CODE	COURSE TITLE	L	P	CU/CH	TOTAL HOURS /SEM
VLT 221	Record Keeping and Data Management	1	0	1	15
VLT 222	Seminar	0	1	1	15
VLT 223	Project	0	3	3	45
VLT 224	Clinical Posting	0	36	36	360 (10wks)
					435

ND I

FIRST SEMESTER

2.0 Understand common terminologies in veterinary laboratory technology

3.0 Know the roles of veterinary laboratory technologist in veterinary services					
PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY					

COURSE TITLE: IN VETERINARY LABO		LOGY	Cours	se Code: VLT 111	Contact Hou	rs: :- 1	hr/wk(15hr/w	k)
COURSE SPECIFICA	TION: Theoretical co	ontent		Pra	ctical Content:			
General Objective: 1.0		veterinary la	aborator	ry technology				
Specific Learnin	O I	Learning			Teachers	Lea	rning	Evaluation
Objectives:	Activities	Resources		Learning Objectives:	Activities	Res	ources	
1.1 Define veterina		Books,						Ask questions
laboratory technology	veterinary	journals,						Class
1.2 Explain the history		periodicals	8,					assessment
veterinary laborato	ry technology is	internet,						Test
technology		power poin	nt.					Assignment
1.3 Explain the curre								
status of veterina	ry							
laboratory technology	T. 1 4 1	4 • 1	•	1	1 1			
General Objective: 2.0					Teachers Act		Lagunina	Evaluation
Specific Learnin Objectives:	Activities	Learning Resource	,	Specific Learning	Teachers Act	ivities	Learning Resources	Evaluation
·		Resource	es	Objectives:			Resources	
2.1 List relevant terminologies used in								Examination
veterinary laborator								Quiz
technology 2.2 Explain eac	h							Assignment
terminology listed								Test
General Objective: 3.0 Know the role of veterinary laboratory technology in Veterinary services.								
Specific Learning	Teachers Activities	Learning		Specific	Teachers A	ctivities	Learning	Evaluation
Objectives:		Resourc	es	Learning			Resources	
				Objectives:				
3.1 Explain the	Give examples to						1	Test
sequence of	students.	Textbool	,					
arrangement of		journals,						
-		manuals						

equipment in the	Explain how			Report
Laboratory.	consumables in the			writing
3.2 Outline how	laboratory are taken.			
consumables in				
the laboratory are				
taken and updated				
on routine basis.				
3.3 Explain how to				
document				
samples				
examined on				
daily basis.				

PR	PROGRAMME: NDVETERINARY LABORATORY TECHNOLOGY										
CO	COURSE TITLE: LABORATORY ANIMAL STUDIES										
CO	COURSE CODE: VLT 112										
DURATION Lecture:- 1 hr Tutorial:- Hours Practical:- 2hours Total:- 3Hrs/Wk(45 hours/sen											
CR	EDIT UNITS: 3	CU		•							
GO	AL: This course	is designed to train students to acqu	uire knowledge of managing la	aboratory animals							
GE	NERAL OBJEC	TIVES: On completion of the cour	rse, the student should be able	to:							
1.0		e concept of the use of animals in	•								
2.0	Know commo	n animals that are used in the lab	oratory								
3.0	Know how to	manage common laboratory anim	nals								
4.0	Know commo	n signs of distress and disease in	laboratory animals								

COURSE TITLE: LABORATOR	Y ANIMAL STUDII	ES	Course	Code: VLT 112	Con	tact Hours: 3 ho	ours/week	
COURSE SPECIFICATION: The	oretical content		Practical Content:					
General Objective: 1.0: Understand the concept of the use of Animals in the Laboratory								
Specific Learning Objectives:	Teachers Activities	Learning Resour	rces	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation	
1.1 Explain Laboratory Animals	Define Laboratory Animals to students	Text books Journals Pictures Internet					Tests Assignment Examination Quizzes	
1.2 Outline the brief history of the use/study of Animals in the laboratory1.3 State the rationale for use of laboratory animals	Explain the history of Laboratory animals							
1.4 State the current trends in the use of Laboratory Animals	Explain to students rationale for use of animals in the laboratory.							
	Explain new trends including challenges in the use of laboratory animals							

Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
2.1 List the species and breeds of common Laboratory Animals	Explain to students different species and breeds of animals used in the Laboratory	Text books Journals Pictures Internet	Identify various laboratory animals and their breeds	Guide students to identify various laboratory animals and their breeds.		Practical Note books
2.2 Describe the features of the various animals				then breeds.		
2.3 Differentiate between the different breeds	Explain the distinguishing features of various laboratory animals and their breeds					
General Objective: 3.0 Know how	to manage common	Laboratory Animals				
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
3.1 Explain how to manage common Laboratory Animals with emphasis on: -Housing -Feeding -Watering - Breeding	Lecture students on how to manage Laboratory Animals with emphasis on housing, feeding and watering and breeding	Text books Journals Pictures Internet	3.1 carryout: -Feeding -Housing -Watering of different Laboratory Animals	Guide students on how to feed, house and water different laboratory animals	Laboratory animals facilities	-Quiz -Test -Assignment -Practical Examination
3.2 List equipment required for management of common Laboratory Animals	Describe to students various housing, feeding					

	and watering utensils for laboratory animals					
General Objective: 4.0 Know com Specific Learning Objectives:	mon signs of Distress Teachers Activities	and Disease in Laboratory A Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
4.1 Explain signs of distress in Laboratory Animals	Define Distress	Text books Journals Pictures				Quiz -Test
4.2 Outline sources of distress in Laboratory Animals -Heat (weather) -injury -Diseases -Attack by predators	Explain the various sources of distress to animals	Internet				-Assignment -Practical Examination
-Improper handling etc 4.3 Outline basic steps to be taken to alleviate distress in Laboratory animals	Explain to students steps to be taken to alleviate stress in Laboratory Animals					

PROGRAMME: N	ND VETERINARY LABORAT	TORY TECHNOLOGY		
COURSE TITLE:	VETERINARY MEDICAL T	ERMINOLOGY		
COURSE CODE:	VLT 113			
DURATION	Lecture:- 15 hrs	Tutorial:- Hours	Practical:- hours	Total:- 15 hrs/Semester
CREDIT UNITS:	1	-1	- 1	'
GOAL: This cour	se is designed to acquaint the	students with terminologie	s used in the laboratory	
GENERAL OBJE	CTIVES: On completion of the	he course, the student shou	lld be able to:	
1.0 Appreciate the	compound nature of the termin	ologies		

COURSE TITLE: VETERINAF	RY MEDICAL TER	MINOLOGY.	Course	Code: VLT 113	Contact Hours:		
COURSE SPECIFICATION: T	Theoretical content			Practical Content:			ıt:
General Objective 1.0: Apprecia	te the compound na	ture of the termin	nologies	•			
Specific Learning Objectives:	Teachers				Teachers	Learning	Evaluation
	Activities			Learning Objectives:	Activities	Resources	
1.1 Identify the terminology	Motivate the	Medical diction	nary				Continuous
cystoscopy	students to pick						Assessment
1.2 Break up the word into	on any	Medical journa	ls				
component parts	terminology for						End of
- cysto	break down	Medical Text b	ooks				semester test
- Scopy							
		Power point					
1.3 Define each component							
- Cysto							
- Scopy							
- Bladder							
- View							
1.4 Give the meaning of							
terminology –							
View the bladder							

PROGRAMME: N	ND VETERINARY LABORA	TORY TECHNOLOGY		
COURSE TITLE:	ANIMAL HANDLING AND) WELFARE I		
COURSE CODE:	VLT 114			
DURATION	Lecture:- 2 hrs	Tutorial:- Hours	Practical:- 2hrs	Total:- 60 hrs/Semester
CREDIT UNITS:	4	•		·

GOAL: This course is designed to expose students to safe handling and restraint of animals

GENERAL OBJECTIVES: On completion of the course, the student should be able to:

- 1.0 Know how to safely handle, restraint common laboratory, companion and food animals
- 2.0 Know how to provide core to common laboratory, companion and food animals and identify when they stressed
- 3.0 Know the health care needs of common laboratory, companion and food animals and risk involved during interaction with them
- 4.0 Know how to undertake safety transportation within and outside the laboratory/hospital environment and home slaughter of the food animals

PROGRAMME: ND VETERI	NARY LABORATORY	TECHNOLOGY				
COURSE TITLE: ANIMAL H	IANDLING AND WELI	FARE I C	ourse Code: VLT 114	Contact I	Hours: 60 hrs/S	Semester
COURSE SPECIFICATION:	Theoretical content	<u> </u>		Practi	ical Content:	
General Objective1.0: Know h	ow to safely handle, rest	raint common labora	tory, companion and fo	od animals		
Specific Learning Objectives:	Teachers Activities	Learning Resourc	es Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
Explain how common laboratory, companion and food animals are handled and restrained Outline basic equipment and facilities for restrain and handling of laboratory	Explain how common laboratory, companion and food animals are handled and restrained List basic equipment and facilities used for handling and restrain	Text books Journals Pictures Internet	Handle and restrain all species of animals among laboratory companion and food animals Display all equipment and facilities used in restrain and handling Display better housing feeding and watering	Show and demonstrate to students how animal should be handled and restrained Demonstrate/guide how all equipment and facilities for restraint and handling are used		
General Objective 2.0:Know h			•			1
Specific Learning Objectives:	Teachers Activities	Learning Resourc	es Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
2.1 Discuss how laboratory, companion and feed animals are catered for with all available	Explain to students how animals should be kept homely	Text books Journals	Describe the pain, fear, discomfort in animal	Take students to demonstration farms and identify the idle condition	Manuals User guide	Assignment Test, Quizzes, Examination
		Pictures		and the for	Demos	

requirements (housing, feeding watering) 2.2 Explain how stress, fear, pain and discomfort can be identified in companion, laboratory and food animals 2.3 Discuss how to alleviate, control or improve on conditions in 2.2 above						
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning	Teachers Activities	Learning Resources	Evaluation
			Objectives:			
3.1 Explain basic health care needs of laboratory, companion and food animals3.2 Describe the risks involved during interaction with laboratory, companion and food animals	Describe to students basic health care needs Indicate to students dangers involved during interaction with animals	Text books Journals Pictures Internets	Describe basic health care needs Describe how to appraise animals	Show how basic health care needs are provided Show students direction to approach or how to handle animals	Farms Stables Kernels	Assignment Test, Quizzes, Examination (theory and practicals)
3.3 List basic first aid facilities precautionary measures for emergency situations in handling and restrain	Describe to students first aid measures precautions and its components		Describe the use of first aid and precautionary measures	Show students all first aid facilities and precautionary measures available and how to apply		
General Objective 4.0:Know h	now to undertake safety to	ransportation within and ou	itside the laborate	ory/hospital environm	ent and home	slaughter of the

food animals

Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
4.1 Discuss how to undertake transportation within and outside the laboratory/hospital environment 4.2 Explain humane slaughter of food animals and mercy killing where necessary	Describe how animals can be safely moved from one place to another Define human slaughter and mercy killing and how it is done	Text books Journals Pictures Internet	Demonstrate how animals can be transported effectively and safely Demonstrate how humane slaughter and mercy killing is being done	Show to students how to transport animals safely. Show how food animals are slaughtered and how mercy killing is being done	Loading- Bay Lairage Slaughter - slabs Abattoirs farms text books, Journals Pictures internet	Assignment Test, Quizzes, Examination (theory and practicals)

PROGRAMME:	ND VETERINARY LABO	RATORY TECHNO	LOGY	
COURSE TITLE	E: INORGANIC CHEMIST	RY		
COURSE CODE	E: VLC 111			
DURATION	Lecture:- 1Hour	Tutorial:-	Practical:- 2 Hours	Total:- 3 Hrs/wk(45Hrs/sem)
CREDIT UNITS	: 3			
GOAL: This cou	urse is designed to enable the	students acquire know	ledge of the basics of inc	organic chemistry
GENERAL OBJ	ECTIVES: On completion o	f the course, the stude	nt should be able to:	
1.0 Understand at	oms, molecules, composition	and structure		
2.0 Understand m	ole concept and Stoichiometry	y in Chemical Reaction	ns	
3.0 Understand Pe	eriodicity and Chemical Bond	ing		
4.0 Understand th	e Chemistry of elements in gr	oups I, II, VII and Trai	nsition metals	
5.0 Understand th	e concept of Acids, Bases and	Salts		

COURSE TITLE: INORGA	NIC CHEMIS	TRY	Course	Code: VLC 111	Contact H	lours:3Hrs/wk	(45hrs/sem)
COURSE SPECIFICATION	: Theoretical c	ontent		Practical Content:			
General Objective: 1.0 Under	rstand Atoms,	Molecules ant the	eir Struct	tures			
Specific Learning	Learning Res		Teachers	Learning	Evaluation		
Objectives:	Activities			Learning Objectives:	Activities	Resources	
1.1 Outline the history of atomic and molecular theories. 1.2 Explain the atomic model based on works of Faraday, Thompson, Millikan, Rutherfords and Moseley. 1.3 Explain the experimental basis of atomic theory using the Bohr's theory of hydrogen atom and many electron atoms. 1.4 Describe the determination of relative atomic and molecular masses. 1.5 Explain isotopes and their uses	Lecture	Relevant textl Journals, Inter Classroom spa	rnet,	View the visible emission spectra of several metals in some of their compounds Interpret the mass spectrum of representative elements such as Oxygen, Carbon, Chlorine etc.	Guide and supervise lab technicians, technologists and students	direct vision spectroscope Bunsen burner, nichrome wire fixed to a cork handle, concHCl, solid chlorides of : barium, calcium, potassium, sodium and strontium beakers and watch glasses	Tests, classwork, assignment, etc

General Objective: 2.0 Understand Mole Concept and Stoichiometry in Chemical Reactions							
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation	
2.1 Be able to define the Mole 2.2 Describe molar mass 2.3 Interconversion of Moles, mass, and number of species 2.4 Calculation of mass percent from the chemical formula 2.5 Understand empirical formula 2.6 Understand molecular formula 2.7 Write and Balance Chemical Equations 2.8 Be able to write balanced chemical equations and to balance chemical equations. 2.9 Be able to express concentration in terms of molarity 2.10 Be able to perform interconversions of Mole-mass number for species in solution	Definition, illustrations Illustrations Calculates	Classroom space, relevant textbooks, journals, internet etc	Determine the formula of a compound from experimental data Be able to prepare a standardized solution of NaOH or HCl	Guide students	NaOH crystals, Volumetric Apparatus etc	Conduction of tests, assignments	

2.11 Be able to use stoichiometry of chemical reactions in solution						
Objective: 3.0 Understand P	eriodicity and (Chemical Bonding	-	•	·	•
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
3.1 State the different main energy levels of an atom, namely K, L. M 3.2 Explain Quantum Numbers 3.3 Correlate the energies of the electron in the K,L,M,N,shells with the values of the principal quantum no n= 1,2,3,4, 3.4 State Hund's rule, Pauli exclusion Principle, Auf-Bau principle and Heisenberg Uncertainty principle 3.5 Explain 3.4 above in relation to the concept of orbitals including subsidiary energy (s,p,d, f orbitals) 3.6 Explain the significance of the four quantum numbers	Lecture	Classroom resources Classroom resources	View the visible emission spectra of several metals in some of their compounds Interpret the mass spectrum of representative elements such as Oxygen, Carbon, Chlorine etc.	Guide and supervise lab technicians, technologists and students	Laboratory space and apparatus	Tests, Classwork, assignments, etc

3.7 Place elements in the Periodic table into groups and Periods					
3.8 Describe the shapes of s and p orbitals					
3.9 Explain Electronic Configuration within groups					
3.10 Write the electronic configuration for the first twenty elements					
3.11 Relate electronic configuration to the position of elements in the periodic table.				,,	,,
3.12 Describe trends in the Periodic table such as Atomic size, Ionization energy, Electron Affinity, Reactivity etc			,,		
3.13 Explain Valency and Chemical Bonding		Prepare iron			
3.14 Explain the Octet and Duplet rules		sulphide from iron and sulphur			
3.15 Explain and Distinguish between the following types of bonds: Ionic, covalent,		*			

metallic, coordinate covalent bonds						
3.16 Understand the						
formation of the types of bonds in 3.15 above						
bonds in 3.13 above						
3.17 Explain hybridization of						
atomic orbitals (sp ³ , sp ² ,						
spetc)						
General Objective: 4.0 Under	rstand the Chemi	stry of Elements in Grou	ips I, II, VII and	Transition Me	etals	
Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning Objectives:	Activities	Resources	
4.1 List the elements in group	*	C1			Mg, Ca, Sr,	T
I 4.2 Describe the occurrence	Lectures by illustrations.	Classroom resources, relevant textbooks.	Investigate the reactivity of	Guide and supervise lab	Ba, water, dilute	Tests, Classwork.
of Group I elements in nature	enumerating	Periodic table,	group 2 metals	technicians,	hydrochloric	assignments,
4.3 Describe the physical and	chamerang	internet, etc	(i) Mg. Ca, Sr,	technologists	acid test	etc
chemical Properties of group			and Ba with	and students	tubes etc,	
I elements			water		and	
4.4 Repeat 4.1, 4.2 and 4.3			(ii) Mg and Ca		laboratory	
above to elements in group II,			with dilute		space	
and VII respectively			HC1			
4.5 Compare the acid	T . 1					
strengths of the first four	Lecture and		Desertivity of			
halogens 4.6 Explain the meaning of a	explain transition		Reactivity of transition			
Transition metal	metals.		metals			
4.7 Write the electronic	metals.		metals			
configuration of transition						,,
metals of given atomic	Relate their		Obtain a UV	,,		,,
number	properties to		spectrum of a			

4.8 Explain the characteristic	electronic		range of				
properties of the transition	configuration,		coloured				
metals viz:	ionization		transition				
i. Metallic Character	energies, bond		metal				
ii. Variable Oxidation States	energies etc		complexes and				
iii. Complex ion formation			determine the				
iv. formation of coloured ions			wavelength of				
v. Paramagnetism			maximum				
4.9 Relate the characteristic			absorbance.				
properties of the transition							
elements above to							
(a) electronic configuration.							
(b) Atomic and ionic radio			Add Cobalt (ii)	,,		,,	
(c) Ionization energies			chloride to				
(d) Lattice energies and bond			water and				
energies			obtain UV				
(e) Availability of vacant			spectrum.				
orbital for complex			Relate colour				
formation			to absorption				
Compared Objectives 5 0 Understand the Compares of Apide Doses and Colte							

General Objective: 5.0 Understand the Concepts of Acids, Bases and Salts

Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning Objectives:	Activities	Resources	
5.1 Define an acid and a base according to Arrhenius, Bronsted – Lowry and Lewis concepts. 5.2 Identify acids and bases in chemistry equations. 5.3 Explain the meaning of the terms conjugates acid and conjugate base 5.4 Distinguish between a strong and weak acid or	Define acid, bases and salts and teach to identify them in equations	pH meter, conductivity meter, litmus strips etc	Carry out acid base titration by using conductance meter Determination of pH of	Guide and supervise lab technicians, technologists and students	Laboratory space and apparatus	Tests, classwork, assignments, practical reports.

base. 5.5 Write the expression for the dissociation constant for an acid HA (aq) 5.6 Give the equation for the degree of dissociation and concentration, M. (mole dm3) for a dilute solution of a weak acid. 5.7 Explain Ostwald's Dilution law and dissociation constant, K. 5.8 Calculate the degree of dissociation of a weak acid given the molarity and dissociation constant. 5.9 State the value of the ionic product of water. 5.10 Explain the concept of hydrogen on concentration and pH 5.11 Calculate the pH value of an acid or base given the hydrogen ion concentration 5.12 Identify various types of indicators and the use in the measurement of pH.	Explain dissociation constant and derive expression for it Work out simple calculations on degree of dissociation of weak acid	Identify indicators and use indicators in acid base titration	different solutions by using the different types of methods Measure the pH of solutions by using colour charts, indicators and pH meter Determine experimentally the strengths of acids and bases in relation to structure e.g. in the series CH3COOH, HCL, NH4, OH, NaOH	Guide and supervise lab technicians, technologists and students	"	,,,

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY

COURSE TITLE: MECHANICS AND PROPERTIES OF MATTER AND HEAT ENERGY

COURSE CODE: VLP 111

 DURATION
 Lecture:- 2 Hours
 Tutorial: Practical:- 3 hours
 Total:- 5Hrs/Wk(75hrs/Sem)

CREDIT UNITS: 5 CU

GOAL: This course is designed to develop the students understanding and applications of basic concepts in mechanics and properties of matter and heat energy.

GENERAL OBJECTIVES: On completion of the course, the student should be able to:

- 1. Understand the various fundamental and derived units.
- 2. Understand vectors and scalars and their application.
- 3. Understand the laws of motion.
- 4. Understand the basic concept of motion.
- 5. Understand the concept of work, energy and power.
- 6. Understand frictional forces and their effects.
- 7. Know the conditions for equilibrium of a stationary body under the action of coplanar forces.
- 8. Understand the principles of simple machines and their uses.
- 9. Understand circular motion.
- 10. Understand the behaviour of fluids at rest.
- 11. Understand the phenomenon of surface tension in liquids.
- 12. Understand the concept of elasticity

13. Understand the concept of viscosity
14. Understand the concept of temperature and its measurement
15. Understand the phenomenon of expansion and its effects
16. Understand the concept of heat as a form of energy
17. Understand the three states of matter and change of state
18. Understand the behaviour of gases in terms of atomic or molecular motions
19. Understand the modes of heat transfer.

	COURSE TITLE: MECHANICS AND PROPERTIES OF MATTER AND HEAT ENERGY				Contact I	Contact Hours:5Hrs/Wk(75hrs/Sem)			
COURSE SPECIFICATION	N: Theoretical con	itent			Pı	ractical Conte	nt:		
General Objective: 1.0 Understand the various fundamental and derived units									
Specific Learning	Teachers	Learning Re	sources	Specific	Teachers	Learning	Evaluation		
Objectives:	Activities			Learning Objectives:	Activities	Resources			
1.1 Identify the fundamental and derived quantities e.g. mass, length, time, temperature and charge. 1.2 List the S.I. units in which (1.1) quantities are measured. 1.3 State the dimensions of (1.1) above 1.4 Derived the dimensions some quantities in terms of fundamental quantities. 1.5 State the merits and demerits of dimensional analysis 1.6 Identify dimensionless quantities 1.7 Explain the use of measuring devices or apparatus; and their limitations e.g micrometer	Give examples and show the relation between them Check physical laws, equations using dimensional analysis. Measure small distances using micrometer screw guage and Vernier callipers.			To know how measure length, areas, volume of regular and irregular objects		Meter rule, Vernier calipers, micrometer screw gauge			

stand vectors and	scalars and their applicati	on.			
Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Activities		Learning Objectives:	Activities	Resources	
Illustrate with		Determine			
examples		experimentally			
-		the use of			
Solve		force board to			
numerical		calculate			
problem using		resultant and			
expressions in		equilibrium of			
2.5		two or more			
Express 2.5 in		vectors (force)			
terms of unit					
vectors		Demonstrate			
Solve		experimentally			
numerical		the			
and 2.7		law of forces			
Solve					
numerical					
problems to					
calculate					
resultant and					
equilibrium					
acting from a					
point using 2.8					
	Teachers Activities Illustrate with examples Solve numerical problem using expressions in 2.5 Express 2.5 in terms of unit vectors Solve numerical problems in 2,6 and 2.7 Solve numerical problems to calculate resultant and equilibrium forces/vectors acting from a	Teachers Activities Learning Resources Illustrate with examples Solve numerical problem using expressions in 2.5 Express 2.5 in terms of unit vectors Solve numerical problems in 2,6 and 2.7 Solve numerical problems to calculate resultant and equilibrium forces/vectors acting from a	Activities Illustrate with examples Solve numerical problem using expressions in 2.5 Express 2.5 in terms of unit vectors Solve numerical problems in 2,6 and 2.7 Solve numerical problems to calculate resultant and equilibrium of tocalculate resultant and equilibrium of two or more vectors (force) Demonstrate experimentally the parallelogram law of forces Solve numerical problems to calculate resultant and equilibrium forces/vectors acting from a	Teachers Activities Learning Resources Learning Objectives: Illustrate with examples Solve numerical problem using expressions in 2.5 Express 2.5 in terms of unit vectors Solve numerical problems in 2,6 and 2.7 Solve numerical problems to calculate resultant and equilibrium forces/vectors acting from a Determine experimentally the use of force board to calculate resultant and equilibrium of two or more vectors (force) Demonstrate experimentally the parallelogram law of forces	Teachers Activities Learning Resources Specific Learning Objectives: Illustrate with examples Solve

	1		•			,
	solve					
	numerical					
	problems in 2.9					
	and calculate					
	the magnitude					
	and magnitude					
	calculate the					
	relative					
	velocities of					
	two cars					
	moving at a					
	particular					
	speed in i)					
	opposite					
	direction ii)					
	the same					
	direction					
General Objective: 3.0 Under	stand the laws of r					
Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning	Activities	Resources	
			Objectives:	Activities	Resources	
3.1 Define motion	Illustrate with					
3.2 List various types of	examples					
motion.	_					
3.3 Define distance and	Sketch graphs					
displacement.	to illustrate the					
3.4 Define speed, velocity	terms					
and acceleration						
(instantaneous, uniform	Solve					
average).	numerical					
3.5 Distinguish between:	problems to					
i) Distance and displacement	calculate					
ii) Speed velocity	distance, time,					
ii) Speed velocity	velocity and					
	velocity and					

iii) instantaneous and	acceleration in					
average velocities	3.6 and 3.7					
iv) instantaneous and average						
acceleration v) instantaneous	Solve					
and uniform velocities	numerical					
v) instantaneous and uniform	problems to					
acceleration	calculate time					
3.6 Derive the three	of flight, time					
equations of uniformly	to flight,					
accelerated motion.	maximum					
3.7 Draw the displacement-	height and					
time and velocity- time	range					
graphs						
3.7 Draw the displacement-						
time and velocity- time						
graphs						
3.8 Define Projectile						
3.9 List examples of						
projectile and areas of their						
applications.						
3.10 Derive the equations of						
projectile in one and two						
dimensions.						
General Objective: 4.0 Under			T	T	T	
Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning Objectives:	Activities	Resources	
4.1 Explain the concept of	Illustrate with	Trolley loaded with				
mass and inertia.	examples	mass				
4.2 State Newton's first law.	Illustrate with					
4.3 Define momentum and	examples					
impulse .	_					
4.4 State S.I units of 4.3	Illustrate with					
4.5 State Newton's second	examples, the					
law						

4.6 Relate Newton second	relation mass					
law and changes in	and weight					
momentum.						
4.7 Deduce F = ma from the						
second law	Illustrate with a					
4.8 State the S.I unit of force	senerio where					
and define 1N	Newton three					
4.9 Distinguish between	laws are					
mass and weight.	applicable.					
4.10 State Newton's third						
law.	Solve					
4.11 State the law of	numerical					
conservation of linear	problems					
momentum.						
4.12 Derive equations 4.10	Illustrate with					
and 4.11.	examples					
4.13 Distinguish between						
elastic and inelastic collision.	Calculate the					
4.14 Define coefficient of	final					
restitution.	momentum:					
	velocities of					
	bodies in					
	Elastic and in					
	elastic					
	collisions.					
General Objective: 5.0 Under	rstand the concep	ot of work, energy and p	ower.			
Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities	_	Learning	Activities	Resources	
			Objectives:	Activities	Resources	
5.1 Define work, (F.S),	Solve					
energy and power (F.V)	numerical					
5.2 State the S.I units in (5.1)	problems using					
	area under					
	I			1	l	

5.3 Relate work done to	force-			
changes in kinetic and	displacement			
potential energies.	graph			
5.4 Identify other forms of				
energy.	Illustrate the			
5.5 State the law of	law of			
conservation of energy.	conservation of			
5.6 Explain conversion and conservation of all forms of	energy to			
energy.	freely falling			
5.7 Derive an expression for	bodies and			
the kinetic energy of moving	perfectly			
object and potential energy of				
object by reasoning of	collisions.			
position.				
5.8 Explain what is meant by				
conservation of a system of				
particles not acted by an				
external force				
5.9 List the power ratings of machines and appliances				
machines and apphances				
	1		l	

General Objective: 6.0 Unde	erstand frictional for	orces and their effects.				
Specific Learning Objectives: 6.1 Derive friction	Teachers Activities Illustrate with	Learning Resources	Specific Learning Objectives: Determine	Teachers Activities	Learning Resources	Evaluation
 6.2 State advantages and disadvantages of friction 6.3 State the laws of static friction. 6.4 State ways of reducing friction between two solid surfaces in relative motion 6.5 Distinguish between static and dynamic friction 6.6 List the applications in practical situation 	diagrams and examples Solve problems on velocities of solid bodies moving in horizontal and inclined planes where friction is not negligible		experimentally the coefficient of friction and verification of laws of friction		planed, rectangular object, masses, scale pan and thread	
General Objective: 7.0 Know		r equilibrium of a stationar	y body under the	action of copla	nar forces.	
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
7.1 Define moment of a force (torque) about a point and couple.	Illustrate with diagrams and solve	Metre rule, knife edge, masses, solid cone, bottle e.t.c	Determine experimentally C.G of		Force board, weight, plane	

7.2. Define centre of gravity (C.G)of a body.7.3 Explain and distinguish the terms; stable, unstable and neutral equilibrium7.4. State the conditions	numerical problem in 7.1 Solve problem on centre of gravity of composite	Beam balance,massese.t.c is used to apply equilibrium forces calculate unknown mass using pivoted	irregular shaped objects Determine		paper and thread	
necessary for stable equilibrium 7.5 State the principles of triangle of forces and Lami's theorem 7.6 State the necessary	Solve numerical problems to	meter rule	experimentally the law 7.8			
conditions for a system to be in equilibrium under the action of three non-parallel forces. 7.7 Define sensitivity of a	calculate equilibrium positions of various shapes;					
beam balance. 7.8 State the law of triangle forces	square, circle, rectangle e.t.c with given dimensions					
General Objective: 8.0 Under Specific Learning	stand the principle Teachers	s of simple machines and Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning Objectives:	Activities	Resources	
8.1Explain the concept of simple machines 8.2 List examples of simple machines 8.3 Define effort E, load L, mechanical advantages M.A. velocity ratio V.R and the efficiency of a machine.	Solve numerical problems to calculate; E,L,M.A,V.R and efficiency	Wheel barrow, bottle opener, scissors pliers e.t.c				

8.4Derive expression that connect effort E, load L, mechanical advantages M.A. velocity ratio V.R and the efficiency of a machine. 8.5 Explain why efficiency is not hundred percent.						
General Objective: 9.0 Unders	Teachers	on. Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning Objectives:	Activities	Resources	
9.1 Define the terms angular displacement; angular velocity, angular acceleration, tangential velocity and radial acceleration. 9.2 Derive expressions for angular velocity and angular acceleration. 9.3 Differentiate between: i) Linear displacement and angular displacement ii) Linear velocity and angular velocity iii) Linear acceleration and angular acceleration iv) Radial acceleration and tangential acceleration.	Illustrate with diagrams Calculate the distance travelled and velocities of bodies in horizontal and circular motion. Solve numerical problems involving banked tracks.		Demonstrate circular motion in both horizontal and vertical circles using a stone tied to a string.			

9.5 Explain how centrifuge			
works			
9.6 List examples of bodies			
performing circular motion.			
9.7 Explain the reason for			
banking of roads.			

Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
10.1 Define density and relative density of fluids 10.2 Derive an expression for the pressure exerted by a column of a fluid. 10.3 State Pascal's law of transmission of pressure in fluids 10.4 Explain the principles of hydraulic press. 10.5 State Archimedes" principles 10.6 State the law of floatation.	Calculate separately the following parameters for bodies partially or wholly immersed in a liquid by means of a string applying the law of floatation and for Archimedes principles: i) the tension of the string ii) yolume of	Measure pressures using manometers and barometers	Describe manometers and barometers (simple aneroid and for tin barometers) Verify experimentally the principle in 10.4 above Measure relative density of a liquid by			

gra bod	specific avity of the		mentioned in			
gra boo	avity of the					
			10.5 above.			
iv)	dy					
	relative		Measure			
der	nsity of the		relative			
liqu			density using			
			hydrometers.			
			Measure the			
			relative			
			density of a			
			liquid using a			
			U-table.			
			U-table.			
General Objective:11.0 Understar	nd the phenome	enon of surface tension in	liquids.			
	eachers	Learning Resources	Specific	Teachers	Learning	Evaluation
	ctivities		Learning	Activities	Resources	
o Sjeed vest	00111000		Objectives:	11001/10105	11050 011 005	
11.1 Explain the phenomenon Illu	lustrate with		J			
	agram					
11.2 Define and state unit of						
	olve simple					
	oblem on					
	rface tension					
interactions.						
11.4 State methods of						
reducing surface tension						
11.5 List effects of surface						
tension						
11,6 State applications of						
	lustrate with					
	agram					
11.8 Explain capillary	agrain					
11.0 Explain capillary						

11.9 List various applications of capillarity.			
11.10 Explain the effect of			
temperature on surface			
tension.			

General Objective: 12.0 Under Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities	Learning Resources	Learning Objectives:	Activities	Resources	Evaluation
12.1 Explain elasticity						
12.2 Define stress and strain	Spiral springs,					
12.3 State Hooke's law	masses					
12.5 Determine the elastic			Determine			
constant of a spring			experimentally			
12.6 Draw and explain the	Solve		the law in 12.3			
load extension curve and	numerical		and elastic			
indicate elastic limit.	problems		constant of a			
12.7 Define young's	involving		spring			
modulus.	Hooke, area					
12.8 Derive the expression	under force-					
for the energy stored in an	extension					
elastic spring or string,	graph e.t.c					
General Objective: 13.0 Unde	erstand the concep	ot of viscosity				
Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning	Activities	Resources	
10.1 F. 1			Objectives:			
13.1 Explain viscosity			1			

13.2 Compare viscosity with solid friction 13.3 Define the coefficient of viscosity 13.4 Explain the motion of a ball bearing falling through a viscous fluid. 13.5 Define terminal velocity 13.6 List applications of viscosity	Solve simple problem on viscosity		Determine experimentally coefficient of viscosity, terminal velocity			
General Objective:14.0 Unde	rstand the concept	of temperature and its m	easurement			
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
14.1 Define temperature using concept of thermal equilibrium 14.2 Define temperature in terms of thermometric properties, length of liquid column, pressure of a gas under constant pressure, resistance of a wire, e.m.f. of thermocouple, I radiation from a hot body. 14.3 Define temperature scales (Celsius scale, kelvin scale, ideal gas scale). 14.4 Convert Celsius to Kelvin scale. 14.5 Compare the Ideal gas scales and other scales	Thermometer Solve numerical problems on different scales of temperature and their conversion					

14.10 Construct and calibrate a liquid in glass thermometer and resistance thermometers 14.11 Convert temperature from Fahrenheit scale to Celsius and thermodynamic	thermometers 14.9 List the limitations of various types of			
and resistance thermometers 14.11 Convert temperature from Fahrenheit scale to Celsius and thermodynamic	various types of thermometers			
14.11 Convert temperature from Fahrenheit scale to Celsius and thermodynamic	14.10 Construct and calibrate a liquid in glass thermometer			
Celsius and thermodynamic	and resistance thermometers 14.11 Convert temperature			
	scales and vice visa using the			
appropriate formula.	ě l			

15.1 Explain the effect of	Calculate the		Determine			
heat on solids, liquids and	increase in		experimentally			
gases	volume, length		linear			
15.2 Define linear,	and area		expansivity of			
superficial and cubical	expansion of		solid in the			
expansivities	solids, liquids		form of a rod			
15.3 State the relationship	using correct		Determine			
between the expansivities	formulae are		experimentally			
15.4 Distinguish between	new length,		the cubical			
real and apparent expansion	area and		expansivity of a			
of liquids.	volume in		liquid using a			
-	expansivity,		specific gravity			
	superficial		bottle.			
	expansivity					
	cubical					
	expansivity					
	temperature					
	change					
General Objective:16.0 Und	lerstand the conce	pt of heat as a form of ene	rgy			
Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning	Activities	Resources	
			Objectives:			
16.1 Define heat energy			Determine			
16.2 Distinguish between			specific heat			
heat energy and temperature			capacity of			
16.3 Define heat capacity			solids and			
and specific heat capacity			liquids			
16.4 State Newton's laws of			experimentally.			
cooling.						
16.5 Explain cooling			Verify Newton's			
corrections in measurements			law of cooling			
of quantity of heat			experimentally.			
		1		1		
			Determine the			

16.6 Apply these			capacity of			
corrections (16.5 above) in			liquids using			
heat experiments			Newton's law			
16.7 Calculate the heat			Newton's law of			
capacity of different solids			cooling.			
and liquids from			-			
experimental results.						
General Objective: 17.0 Und	derstand the three	states of matter and chang	e of state			
Specific Learning	Teachers	Learning Resources	Specific	Teachers	Learning	Evaluation
Objectives:	Activities		Learning	Activities	Resources	
			Objectives:			
17.1 Define melting point	Calculate the		Determine			
and boiling point	heat capacity,		experimentally			
17.2 Explain the effect of:	specific heat		melting point			
i) pressure on boiling and	capacity and		and boiling,			
freezing points	specific latent		point of various			
ii) impurities on boiling	heat of fusion		substances e.g.			
point and freezing points	of solids given		naphthalene			
17.3 Explain latent heat of	necessary data					
fusion and vapourisation.	heat of fusion.		Determine			
			experimentally			
	Calculate the		specific heat			
	specific latent		capacity			
	heat of		substances			
	vapourisation					
	of liquids given					
	necessary data		Determine			
	Compare		experimentally			
	results above.		specific latent			
			heat of fusion			
			of ice.			
			Determine			
			experimentally			
			specific latent			

General Objective:18.0 Und	erstand the behavi	our of gases in terms of a	heat of vapourisation of and some other liquids tomic or molecular	r motions		
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
18.1 Define atom, molecule, Avogadro constant, relative molecular mass, mole, molar mass, molar volume and S.T.P. 18.2 Differentiate between: i) number of moles, number of molecules and Avogadro's constant ii) number of moles, mass of the gas and molar volume 18.3 State the assumptions of the kinetic theory of gases. 18.4 Explain Brownian motion 18.5 Explain Maxwellian distribution of velocities (quantitatively). 18.6 Explain the most probable speed, the mean speed and the mean square speed 18.7 Derive the expression for the pressure exerted by an ideal gas.	Solve relevant numerical problems in 18.2- 18.11 Calculate the volume of gases.		Verify the gas laws experimentally			

18.8 Relate the kinetic energy of a gas to its temperature 18.9 Derive the equation of state of an ideal gas using the kinetic theory. 18.10 Deduce other gas laws from the equation of state						
e.g. Boyles law and Charles law. 18.11 Distinguish between						
real and ideal gas 18.12 Deduce van der waal's equation of state of real gas						
General Objective:19.0 Unde	erstand the modes					
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
19.1 Explain the terms: Conduction, conversion and radiation of heat energy 19.2 Explain 19.1 above in terms of molecular theory 19.3 Define and give examples of good and bad conductors of heat energy 19.6 List some applications of heat transfer e.g. thermos flask, green house, land and sea breeze etc.			Demonstrate experimentally convection current in water Demonstrate experimentally that black bodies are better absorbers of radiation energy than polished or shiny surfaces.			

ND I SECOND SEMESTER

PROGRAMME: NI	PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY								
COURSE TITLE: INTRODUCTION TO ANIMAL DISEASES									
COURSE TITLE.	INTRODUCTION TO ANIMA	L DISEASES							
COURSE CODE: V	/LT 121								
DURATION	ON Lecture:- 1 Hour Tutorial:- Practical:- 1 Hour Total:- 2Hrs/wk (30Hrs/Wk)								
Dellition	2000000 121001	1 4101 441	Truckieury 1110ur	10000 2116/11 (00116/11)					
CREDIT UNITS: 2	2 CU								
COAL. This course	is designed to provide the stude	sta with basia knowlada	of animal disasses						
GOAL: This course is designed to provide the students with basic knowledge of animal diseases									
GENERAL OBJEC	GENERAL OBJECTIVES: On completion of the course, the student should be able to:								
	1	<u> </u>							

- 1.0 Know the terminologies used in animal diseases
- 2.0 Know the major viral diseases of domestic and laboratory animals
- 3.0 Know the major bacteria diseases of domestic and laboratory animals
- 4.0 Know the major fungal diseases of animals
- 5.0 Know the animal protozoan and rickettsia diseases of economic importance in Nigeria.

6.0 Know the major reproductive problems of animals	
7.0 Know animal helminths diseases of economic importance in Nigeria	
8.0 Know animal disease control procedures	
9.0 Know the common diseases of fish in Nigeria	

PROGRAMME: ND VETER	RINARY LABORATOR	Y TECHNOLOGY	7			
COURSE TITLE: INTROD	OUCTION TO ANIMAL	DISEASES	Course Code: VLT 121	Contac	t Hours:2Hrs/wk (30	Hrs/Wk)
COURSE SPECIFICATION	PECIFICATION: Theoretical content Practical Content:					
General Objective: 1.0: Kno	ow the terminologies use	ed in animal disea				
Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers	Learning	Evaluation
Objectives:		Resources	Objectives:	Activities	Resources	
1.1 Define common terms:	Define	Animal	1.1 Identify portals of	Supervise the	Animal samples	Assignments
(a) Disease	terminologies with	Samples (Sick	disease entry in the	identification of		
(b) Health	examples	and Healthy)	body	portals of disease entry in the body	Restraining rope	Tests
(c) Health Maintenance	Outline the factors	Pictures		chu y in the body		Quizzes
(d) Endemic	in 1.2	Pictures				Quizzes
(e) Sporadic						Examination
(f) Disease control	Give examples of					
(g) Disease eradication	specific diseases					
(h) Infection						
(i) Diagnosis	Outline the direct					
(j) Pathogensis	and indirect effects of ill-health in 1.4					
1.2 Explain certain factors	of ill-health in 1.4					
that may predispose						
animals disease e.g.						
nutrients status,						
transportation stress,						
sex, traumatic injury,						
etc.						
1.3 Classify the factors in						
1.2. above into intrinsic						
and extrinsic						
1.4 Explain the direct and indirect effect of						
mairect effect of						

animal ill-health on the						
economy of farmers						
and the nations						
1.5 List the clinical signs of						
diseased or sick						
animals e.g. anorexia,						
rough hair coat,						
emaciation, bilateral						
lacrimation, pale mucus						
membrane, etc.						
General Objective: 2.0:Know	v the major viral diseas	es of domestic and	laboratory animals		I I	
Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers	Learning	Evaluation
Objectives:		Resources	Objectives:	Activities	Resources	
2.1 List the major viral	Describe the general	Animal Samples	2.1 Identify common	Explain the	Animal Samples	Assignments
diseases of animals with	nature of viral	_	viral diseases	clinical signs,	-	
their aetiologic agents	diseases including	Syringes		control measures	Syringes	Tests
2.2 Describe the methods of	transmission,					
transmission, clinical signs,	clinical signs,	Needles			Needles	Quizzes
management and control of						
viral diseases in 2.1 above	treatment and	Drenching gun			Drenching gun	Examination
2.3 List some viral diseases	control					
with nervous involvement	F 1	Birds and vials			Birds and vials of	
e.g. Rabies and describe	Explain some viral	of vaccine			vaccine	
their clinical signs	diseases in 2.3 with					
2.4 Identify viral diseases	nervous signs e.g.					
characterized by skin lesions	canine distemper					
and describe their aetiology,	and rabies					
clinical signs and control						
2.5 Identify some specific	Explain some viral					
viral diseases of cattle e.g. rinderpest, humvy skin	diseases of sheep					
disease, foot and mouth	and goats					
disease, root and mouth						
clinical signs and control						
2.6 List some specific viral						
diseases of sheep and goats						
diseases of sheep and goats				I	I	

2.7 List some viral diseases of poultry						
General Objective: 3.0 Know Specific Learning	the major bacteria disc	eases of domestic a	nd laboratory animals Specific Learning	Teachers Activities	Learning	Evaluation
Objectives:		Resources	Objectives:		Resources	
3.1 List the major bacteria diseases of animal indicating their aetiology 3.2 Describe the transmission, clinical signs, diagnosis treatment and control of the following major bacterial diseases of animals e.g. Anthrax, Haemorrhagicsepticaecus, black lag, Tetanus, Tuberculosis, Brucellosis, Collibacillosis, etc. 3.3 Identify the clinical signs of diseases in 3.2. above in animals 3.4 List and describe major bacterial diseases of swine e.g. swine dysentery, erysipelas, enzootic pneumonia, etc. 3.5 Identify clinical signs of diseases in 3.4 above in swine	Explain the major bacterial diseases of animals and their causes Discuss the mode of transmission, diagnosis, treatment and control of these diseases Explain mastitis types and their causes, treatment and prevention Explain the causes, mode of transmission, clinical signs, treatment and control of diseases in 3.7.	Animal Samples Syringes Needles Sample bottles, Diagnostic kit Microscope Milking cup Restraint rope Test tube	3.1 Identify common bacterial diseases	Explain the clinical signs, treatment and control measures	Animal Samples Syringes Needles Sample bottles, Diagnostic kit Microscope Milking cup Restraint rope Test tube	Assignments Tests Quizzes Examination
3.6 List disease associated						

General Objective: 5.0 Know Specific Learning Objectives:	v the animal protozoan Teachers Activities	and rickettsia disea Learning Resources	ses of economic importance Specific Learning Objectives:	in Nigeria Teachers Activities	Learning Resources	Evaluation
(iii) Chronic respiratory diseases (CRD) of poultry 3.8 Identify clinical signs of CBPP, CCPP and CRD in animals General Objective: 4.0 Knox Specific Learning Objectives: 4.1 List some of the common fungal diseases of domestic animals and their aetiology e.g. Aspergillosis, ringworms, etc. 4.2 Identify clinical signs of fungal infection on animals 4.3 Name some common antimycotic drugs e.g. Amphoterrain B, Nystatin, Escabiol, etc.	Teachers Activities Explain the clinical signs of fungal infection on animals and some common antifungal drugs	Learning Resources Animal Samples Sample bottles, Diagnostic kit Microscope Petri dish Scapel blade	Specific Learning Objectives: 4.1 Identify common fungal diseases	Teachers Activities Explain the clinical signs, treatment and control measures	Learning Resources Animal Samples Sample bottles, Diagnostic kit Microscope Petri dish Scapel blade	Evaluation Assignments Tests Quizzes Examination
with Mycoplasma infections in animals 3.7 Describe the aetiology occurrence, clinical signs, diagnosis, treatment and control of: (i) Contagious bovine pleuropneumonia (CBPP) (ii) Contagious caprine pleuropneumonia (CCPP)						

5.1 List animal protozoan diseases of economic importance in Nigeria. Trypanosomaisis, Babesiosis, Corodriosis etc. 5.2 Describe method of transmission, clinical sings, treatment and control. 5.3 Identify clinical signs of diseases in 5.1 above in animals. 5.4 Identify the organisms listed in 5.1 above.	Determine the economic importance of protozoan diseases in Nigeria. Determine clinical signs of diseases in 5.1 above.	Blood samples Test tube Microscopes Slides Pictures	5.1 Identify common protozoan and rickettsia diseases	Explain the clinical signs, treatment and control measures Help students to the identify common protozoan and rickettsia organisms in microscope	Blood samples Test tube Microscopes Slides Pictures	Assignments Tests Quizzes Examination
General Objective: 6.0 Know				T	1	
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
6.1 Define infertility and sterility. 6.2 Describe types of infertility and their causes e.g Anestrus etc. 6.3 Describe reproductive diseases of animals in terms of causes and clinical signs e.g. I. Cystic ovarian complex II. Pseudo-pregnancy (false pregnancy) III. Phymosis and paraphymosis etc. 6.4 Explain enzootic abortions in farm animals,	Explain infertility types and causes Discuss enzootic abortion in farm animals and the economic importance	Vaginal speculum, Diagnostic kit, microscope, test tubes, Bulls and pregnant animals.	6.1 Identify with examples common reproductive diseases of animals	Show the students the pictures/ animals	Vaginal speculum, Diagnostic kit, microscope, test tubes, Bulls and pregnant animals.	Assignments Tests Quizzes Examination

their causes, clinical signs						
and economic importance.						
G 1011 11 E 0 TT						
General Objective: 7.0 Know Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers Activities	Learning	Evaluation
Objectives:	Teachers Activities	Resources	Objectives:	Teachers Activities	Resources	Evaluation
7.1 List animal helminth	Lecture with	Faecal samples,	7.1 Identify and Classify	Carry out practical	Model	Assignments
diseases of economic	pictures.	microscopes	various worms e.g.	demonstration of	poster of the	
importance in Nigeria] -		trematodes, cestodes and	various worms/eggs	worms/eggs	Tests
7.2 Describe the life cycle,			nematodes	e.g. trematodes,	Clarate.	0
transmission, clinical signs,				cestodes and nematodes	Charts	Quizzes
control, e.gcestodes,				nematodes	Projectors	Examination
nematodes, trematodes etc						
General Objective: 8.0 Know	w animal disease contro	l procedures		•		
Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers Activities	Learning	Evaluation
Objectives:		Resources	Objectives:		Resources	
8.1 List the general methods	Describe the	Automatic	8.1 Identify disease	Assist students in	Automatic	Assignments
of animal disease control e.g	common methods of	syringes, Dewormers,	control procedures	carrying out	syringes, Dewormers.	Tests
 Vaccination e.g 	disease control in a	Restraining rope,		species of animals	Restraining	Tests
Rinderpest.	named farm	animal, etc.			rope,	Quizzes
 Test and slaughter 					animal, etc.	
methods e.g CBPP						Examination
 Mass treatment and 						
mass vaccination e.g						
FMD						
 Vector control 						
e.gTrypanosomiasis						
 Quarantine e.g FMD 						
 Reservoir control 						
e.g Rabies		1				

 Isolation e.g Tuberculosis Strategic Deworming e.g. Haemonclosis 						
General Objective: 9.0 Know	v the common diseases	of fish in Nigeria				
Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers Activities	Learning	Evaluation
Objectives:		Resources	Objectives:		Resources	
9.1 List the viral diseases of	Explain the content	Fish	9.1 Identify viral,	Supervise the	Fish	Assignments
fish and their causes,	in 9.1 to 9.2	Pictures	bacterial, fungal,	identification of	Pictures	
clinical signs and control		Microscope	nutritional diseases of	diseased fish.	Microscope	Tests
9.2 List the major bacterials and fungal diseases of			fish			Quizzes
fish, their agents,						Examination
clinical signs, treatment						
and control						

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY

COURSE TITLE: VETERINARY MEDICINAL PRODUCTS AND DRUG USE(VMPS)

COURSE CODE: VLT 122

DURATIONLecture:- 1hrTutorial:- 0Practical:- 1hourTotal:- 2hrs/wk (30hrs/sem)

CREDIT UNITS: 2 CU

GOAL: This course is designed to enable students acquire knowledge and skills in veterinary medical products/drugs and their uses.

GENERAL OBJECTIVES: On completion of the course, the student should be able to:

- 1.0 Know basic medicinal products used in veterinary laboratory practice.
- 2.0 Know the available dosage forms of medicinal products and drugs.
- 3.0 Know how to administer medicinal products and drugs to patients.
- 4.0 Know how to properly keep different veterinary medicinal products and drugs.
- 5.0 Know how to weigh veterinary medicinal products/drugs and their units of measurements.

 	_	

	VETERINARY MEI DRUG USE(VMPS)	DICINAL Co	ourse Code: VLT122	Contac	t Hours: :- 2hrs/v	vk (30hrs/sem)
COURSE SPECIFI	CATION: Theoretica	al content]	Practical Content:	1
General Objective:	1.0: Know basic medi	cinal products us	ed in veterinary labora	tory practice.		
Specific Learning	Teachers	Learning	Specific	Teachers	Learning	Evaluation
Objectives:	Activities	Resources	Learning Objectives:	Activities	Resources	
1.1 Define veterinary medicinal products 1.2 Identify the types of veterinary medicinal products 1.3 Explain origin of veterinary medicinal	Define veterinary medicinal products Explain the various types of veterinary medicinal products Discuss the origin of various types of	Textbooks Journals Internet Pictorial chart	1.1 Show examples of veterinary medicinal products	Guide students to understand veterinary medicinal products	Practical guide manuals	Assignment Test Practical report Examination (theory and practical)

	medicinal products							
Canaral Objectives	2.0: Know the available	a docage forms of	Emedicinal products as	nd drugs				
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning	Teachers Activities	Learning Resources	Evaluation		
Objectives:	Activities	Resources	Objectives:		Resources			
2.1 Define drug	Define drug	Textbooks	2.1 Show	Demonstrate	Practical	Assignment		
dosage forms.	dosage forms.	Journals	various dosage	different dosages	guide	Test		
2.2 Outline the types		Internet Pictorial chart	forms. 2.2 Show	forms to students	Manuals	Test		
of dosage forms	•	1 ictoriai chart	equipment			Practical reports		
C			used in the			Examination (the arm)		
			delivery of			Examination(theory and practical)		
			various dosage forms.			and practical)		
General Objective:	General Objective: 3.0 Know how to properly keep different veterinary medicinal products and drug use							
Specific Learning	Teachers Activities	Learning	Specific	Teachers Activities		Evaluation		
Objectives:		Resources	Learning Objectives:		Resources			
	Explain how various	Textbooks	3.1 Demonstrate	Identifying all the	Practical	Examination		
. 1	dosage forms are	Journals	how they are	equipment used in	Guide	Ouiz		
administered	administered	Internet Pictorial chart	administered	delivery	Manuals	Quiz		
3.2 Identify		Tietoriai enart				Assignment		
requirements used in delivery of				Use the equipment hands on		Practical		
veterinary medicinal						Reports		
products						Test		
	4.0 Know how to prope			•		1		
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Resources	Evaluation		

3.1 Explain how		Textbooks	3.1 Identify	Show students the	Charts	Examination
to properly and		Journals	facilities for	facilities for storing		
safely handle	Explain how	Internet	storing drugs and	veterinary medicinal		Quiz
veterinary	veterinary medicinal	Pictorial chart	how they are stored	products and how		Assignment
medicinal	products are stored		stored	they are stored		1 10018
products						Practical
3.2 Explain how						Damanta
to store different						Reports
veterinary						Test
medicinal						
products						
3.3 List basic						
facilities used in						
storage of						
veterinary						
medicinal						
products						
General Objective	: 5.0 Know how to weigh	n veterinary medici	nal products/drugs a	nd their units of measure	ements.	
Specific	Teachers Activities	Learning	Specific	Teachers Activities	Learning	Evaluation
Learning		Resources	Learning		Resources	
Objectives:			Objectives:			
5.1 Describe	Explain with	Textbooks	5.1 Identify the	Show the students	Assignment	Examination
different units	examples various	Journals	basic weighing	basic weighing	Test	Ouiz
of metric	types of units	Internet	equipment	equipment.	Practical	Quiz
system		Pictorial chart	5.2 Demonstrate	Guide students on	report Examination	Assignment
			5.4 Demonstrate	Guide students on	Examination	

the use of the equipment

5.2 Identify the

basic weighing equipment

(theory and practical

Test

the use of basic

weighing equipment.

PROGRAMME:	ND VETERINARY LABOR	ATORY TECHNOLOGY

COURSE TITLE: ANIMAL ANATOMY AND PHYSIOLOGY

COURSE CODE: VLT 123

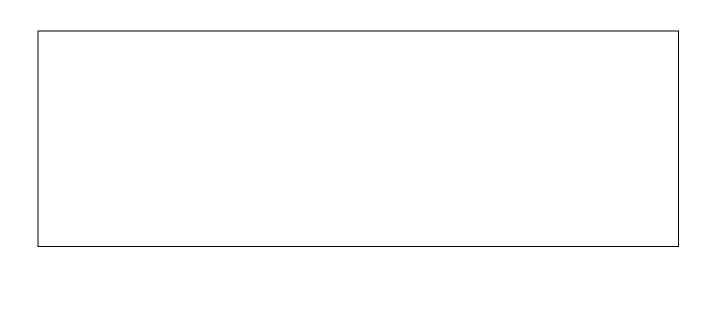
DURATION Lecture:- 2 Hours Tutorial:- Practical:- 2 Hours Total:- 4Hrs/Wk(60Hrs/Sem)

CREDIT UNITS: 4 CU

GOAL: This course is designed to acquaint the students to normal anatomic and physiology of animals

GENERAL OBJECTIVES: On completion of the course, the student should be able to:

- 1.0 Understand anatomic structures of the bovine by systems Head, eye, GIT, Skeletal, Genitalia, Muscular
- 2.0 Use Cadaveric dissection to identify various body organs by system
- 3.0 Know the vital parameters of all domestic animals
- 4.0 Use apparently healthy domestic animals to evaluate the normal range of vital parameters



PROGRAMME: ND	VETERINARY LABOR	ATORY TECH	INOLOGY			
COURSE TITLE: A PHYSIOLOGY	NIMAL ANATOMY AN	ID	Course Code: VLT 123	Contac	t Hours:4Hrs/Wk(601	Hrs/Sem)
COURSE SPECIFIC	CATION: Theoretical c	ontent		Pı	actical Content:	
General Objective: 1	.0: Understand anatomic	e structures of t	he bovine by systems – He	ad, eye, GIT, Ske	letal, Genitalia, Muscu	ılar
Specific Learning	Teachers Activities	Learning		Teachers	Learning	Evaluation
Objectives:		Resources	Learning Objectives:	Activities	Resources	
1.1 Identify various	Explain the various	Drawings of				Assignments
anatomic organ body systems	body structures	Bovine Organ	ns			Tests
1.2 Identify the normal organs	Show the normal organs in pictures	Textbooks				Quizzes
"in-situ" in pictures		Journals				Examination
		Power point				Spot Test
General Objective: 2	.0:Use Cadaveric dissec	tion to identify	various body organs by sy	stem	•	
Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers	Learning	Evaluation
Objectives:		Resources	Objectives:	Activities	Resources	
			2.1 Identify body cavities	Identify each	Bovine Cadaver	Assignments
			2.2 Identify organs	organ		Tests
			in each body cavity 2.3 Observe the			Quizzes
			relationship between the organs			Examination
Ceneral Objective: 3	.0 Know the vital param	eters of all don	on the cadaver			
Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers	Learning	Evaluation
Objectives:	2 decire o factivities	Resources	Objectives:	Activities	Resources	Z / MIMMOOII

3.1 List the common	Give Tables of the	Textbooks				Assignments
vital parameters in	Parameter	Journals				
domestic animals		Powerpoint				Tests
(Rectal Temperature,						Ouizzes
Respiratory rate and						Quizzes
Pulse rate						Examination
•		ĭ	ls to evaluate the normal			
Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers	Learning	Evaluation
Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Activities	Resources	Evaluation
	Teachers Activities	0	Objectives: 4.1 Use cow, goat,		0	Assignments
	Teachers Activities	0	Objectives:	Activities	Resources	
	Teachers Activities	0	Objectives: 4.1 Use cow, goat,	Activities Demonstrate	Resources	
	Teachers Activities	0	Objectives: 4.1 Use cow, goat, sheep, dog, cat,	Activities Demonstrate how to take	Resources	Assignments
	Teachers Activities	0	Objectives: 4.1 Use cow, goat, sheep, dog, cat, chicken to take	Activities Demonstrate how to take	Resources	Assignments
	Teachers Activities	0	Objectives: 4.1 Use cow, goat, sheep, dog, cat, chicken to take pulse, respiration	Activities Demonstrate how to take	Resources	Assignments Tests
	Teachers Activities	0	Objectives: 4.1 Use cow, goat, sheep, dog, cat, chicken to take pulse, respiration and temperature in	Activities Demonstrate how to take	Resources	Assignment

PROGRAMME: NI	O VETERINARY LABORATORY	TECHNOLOGY				
COURSE TITLE: V	eterinary Laboratory Procedure	1 (Instrument and Instrume	entation)			
GOLIDGE GODE 1	T T 101					
COURSE CODE: V	LT 124					
DURATION	Lecture:- 1Hour	Tutorial:- Hours	Practical:- 2 hours	Total:- 3 Hours		
DURATION	Lecture IIIour	Tutoriai Hours	Tractical 2 hours	Total 5 Hours		
CREDIT UNITS: 3	3					
GOAL: This course is designed to enable the students acquire knowledge of the various laboratory procedures in the Veterinary Laboratory						
GENERAL ORIEC	TIVES: On completion of the cou	rse the student should be able	to:			
GENERAL OBJEC	11 v Es. On completion of the cou	ise, the student should be able	ιο.			

- 1.0 Know the major equipment and Instruments in a Veterinary Laboratory
- 2.0 Know the uses of the equipment/Instrument in a Veterinary Laboratory
- 3.0 Know the care and maintenance of the equipment and Instruments in a Veterinary Laboratory

COURSE TITLE: Veterinary Lab and Instrumentation)	boratory Procedure 1	(Instrument	Course	Code: VLT 124	Conta	act Hours:hrs/	Semester (
COURSE SPECIFICATION: The	oretical content				Pra	ctical Content	:
General Objective: 1.0 Know the	major equipment and	Instruments in a	Veterinary	Laboratory			
Specific Learning Objectives:	Teachers			Specific	Teachers	Learning	Evaluation
	Activities			Learning Objectives:	Activities	Resources	
1.1 Outline the major equipment and Instruments in the Veterinary Laboratory	Enumerate, list and describe	Classroom sparelevant charts		Identification of laboratory equipment	Guides students in the identification		
1.2 Identify the equipment and instruments mentioned in 1.1 above							
General Objective: 2.0 Know the						1	1
Specific Learning Objectives:	Teachers	Learning Resou	ırces	Specific	Teachers	Learning	Evaluation
	Activities			Learning Objectives:	Activities	Resources	
2.1 Identify the different types of laboratory glass wares e.g. beakers test tube, funnels, flask etc. 2.2 State the uses of different laboratory wares	Lecture Practical identification and sketch/illustration in the laboratory Laboratory identification	Different glass the laboratory	wares in	See column 1	See column 2	See column 2	

2.3 Outline the uses of other laboratory equipment and instrument General Objective: 3.0 Know the Specific Learning Objectives:	Gets students involved in the preparation and use of cleansing agents. Teacher to demonstrate cleaning of sintered glass ware using chromic water and organic advents. Lecture care and maintenance acceptance and maintenance acceptance accepta	e of the equipment and Instr Learning Resources	uments in a Vetering Specific Learning Objectives:	nary Laboratory Teachers Activities	Learning Resources	Evaluation
3.1 Prepare cleaning reagents	Gets students	Beakers,	Prepare	See column 2	See	
for laboratory wares. 3.2 Clean laboratory wares	involved in the preparation	burette, pipette, test	different types of cleaning		column 3	
using cleansing agents	and use of	tube etc.	agents			
	cleansing agents. Teacher to	Water fittings,				
	demonstrate	gas fittings, light fittings				
	cleaning of	Grease, kipps				
	sintered glass	apparatus				
	ware using	condensers				
	chromic water and organic	Containers, H ₂ SO ₄ ,				
	advents.	alcohol etc.				
		Used or dirty				
		sintered glass				
		wares;				
3.3 Know how to maintain						
laboratory sensitive equipment	Demonstrate	Top loading	cleaning of			
like balances, Microscope etc	cleaning of	balance,	balances.			

balances. Allow students to participate under strict supervision	Analytical balance, Standard masses	Allow students to participate under strict supervision	Allow students to participate under strict supervision	,,	
--	--	--	--	----	--

_				
PROGRAMME: ND VI	ETERINARY LABORATO	ORY TECHNOLOGY		
COURSE TITLE: AND	MAL HANDLING AND W	ELFARE II		
COURSE CODE: VLT	125			
DURATION Le	ecture:- 1 Hours	Tutorial:-	Practical:- 2 Hours	Total:- 3Hrs/Wk(45Hrs/Sem)
CREDIT UNITS: 3 CU	J			
GOAL: This course is d	lesigned toexpose students to	the handling and huma	ne management of aquatic, exotic and	wild animals
GENERAL OBJECTIV	TES: On completion of the co	ourse, the student shoul	d be able to:	
2.0 Know how to provi 3.0 Know the health ca	•	and wild animals and	imals identify when they are stressed d risks involved during interactions	·

COURSE TITLE: A WELFARE II	NIMAL HANDLING AN	ND	Cour	rse Code: VLT 125	Contact 1	Hours:3Hrs/Wk(45H	(rs/Sem)
COURSE SPECIFIC	CATION: Theoretical c	ontent			Prac	ctical Content:	
General Objective: 1	.0: Know how to handle	and restrain	some aq	quatic, exotic and wi	ld animals		
Specific Learning	Teachers Activities	Learning		Specific	Teachers	Learning	Evaluation
Objectives:		Resources		Learning Objectives:	Activities	Resources	
1.1 Explain how aquatic, exotic and some wild animals are handled and restrained 1.2 Outline the equipment and facilities that are used in restraining aquatic, exotic and wild animals	Explain how aquatic, exotic and some wild animals are handled and restrained Explain the equipment and facilities that are used in restraining aquatic, exotic and wild animals	Textbooks Journals Pictures Internet		1.1 Handle and restrain some selected aquatic and wild animals 1.2 Identify equipment and facilities used in restraining aquatic, exotic and wild animals 1.3 Demonstrate how to use restraint equipment	Show how to handle and restrain aquatic, exotic and wild animals Show students the various equipment and facilities used in restraining aquatic, exotic and wild animals Guide students on how to use restraint	Practical Manuals and note books, Animals Restraint equipment and facilities	Assignments Tests Quizzes Examination
	.0: Know how to provid Teachers Activities				and identify when the		Evaluation
Specific Learning Objectives:	l eacners Activities	Learning Resources		Specific Learning Objectives:	Activities	Learning Resources	Evaluation
2.1 Explain how aquatic, exotic and wild animals are	Explain how aquatic, exotic and	Textbooks Journals Pictures	:	Demonstrate how to feed, water and house aquatic,	Guide students on how to feed, water and house	Zoo, Wildlife park	Assignments

managed (housing,	wild animals are		exotic and wild	aquatic, exotic		Quizzes
feeding, watering)	managed		animals	and wild animals		
2.2 Outline the signs						Examination
of fear, stress, pain	Explain to students					
and discomfort in	how to recognize the					
aquatic, exotic and wild animals	signs of fear, stress, pain and discomfort					
wild allillais	in aquatic, exotic					
	and wild animals					
General Objective: 3	.0 Know the health care	needs of aquatic, e	xotic and wild animals	and risks involved of	luring interactions	
Specific Learning	Teachers Activities	Learning	Specific Learning	Teachers	Learning	Evaluation
Objectives:		Resources	Objectives:	Activities	Resources	
3.1 Outline some	Explain some					Assignments
common diseases	common diseases of					
of aquatic, exotic	aquatic, exotic and					Tests
and wild animals	wild animals					Ouinne
3.2 List basic first aid						Quizzes
facilities used in	Explain some basic					Examination
aquatic, exotic	first aid facilities					2
and wild animals	used in aquatic,					
3.3 Outline some of	exotic and wild					
the risk involved	animals					
in handling	management					
aquatic, exotic						
and wild animals	Explain the risk					
3.4 List first aid	involved in handling					
requirement for	aquatic, exotic and					
handlers of	wild animals					
aquatic, exotic						
and wild animals	Explain the basic					
	first aid requirement					
	for handlers of					
	aquatic, exotic and					

wild animals

General Objective: 4.	O Know how exotic and O	l wild animals are p	ut to humane death			
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
4.1 List options for	Explain basic forms		0.0,0000.000			Assignments
puttingaquatic, exotic and wild animals to	of euthanasia					Tests
humane death						Quizzes
						Examination

PROGRAMME:ND VETERINARY LABORATORY TECHNOLOGY

COURSE:INTRODUCTORY BIOCHEMISTRY

CODE:VBC 121

DURATION Lecture: 1Hr Tutorial: Practical: - 2HRS Total: 3Hrs/Wk(45Hrs/Wk)

UNITS: 3 CU

GOAL: This course is designed to enable students to acquire a basic knowledge of Life science chemistry

GENERAL OBJECTIVES:On completion of this course, the student should be able to:

- 1.0 Understand the molecular organization of the living cell
- 2.0 Understand the importance of water and the concepts of pH and buffers
- 3.0 Understand the properties, sources, uses and structure of carbohydrates
- 4.0 Understand nature, biological, and industrial importance of lipids
- 5.0 Understand the structure, properties, behavior and functions of proteins
- 6.0 Understand the classification of amino acids
- 7.0 Understand the structure and behavior of proteins

PROGRAMME: ND VETERIN	NARY LABORATOR	Y TECHNOL	OGY			
COURSE TITLE: INTRODU BIOCHEMISTRY	CTORY	Co	urse Code: VBC121	Contact Hours: 3Hrs/Wk(45Hrs/Sem)		
COURSE SPECIFICATION:	Theoretical content	t		Practio	cal Content:	
General Objective: 1.0: Under	stand the molecula	r organizatio	n of the living cell			
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
1.1 List cell organelles 1.2 Explain centrifugation 1.3 Explain the structure, functions and fractions of intracellular organelles. 1.4 Describe chemical composition of the (i.e. carbohydrate, protein, lipids, DNA, RNA, nucleoproteins etc.)	Lectures On cell organelles	Internet Journals books	Centrifugation of fractions.	Demonstrate Practical on cell fractionation	Black Board Centrifuge Experimental animal Dissecting set Homogeniser Glasswares.	Assignments Tests Quizzes Examination
General Objective: 2.0: Under	stand the importan	ce of water a	nd the concepts of pH and	d buffers		I.
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
2.1 Explain the importance of water as a major Cellular component.	water and the	Internet Journals books	Choose the Appropriate acid and its salts (base	Demonstrate the use of the pH metre.	Indicator papers pH metre	Assignments Tests

2.2 List the properties of water. 2.3 List the common laboratory and physiological buffer systems with their components. 2.4 Explain how the buffers above function to resist pH changes particularly in physiological systems. Conduct practicalson the measurement of pH of solutions Conduct practicalson the measurement of pH of solutions. Glassware's/Tiles Examination Glassware's/Tiles Examination Glassware's / Tiles Examination
--

1	Evaleia cerbohydretes es	Lagtures on the	Internet	Test for	Conduct	ClasswarasPassants	Test
1		Lectures on the chemistry of	Journals		Practical test for	GlasswaresReagents such as molish,	Assignment
	Poly hydroxylaldehydes	carbohydrate	books	Carbohydrate in the		Fehling's etc.	Quizzes
	and their derivatives.	J		laboratory by e.g.	carbohydrates	8	Examinations
.2	List the general properties			Fehling's etc.			
	of carbohydrates.	Lectures on the					
.3	Explain the general	importance of					
	properties of	•					
	carbohydrates.	carbohydrate					
.4	List common sources of carbohydrates.						
.5	List domestic						
.5	and industrial						
	uses of						
	carbohydrate						
	S						
.6							
.0	carbohydrates as						
	mono-di-						
.7	Oligo and						
. '							
.8	polysaccharides'. Draw structural						
.0	formula of named						
	examples of the						
	families in 3.6 above.						

neral Objective 4.0: Unders	tand nature, biolog		al and industrial important	ce of lipids	Loi	In .
.1 Define lipids as fats and	Lecture on the	Internet	Test for fats in	Assist	Glasswares	Test
fat like substance.	importance of lipids	Books Journals	The laboratory	Students to carry	Bunsen burner	Assignment Ouizzes
.2 Define fat as mono-	npius	charts	e.g.by solubility test.	out laboratory	Water bath	Examinations
di-and tri-carboxylic		Citarts		assignment.	Saturated	Examination
esters of glycerides			Practical tests test for		And	
e.g.monoglycerides,			fats		Unsaturated fat	
diglyceridesand			Carry out simple Chemical tests for		Liquid and	
triglycerides.			triacylglycerides		Solid fats	
.3 List natural sources			unacyigiyeendes			
of fats.						
.4 Classify lipids into						
simple and complex						
lipids.						
.5 List members of						
classes in 4.4 above.						
.6 Describe with						
equation the						
hydrolysis of						
triglycerides.						
.7 Describe the						
hydrolysis of TAG						
with alkali to yield a						
mixture of soap and						
glycerol–						
(saponification)						
.8 Define						
saponification						
number, iodine						
number and free						
fatty acids(FFA)						
value of fats and						
oils(acylglycerols)						
0110(40)151)001010)		1	1	i e	i e	1

General Objective 5.0: Und	erstand the structure	, properties,	behavior and functions	of proteins		
5.1 Classify proteins as	Lectures on	Charts	Identify	Practical	Protein	Test
globular or fibrous.	classification of	Books	Proteins in the	Identification	Sample Million's	Assignment
5.2 List natural courses of	proteins	internet	laboratory	of protein	reagent Biuret	Quizzes
proteins	•			•	U	Examinations
5.3 State the characteristics	Lectures on the				reagent tiles.	
properties of the classes	functions of				dropers.	
in 5.1above.	proteins					
5.4 Explain with examples					Glassware	
the role of different					Colorimeteror	
proteins in the						
functioning of living					Spectrophotometer Waterbath	
matter e.g. transport,					waterbatti	
structural, catalytic,						
regulatory, defense, etc.						
5.5 Define prostheticgroup						
as a non-protein						
moiety of a complex						
protein.						
5.6 Describe proteins in						
terms of their						
prosthetic Groups e.g.						
haemoproteins,						
glycoproteins,						
lipoproteins etc.						
5.7 Describe the structure						
of a protein as a chain						
of amino acids which						
are chemically linked						
together by chemical						
bonds between						
carboxyl –alpha amino						
Groupson amino						
acids(Co-NH)						
5.8 Draw the general						
structural formula for						
alpha amino						
*		1		1		1

General Objective 6.0: Under	rstand the Classific	ation of Ami	no Acids and their struc	ctures		
6.1 Classify amino acids on the basis of the chemical	Lectures on amino		Identify amino Acid generally and specifically.	Amino acid standards and test samples, Ninhydrin.	Protein Sample Million's reagent Biuret reagent tiles. dropers.	Test Assignment Quizzes Examinations
General Objective 7.0: Under	rstand the structure	and behavio	rof Proteins	1	<u> </u>	l
7.1 Explain the primary, secondary, tertiary and quaternary structure of proteins. 7.2 List the types of interactions involved in: Secon dary Tertiar y and	structures of protein	Charts Text books journal				Test Assignment Quizzes Examinations

 Quater 			
nary			
structu			
res of			
protein			
s.			
7.3 Describe denaturation of			
proteins with examples			

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY										
COURSE TITLE: ORGANIC CHEMISTRY										
COURSE CODE: VLC 121										
DURATION	DURATION Lecture:- 1Hour Tutorial:- Hours Practical:- 2 hours Total:- 3 Hours									
CREDIT UNITS:	3									
GOAL: This cour	se is designed to enable the	students acquire knowle	edge of the basics of Organi	ic chemistry						
GENERAL OBJE	CTIVES: On completion	of the course, the studen	t should be able to:							
1.0 Understand th	ne Classification of Organ	nic compounds								
2.0 Understand th	ne Chemistry of Aliphatic	c Compounds								
3.0 Understand th	ne Chemistry of Aromatic	c Compounds								
4.0 Understand the Composition and Basic Properties of Carbohydrates										
5.0 Understand the general properties of Amino Acids and Proteins										
6.0 Understand the properties of Lipids										

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY										
COURSE TIT	LE: ORGANI	IC	Course Code:	Con	Contact Hours:3hrs/wk					
CHEMISTRY			VLC 121		(45hrs/sem)					
COURSE SPE	CIFICATION	:		Pra	ctical Conte	nt:				
Theoretical co	ntent									
General Objec	tive: 1.0 Unde	rstand the C	Classification of	Organic Com	pounds					
Specific	Teachers	Learning	Specific	Teachers	Learning	Evaluation				
Learning Objectives:	Activities	Resource s	Learning Objectives:	Learning Activities Resource						
					S					
1.1 List the										
major	Lectures		Determine	Guide and						

classification of	Classroo	qualitatively	supervise	Laborator	Tests,
organic	m	the	students	y space	Assignment
compounds.	resources	elements		and	s, term
1.2 Define		present		apparatus	papers, etc
homologous		in an organic			
series as		compound.			
consisting of					
compounds		Identify			
with each		functional			
successive		groups			
member		in organic			
differing with –		compounds			
CH2 –		via			
1.3 State the		qualitative			
members of a		chemical			
homologous		tests			
series and		(reactions)			
their physical					
properties.					
1.4 Define the					
functional					
group.					
1.5 Identify					
functional					
groups in					
alkanols,					
alkanals,					
alkanones,					
armines,					
alkanoic, acids,					
phenols,					
nitriles ethers,					
esters, amides					
etc.					
1.6 Draw					
structures for					
the functional					
groups in 1.5					
Above.					
	e: 2.0 Understand the C	hemistry of Ali	phatic Compor	unds	

General Objective: 2.0 Understand the Chemistry of Aliphatic Compounds

•			ě	• •		
Specific	Teachers	Learning	Specific	Teachers	Learning	Evaluation
Learning Objectives:	Activities	Resource s	Learning Objectives:	Activities	Resource	
					S	
2.1 Explain the						
general features	List and	Classroo	Determine	Demonstrate	Laborator	Tests,
of aliphatic	illustrate	m space,	qualitatively	, guides and	y space	Assignment
hydrocarbons.		relevant	the	supervise the	and	s,
2.2 State the		textbooks	elements	students	laborator	Reporting
general		, internet	present		у	of
formula,		etc	in an organic		apparatus	practicals.

	Г	T	T	T	T 2 2	
CnH2n+2 to			compound.		, lab	
represent					manual	
Alkane.			Identify			
2.3 Name			functional			
alkanes by			groups	,,		
using the			in organic			
IUPAC			compounds			
nomenclature			via			,,
2.4 List the			qualitative		,,	
industrial uses			chemical		,,	
of alkanes.			tests			
2.5 List natural			(reactions)			
sources of	Explanatio		,			
alkanes	n by					
2.6 State the	describing,					
general	illustrating					
formula,	and					
CnH2n to	enumeratin					
represent	g					
alkenes	5					
2.7 Explain the						
bonding in						
carbon atom as						
Sp2 hybridized						
in alkene						
2.8 Name						
alkenes by						
using the						
IUPAC						
nomenclature						
2.9 Represent						
the addition						
reactions of						
simple alkenes by means of						
chemical						
equation e.g. with Br2 HBr						
and H2.						
2.10						
Undrestand the						
use of curly						
arrows to						
represent						
reaction						
mechanisms						
2.11 Use curly arrows to show						
the mechanism						
of the above						
addition						
auumon				l		

	1			
reactions of				
alkenes				
2.12 Explain				
the use of				
alkenes in the				
production of				
polymers e.g.				
PVC,				
polyethene				
polystyrene etc				
2.13 Explain				
that the carbon				
in alkynes is Sp				
hybridized.				
2.14 Represent				
the addition				
reaction of				
alkynes by				
means of				
simple equation				
e.g. reaction				
with H2,Br2				
and HBr				
2.15 Describe				
chemical tests				
for the				
unsaturation in				
alkenes and				
alkynes.				
2.16 Describe				
the industrial				
uses of alkynes				
e.g. production				
of oxyacetylene				
flame,				
production of				
vinyl chloride				
in the				
production of				
polymers				
2.17 Describe				
the general				
methods of				
preparation of				
the following				
aliphatic				
compounds:				
i. Alkanes				
ii. Alkenes				
iii. Alkynes				
2.18 State the				
functional				
	1		1	

						1
group of						
alkanol as –						
OH						
2.19 State the						
general formula						
•						
of alkanols as						
C_nH_{2n+1} OH.						
2.20 Name						
alkanols using						
IUPAC						
nomenclature						
2.21 Describe						
the general						
methods for the						
preparation of						
the following						
aliphatic						
compounds:						
alkanols,						
alkanals,						
alkanones and						
alkanoic acids						
2.22 Describe						
the						
characteristic						
reactions						
associated with						
the aliphatic						
compounds in						
2.21 above						
2.23 Explain						
2.23 Explain the following						
2.23 Explain the following types of						
2.23 Explain the following types of reactions						
2.23 Explain the following types of reactions applied in						
2.23 Explain the following types of reactions						
2.23 Explain the following types of reactions applied in						
2.23 Explain the following types of reactions applied in organic						
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition						
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction						
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution						
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction						
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination						
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction						
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination	ve: 3.0 Under	rstand the C	Chemistry of Ar	omatic Compo	ounds	
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination reaction	1			_	1	Evaluation
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination reaction General Objecti Specific	Teachers	Learning	Specific	omatic Compo	Learning	Evaluation
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination reaction General Objecti Specific Learning	1	Learning Resource	Specific Learning	_	1	Evaluation
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination reaction General Objecti Specific	Teachers	Learning	Specific	Teachers	Learning Resource	Evaluation
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination reaction General Objecti Specific Learning	Teachers	Learning Resource	Specific Learning	Teachers	Learning	Evaluation
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination reaction General Objecti Specific Learning Objectives:	Teachers	Learning Resource	Specific Learning	Teachers	Learning Resource	Evaluation
2.23 Explain the following types of reactions applied in organic chemistry viz: i. Addition reaction ii Substitution reaction iii Elimination reaction General Objecti Specific Learning	Teachers Activities	Learning Resource	Specific Learning	Teachers	Learning Resource	Evaluation

Specific Learning	Teachers	Learning Resource	Specific	Teachers	Learning	Evaluation
_	I .	1	- I		T	
General Objecti	ve: 4.0 Under	stand the C	omposition and	Basic Propert	ties of Carbo	ohydrates
3.5 above						
mentioned in						
derivatives						
the						
the properties and reactions of						
ketones. 3.6 Describe						
aldelydes and						
phenols,						
of benzene;						
derivatives						
of the following						
the preparation						
3.5 Describe						
reactions.						
etc. Addition						
halogenation,						
sulphonation,						
(nitration,						
reactions						
Substitution						
acetylation),						
(alkylation or						
crafts						
Friedel						
benzene;						
reaction of						
the following						
3.4 Describe						
benzene).						
(especially						
hydrocarbons	illustrations					
aromatic	uses					
properties of	Teacher					
and chemical						
the physical						
3.3 Describe	s					
homologues	homologue					
benzene and its	and its					
structures of	of benzene					
3.2 Write the	structures	İ				

Specific Learning Objectives:	Teachers Activities	Learning Resource s	Specific Learning Objectives:	Teachers Activities	Learning Resource	Evaluatio
					S	
4.1 Define Carbohydrate	Teacher defines					

Amino Acids	on					
the general formulae of	and identificati					
Objectives: 5.1 Describe	Description		Objectives:		S	
Specific Learning	Teachers Activities	Learning Resources	Specific Learning	Teachers Activities	Learning Resource	Evaluation
groups in 4.3 above 4.5 Describe the general properties of the group in 4.3 above General Objecti	ve• 5 A Under	stand the G	eneral Propert	ies of Amino A	cids and Pr	oteins
and polysaccharides . 4.4 Describe the structures of selected members of the						
of carbohydrates. 4.3 Classify carbohydrates into monosaccharid es, disaccharides,	Illustrations		Differentiate experimentall y reducing sugars from non-reducing sugars	Teacher demonstrates , guides and supervises the students in the laboratory		

5.3 Enumerate the physical

and chemical properties of Amino Acids

5.4 Explain the formation of

dipeptides, oligopeptides

and poly peptides 5.5 Describe proteins as

1 0	1	1		I	ı	
made up of						
amino acids						
5.6 List the						
various classes						
of protein e.g						
fibrous and						
globular protein						
5.7 Describe						
the primary,						
secondary and						
tertiary						
structural levels						
of proteins						
5.8 Describe						
Protein						
Denaturation						
5.9 Explain						
Isoelectric						
point of amino						
acids and						
proteins						
General Objecti	ve· 6 0 Under	stand the Ra	sic Properties	of Linids	l	
Specific Specific	Teachers	Learning	Specific Specific	Teachers	Learning	Evaluation
Learning	Activities	Resources	Learning	Activities	Resource	2741441011
Objectives:	Tietrytties	Resources	Objectives:	rictivities	S	
6.1 Define			Objectives.		В	
Lipids		Textbooks		Teacher	Laborator	
6.2 List sources	Teacher	, internet		Demonstrate		
and types of	defines and	etc		s, guides and	y space and	
Lipids		Cic	Tests for	s, guides and supervises		
6.3	enumerates		fats in the	the students	apparatus	
Differentiate			laboratory			
between fats			1aboratory	during the experiment		
and oils				experiment		
6.4 Describe						
the general						
properties of						
lipids e.g						
saponification.						

	E: ND VETERINA LE: OPTICS, WAV			
COURSE COI	DE: VLP 121			
DURATION	Lecture:- 1hr	Tutorial:-	Practical:-	Total:-
		Hours	2hrs	3hrs/wk(45hrs/Sem)
CREDIT UNIT	TS: 3 CU		1	1
	ourse is designed to position and structure			nding of optics, atoms
GENERAL OF	BJECTIVES: On c	ompletion of the c	ourse, the student s	hould be able to:

- 1.0 Understand the principles of reflection and refraction at plane surfaces
- 2.0 Understand the principles of reflection and refraction at curved surfaces.
- 3.0 Understand the working principles of optical instruments and the human eyes
- 4.0 Understand the basic concepts of photometry
- 5.0 Understand the phenomenon of wave optics and sound waves
- 6.0 Understanding the atom, molecules composition and structure and Electromagnetic radiation

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY							
COURSE TITLE: OPTICS, WAVES AND ELECTROMAGNETIC RADIATION	Course Coo	de: VLP 121	Contact Hours: 3hrs/wk(45hrs/Sem)				
COURSE SPECIFICATION: Theoretical content Practical Content:							
Canaral Objectives 1 0 Understand the prince	inles of rofles	etion and rafrac	etion at plana surfaces				

General Objective: 1.0 Understand the principles of reflection and refraction at plane surfaces

	•	•			-	
Specific Learning	Teachers	Learn	Specific	Teachers	Learnin	Evaluatio
Objectives:	Activities	ing Resou	Learning Objectiv	Activitie	g	n
		rces	es:	S	Resourc	
					es	
1.1 Explain reflection of						
ray						
1.2 Explain ray and beam						
1.3 List types of beam						
1.4 State the laws of						
reflection of light						
1.5 State the nature or					Plain	
characteristics of the			Determin		mirror	
image formed by the plane			e		optical	
mirror (size, lateral			experime		pins,	
inversion, virtual,			ntally the		drawing	
position).			laws of		board	

medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and kaleidoscope 1.18 Describe the experiment to determine the angle of minimum deviation General Objective: 2.0 Understand the principles of ref	Determin the efractive index of colid and equid by eal and epparent epth inethod
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and kaleidoscope 1.18 Describe the experiment to determine the angle of minimum deviation	Determin the efractive index of colid and equid by eal and epparent epth inethod
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and kaleidoscope 1.18 Describe the experiment to determine the angle of minimum deviation	Determin the efractive index of colid and equid by eal and epparent epth inethod
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and kaleidoscope 1.18 Describe the experiment to determine the angle of minimum	Determin the efractive index of colid and equid by eal and pparent epth
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and kaleidoscope 1.18 Describe the experiment to determine the angle of minimum	Determin the efractive index of colid and equid by eal and pparent epth
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and kaleidoscope 1.18 Describe the experiment to determine	Determin the efractive index of colid and equid by eal and pparent epth
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and kaleidoscope 1.18 Describe the relation D D T T T T T T T T T T T T T T T T T	Determin the efractive index of colid and equid by eal and pparent epth
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and kaleidoscope	Determin the efractive endex of colid and equid by eal and epparent
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars, optical fibre and	Determin the efractive ndex of olid and quid by eal and
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine periscope, Binoculars,	Determin the efractive index of colid and equid by
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the following; submarine	Determin the efractive ndex of olid and
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the construction of the	Determin the efractive ndex of
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total internal reflection in the	Determin the efractive
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the application of total	Determin the
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism 1.17 Describe the	efraction Determin
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction through a prism	efraction
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection 1.16 Describe refraction	
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total internal reflection D example to the period of the phenomenon of total internal reflection D example to the period of the phenomenon of total internal reflection	
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the phenomenon of total	
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle. 1.15 Explain the	aws at
medium 1.13 Define critical angle 1.14 State the relation between refractive index and critical angle.	tally the
medium 1.13 Define critical angle 1.14 State the relation between refractive index D	xperime
medium 1.13 Define critical angle 1.14 State the relation	
medium 1.13 Define critical angle	Determin et en
medium	
vacuum and in the	
velocities of light in	
index in terms of	
1.12 Define refractive	
refraction	
1.11 State the laws of	
light	
1.10 Define refraction of	
	otated
	nirror is
	when the
	f ray
,	eflection
	ngle of
	tally the
	xperime
1.8 Explain the effect of e	
	Determin et en
1.7 Describe the deviation	
applications	
	urfaces.
	t plane
1.6 Explain the images re	eflection

Specific Learning	Teachers	Learning	Specific	Teache	Learnin	Evaluati
Objectives:	Activities	Resources	Learning	rs	g	on
			Objectives:	Activit	Resourc	
				ies	es	

0.15	I	1			
2.1Describe the					
shape of a curved					
mirror					
2.2 Define pole,					
principal locus,					
centre 01 curvature,					
principal axis, focal					
length, radius of					
curvature of curved					
mirrors.					
			D .		
2.3 State the			Determine		
relation between			experimentally		
radius of curvature			the focal length		
and focal length			of a convex		
2.4 List types of			mirror.		
curved mirror			Determine the		
2.5 Describe the			refractive index		
images formed by			of liquid using a		
curved mirrors			concave mirror		
2.6 Construct ray	Solve				
diagrams for the	numerical				
images in 2.3	problems				
above.	using the		Determine		
2.7 Derive the	_		experimentally		
mirror formula	mirror		the focal length		
2.8 State the real -	formula		of a converging		
is - positive and	(and ray		lens using the		
artesian	diagrams		lens formula		
conventions.			Determine the		
2.9 Define					
			focal length of a		
magnification			converging lens		
2.10 Describe the			by .c		
shape of lenses			magnification		
2.11 List types of			method (Ray		
lenses			box should be		
2.12 Define optical			used)		
centre, principal			Determine		
focus, principal			local length of		
axis, radius of			two thin lenses		
curvature of lenses			in contact using		
2.13 Derive lens			the formula.		
formula					
2.14 Explain			Determine		
defects of lenses			experimentally		
(spherical and			the focal length		
chromatic			of a diverging		
aberation) and their			lens		
corrections			Determine the		
501100115			refractive index		
			of a liquid using		
			a converging		
			u converging		

	lens and a plane mirror		

General Objective 3.0 Understand the working principles of optical instruments and the human eyes

Specific Learning	Teachers	Learning	Spec	Teachers	Learnin	Evaluatio
Objectives:	Activities	Resources	ific	Activitie	g	n
			Lear	S	Resourc	
			ning		es	
			Obje			
			ctive			
			s:			
3.1 Explain the	Calculate		Deri			
magnifying action of	the		ve			
lens	magnifying		expre			
3.2 Explain the	power of the		ssion			
working of:	optical		for			
i) Simple microscope	instruments		angul			
ii) Compound	of 3.2		ar			
microscope			magn			
iii) Astronomical	Calculate		ificat			
telescope iv) Galilean	the		ion			
telescope	magnifying		of a			
v) Terrestrial telescope	power,		lens			
3.3 Describe the	angular					
working of a	magnificatio					
spectrometer	n of optical					
3.4 Draw a labelled	instruments		Meas			
diagram of the human			ure			
eye	Calculate		angle			
3.5 Describe the	the focal		of			
functions of the parts of	lengths of		devia			
the eye	the		tion,			
	objective		mini			
	and eye		mum			

3.6 Explain the defect	lenses of		devia			
of the eye and their	compound		tion			
correction	microscope		angle			
Correction	_		of a			
given the						
	magnificatio		pris			
	n and other		m _.			
	necessary		using			
	parameters.		spect			
			rome			
	2 7 7 1 1 1 1		ter			
General Objective: 4.0					I	- · ·
Specific Learning	Teachers	Learning	Specific	Teachers	Learnin	Evaluati
Objectives:	Activities	Resources	Learning	Activitie	g	on
			Objectiv	S	Resourc	
			es:		es	
4.1 Define radiant	Calculate	Text books	Measure			
power, radiant flux,	the		experime			
luminous flux	luminous		ntally the			
4.2 Define	intensity I		Intensity			
illuminance,	and		of a light			
luminance and	luminous		source			
luminous intensity	flux F, of a					
4.3Describe the	source					
international standard						
source of light	Calculate					
4.4 Define solid angle	the					
4.5 Define luminous	luminance					
efficiency	of a surface.					
4.6 State the						
relationship between						
illuminance and						
luminous flux;						
luminous intensity						
and luminous flux						
4.7 State cosine law						
and inverse square						
law						
4.8 Describe lummer						
- Brolum photometer						
and the flicker						
photometer						
4.9 Compare						
intensities of light						
sources.						
General Objective: 5.0	Understand the	e nhenomenon <i>o</i>	of wave ontic	s and sound	l waves	1
Specific Learning	Teachers	Learning	Specific	Teachers	Learnin	Evaluati
Objectives:	Activities	Resources	Learning	Activitie	g	on
Objectives.	Activities	Resources	Objectiv		Resourc	OII
			es:	S		
5 1 Evalois the town		Toythoolya	cs.		es	
5.1 Explain the term		Textbooks				
waves						

7.0 Cl 'C				
5.2 Classify waves				
into mechanical and				
electromagnetic				
waves				
5.3 Distinguish				
between the two types				
of waves:-				
Longitudinal and				
transverse waves				
5.4 List examples of				
_				
waves 5.5 Define wave				
		D		
parameters e.g.		Determin		
frequency,		e		
wavelength, wave		experime		
velocity, amplitude		ntally the		
5.6 Explain sound		velocity		
waves in air columns		of sound		
and waves in strings		in air		
5.7 Define resonance		using a		
5.8 List examples of		resonance		
resonance in other		tube.		
physical events				
5.9 Identify the		Verify		
factors that affect the		experime		
velocity of sound		ntally		
waves in pipes		relationsh		
5.10 Establish the		ip length,		
relationship between		tension,		
		•		
the frequency of		massper		
waves in a straight		unit		
string and the length		length of		
and tension by		string		
dimensional analysis		using a		
5.11 Explain what is		sonomete		
meant by Doppler		r bridge		
effect	Calculate			
5.12 List examples of	the			
Doppler effect in	velocities of			
sound and light	types of			
5.13 Explain the	waves in			
terms:	different			
i) Reflection	media with	Demonstr		
ii) Refraction	different	ate		
iii) Super Position	parameters.	experime		
vi) Interference and		ntally the		
diffraction as they		relationsh		
relate to waves		ip		
5.14 State the		between		
conditions necessary		the terms,		
for interference and		in5.14		
diffraction to occur		using the		
difficultion to occur		abing the		

5.15 Explain the term		ripple		
beat		tank.		
5.16 Define beat				
frequency				
5.20 Determine beat				
frequency				
5.17 Explain the				
electromagnetic				
spectrum in relation				
to wave lengths and				
frequency				
5.18 Distinguish				
between emission and				
absorption				
5.19 Explain line,				
band and continuous				
spectra				
5.20 Explain the use				
of spectroscopy in				
analysing				
compositions of				
substances.				

General Objective: 6.0 Understanding the atom, molecules composition and structure and Electromagnetic radiation

Specific Learning Objectives	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teache rs Activit	Learnin g Resourc	Evalu ation
Objectives:				ies	es	
6.1 Explain the experimental basis of atomic theory using the Bohr's theory of hydrogen atom and many electron atoms. 6.2 Describe atomic spectra particularly the H atom emission spectrum 6.3 Discuss, qualitatively, the Energy States of the hydrogen	Lecture, diagrams, charts.					

		_		,
atom in the Bohr model				
6.4 Relate these				
Energy States to the				
observed · ·				
emission				
spectra				
6.5 Explain the				
limitations of				
the bohr				
model				
6.6 Describe the				
wave-particle				
duality of				
electrons and				
energy				
6.7 State the		G1 111 11 :		
different		Should be able to		
main energy		tabulate the		
levels of an		electromagnetic		
atom, namely		radiation under the		
K, L, M		heading: ionization,		
6.8 Correlate the		magnetic electric		
energies of		field deflection,		
the electron		charge, frequency,		
in the K, L,		wavelengths e.t.c		
M, N,shells with the				
values of the		Able to use and read		
		detection of		
principal	Lecture,	electromagnetic		
quantum no n=	· ·	radiation		
1,2,3,4,	diagrams, charts.	Tautation		
6.9 Relate the	Illustrate			
lines of the	with			
hydrogen	diagrams			
emission	,photographi			
spectrum to	c plates,			
electronic	thermometer			
energy level.	e.t.c			
6.10 State				
Hund's rule,				
Heisenberg				
uncertainty				
principle				
Pauli				
exclusion				
principle.				
6.11 Explain				
1.10 above in				
relation to the			 	

		1		
concept of				
orbitals				
including				
subsidiary				
energy levels				
(s, p, d)				
orbitals.				
6.12 Explain				
the				
significance e				
of the four				
quantum				
numbers 2.1				
Describe the				
shapes of s				
and p				
orbitals.				
6.13 Sketch				
the s and p				
orbitals				
6.14 Describe				
the				
determination				
of relative				
atomic and				
molecular				
masses.				
6.15 Explain				
isotopes and				
their use				
Describe the				
use of mass				
spectrometer				
as a means of				
proving the				
existence of				
isotopes.				
6.16 Define				
the				
following::				
(i) Atomic				
number,				
(ii) Mass number,				
(iii) Atomic				
mass,				
6.17 Define				
electromagnetic				
radiation				
6.18 List the				
family of				
electromagnetic				
radiation	<u>l</u>			

6.19 State the			
sources and			
origins			
electromagnetic			
radiation			
6.20 Enumerate			
various			
substances to use			
in the detection			
of			
electromagnetic			
radiation			
6.21 Explain the			
production of X-			
rays			
6.22 List types of			
X- rays			
6.23 Define basic			
radiation terms			
associated with			
X- rays: radiation			
dose, maximum			
permissible level			
e.t.c			

ND II

FIRST SEMESTER

PROGRAMME: NDVETERINARY LABORATORY TECHNOLOGY									
COURSE TITLE: Introduction to Veterinary Pathology									
COURSE CODE: VLT 211									
DURATION	Lecture:-	Tutorial:- Hours	Practical:-	Total:- 60 Hours					
			4hours/week						
CREDIT UNITS:	4								
GOAL: This course is designed to introduce students to the basics in Veterinary Pathology									
GENERAL OBJECTIVES: On completion of the course, the student should be able to:									

- 1.0 Know the basic concepts and specialties of Veterinary Pathology
- 2.0 Know the common TERMS IN USE IN Veterinary Pathology
- 3.0 Understand appearance and meaning of common pathology change
- 4.0 Know the basics involved in handling/processing of dead animal
- 5.0 Understand preparation and use of tissue preservation
- 6.0 Know the concept surrounding disposal of dead animals

COURSE TITLE:		Cours	e Code: VLT 211 Contact Hours:						
COURSE SPECIFICAT	ION: Theoretic	cal content		P	Practical Conte	nt:			
General Objective: 1.0									
Specific Learning	Teachers	Learning	Specific	Teachers	Learning	Evaluation			
Objectives:	Activities	Resources	Learning Objectives:	Activities	Resources				
1.1 Understand why we study Pathology	Teacher to explain why	Textbooks Journal	-	-	-	Tests Quiz Examination			
2.02.1 Give definition of common terms in pathology	Teacher to list term and define	Textbooks Journal	-	-	-	Tests Quiz Examination			
General Objective: 3.0									
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation			

3.1 Describe the different pathologic changes (circulatory, inflammatory, growth etc)	Teacher describes these changes	Textbooks Audio-Visual	Identify these pathologic changes	Demonstrate these Pathologic changes	Practical manual, Audio- visual, Post mortem room with	Tests Quiz Examination
4.0 4.1 Describe the processes involved in post mortem examination of animal	Teacher explains the processes	Textbooks Practical manual, Sketches, Drawings, Audio-visual	Identify the pathologic changes	Teacher perform post-mortem examination	Post mortem room with complete	Tests Quiz Examination
General Objective: 5.0			<u> </u>		<u> </u>	
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
5.1 Practice the preparation of preservation5.2 Determine the use of preservatives	Teacher explain preparation Teacher describes	Textbooks Practical Manuals Journal Practical manuals Textbooks	Identify preparation of preservatives Perform fixation of tissues	Demonstrate preparation of fixation Teacher show students how to	Glass container Salts Chemicals Weighing balance Distillate Glass/Plastic container Preservatives Practical	Tests Quiz Examination Tests
	30 3013003			preserve tissue	manual	Quiz Examination
General Objective: 6.0	l					
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
6.1 Explain why dead animals are to be disposed	Teacher outlines why	Textbooks Journals		-	-	Tests Quiz
of carefully. 6.2 Perform the safe disposal	Teacher describes the processes	Textbooks, Audio-visual, Journals	Practice disposal of dead animals	Teacher demonstrate the process	Incinerator Shovel Chemicals Digger	Examination Tests Quiz

of dead animal	l			Examination
animal	s			

PROGRAMME:	ND VETERINARY L	ABORATORY TECHNOI	LOGY						
COURSE TITLI	E: INTRODUCTION T	O VETERINARY MICRO	OBIOLOGY						
COURSE CODE: VLT 212									
DURATION	DURATION Lecture:- 2hrs Tutorial:- Hours Practical:- 2hrs								
CREDIT UNITS	: 4	1	1						
GOAL: This con	urse is designed to introd	duce the students to the basic	c rudiment of microbiolog	sy					
GENERAL OBJ	ECTIVES: On complet	ion of the course, the studen	nt should be able to:						
1.0 Know the mea	aning of veterinary micro	biology and the History of n	microbiology.						
2.0 Know the gen	eral characteristics of mi	croorganisms.							
3.0 Understand M	licrobial growth curve an	d factors influencing microb	oial growth.						
4.0 Understand ko	och's postulates.								
5.0 Know the isol	ation and identification to	echniques.							

COURSE SPECIFICATION: Theoretical content General Objective: 1.0: Know the meaning of veterinary microbiology and the History of microbiology. Specific Learning Objectives: Resource Activiti Specific Learning Objectives: Practical Content: Activities Fesource Objectives:	INTRODUCTION TO VETERINARY MICROBIOLOGY			se Code: VL	Т	Contact Hours: 4hrs/wk (60hrs/sem)					
Learning rs Resource Learning Activities Resource n	COURSE SPECIFICATION: Practical Content: Theoretical content General Objective: 1.0: Know the meaning of veterinary microbiology and the History of										
es s	Specific Learning	rs Activiti	Resource	e L	Learning Objectives: Activities R			Resource	Evaluatio n		

1.1 Explain	Explain	Textboo	-	-	-	
veterinary		ks				Report
microbiology						
100	Diamosi	Journals				Assignmen
1.2 Discuss brief	Discussi on		-	-	-	t
history of microbiology	OII	Internet				Test
			_	_	_	Quiz
1.3 Explain the	Explain					Examinati
scope and relevance of	scope					on
Microbiology	and					
2,7	importa nce					
1.4 Discuss	licc		Identification	Display and	Practical	
Microscopy			of	show parts,	Manual	
and types			microscopes	and types of	Textbook	
	D.		and various	microscopes.	S	
	Discuss		types			
	microsc					
	opy and					
	various					
	types					

General Objective: 2.0: Know the general characteristics of microorganisms.

Specific Learning Objectives:	Teache rs Activiti es	Learning Resources	Specific Learnin g Objectiv es:	Teachers Activities	Learning Resource s	Evaluati on
2.1 Explain the						
characteristics of :-		Textbooks	-	-	-	
	Explana	Internet				Assignm
a) Bacteria	tion	Microorganis				ent
b) Fungi		m				
c) Virus		Multimedia				Test
d) Algae		slides				
e) Protozoa						Quiz
						Examina
						tion
						uon

General Objective: 3.0 Understand Microbial growth curve and factors influencing microbial growth.

Specific Learning Objectives:	Teachers Activities	Learning Resource s	Specific Learning Objectives:	Teachers Activities	Lear ning Reso urce	Evaluatio n
3.1 explain microbial growth	Lectures	Textbooks Internet	Preparation of culture	Demonstratio n	Text book	Assignmen t

3.2 describe the microbial growth curve 3.3 explain the factors that influence microbial growth 3.4 explain a) physical methods b) controlling microbial growth c) chemical methods controlling microbial growth	Lis exp	t and blain the thods	Multimedi a Slide projector Journals	medium e.g NA,BA etc Inoculation of media with table top, skin or saliva swab	demonstration	Pract ials manu al journ als	Test Quiz Examinati on
Specific Learnin		Teache	Learning	Specific	Teachers	Learning	_
Objectives:		rs Activiti es	Resources	Learnin g Objectiv es:	Activities	Resources	e on
4.1 state and explorate the knoch's postulates with specific examples	S	Explain with example s	Textbooks Internet Journals				Assignm ent Test Quiz Examina tion
General Objecti						T	E 1 4.
Specific Learning Objectives:		achers tivities	Learning Resource s	Specific Learning Objectives:	Teachers Activities	Lear ning Reso urce s	Evaluatio n
5.1 Explain isolation of microorgani sm	_	olain lation	Textbooks Journals Internet Multimedi a	Culture on media Colonial	Demonstrate Demonstrate	Pract ical manu als	Practical Report Test
5.2 Explain isolation procedure		plain the cedures	projectors	appearance of bacteria e.g. size,	Demonstrate	book s	Examinati ons

of different		shape, color,		Journ	Assignmen
microorgani		consistency		als	t
sms		and texture.			
5.3 Discuss the	Discuss				
method of	identification		Display and		
identificatio	techniques	Identification	identify the		
n of	1	of stains	stain type		
microorgani					
sms	Explain the				
5.4 List the	staining				
staining	techniques.				
techniques	teeninques.				
for the					
different					
microorgani					
sms.					

PROGRAMME:	PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY							
COURSE TITLE	: INTRODUCTION TO	VETERINARY SUR	GERY					
COURSE CODE:	VLT 213							
DURATION	Lecture: 2 Hours	Tutorial:-	Practical:- 2 Hours	Total:-				
				4Hr/Wk(60Hrs/Sem)				
CREDIT UNITS:	4 CU							
GOAL: This course is designed to enable students acquire basic knowledge in surgery								
GENERAL OBJE	ECTIVES: On completion	on of the course, the stud	lent should be able to:					

- 1.0 Know the history of mycology
- 2.0 Know fungi identification methods
- 3.0 Know the fungal diseases of animal

PROGRAMME:	ND VETERINA	RY LABO) RATO	RY TECHNOL	OGY		
COURSE TITLE TO VETERINARY		ION	Cour	se Code: VLT	Contact I	Hours: 4Hrs/W	(k(60Hrs/Sem)
COURSE SPECIFICATION: Theoretical Practical Content:					ent:		
General Objectiv	ve: 1.0: Know th	e history of	of myco	ology			
Specific Learning Objectives:	Teachers Activities	Learnin Resource	ces	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
1.1 Define Veterinary Surgery 1.2 Explain the History of Veterinary Surgery 1.3 Explain the current status of Veterinary Surgery	Explain Veterinary Surgery	Textboo Internet Journals Projecto					Assignments Tests Quizzes Examination
General Objectiv	ve: 2.0: Know fu	ngi identi	fication	methods		1	
Specific Learning Objectives:	Teachers Activities	Learnin	ces	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
2.1 List common terminologies used in Veterinary Surgery 2.2 Explain each terminology	Explain when to use each terminology	Textboo Internet Journals Projecto	ks	Oxjecuvesi			Assignments Tests Quizzes Examination
General Objectiv	ve: 3.0 Know the	fungal di	seases	of animal			
Specific Learning Objectives:	Teachers Activities	Learnin Resource	ces	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
3.1 Explain how presurgical evaluation of patient is done	Explain the differences in presurgical evaluation and	Textboo Internet Journals Projecto	,				Assignments Tests Quizzes
3.2 Outline the	emergency						

	sequence of	surgical			Examination
	events	procedures			
	before and				
	during				
	surgery				
3.3	Explain how				
	post-surgical				
	evaluation is				
	done				

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY									
COURSE TITLE: VETERINARY NURSING									
COURSE CODI	E: VLT 214								
DURATION	Lecture:-1	Tutorial:- hrs	Practical:- 2hrs	Total:-					
	45hrs/Semester								
CREDIT UNITS	S: 3	·	•	•					
GOAL: This co	ourse is designed to a	cquire skills necessary f	or nursing care						
GENERAL OBJECTIVES: On completion of the course, the student should be able to:									
1.0 Know basic s	teps and instrument	for nursing care							

PROGRAMME:	ND VETERINA	ARY LAB	ORA	ATORY TECHN	OLOGY			
COURSE TITLE	: VETERINARY	7 0	Cour	rse Code: VLT	Contac	t Hours:hrs/S	Semester	
NURSING COURSE SPECI	EICATION, Th	agratical		214	Dwa	tical Conten	4.	
content	FICATION: III	eoreticai			Frac	ticai Conten	ι:	
General Objectiv	e 1.0: Know basi	c stens and	lins	trument for nurs	sing care			
		Learning		Specific Specific	Teachers			
Learning	Activities	Resource		Learning			Lvalaation	
Objectives:				Objectives:	Activities	Resources		
1.1 List	Explain	Text		Demonstrate	Show how		Assignment,	
techniques	wound	books		wound	wound		Quizzes,	
involved in	dressing, drug	Journal		dressing,	dressing,		Test,	
nursing	administration	Internet		Bandaging,	Bandaging,		Examination	
	according to			drug	Drug		theory and	
1.2 Explain	prescription			administration	administration		practicals)	
routes of				and	and			
medication	Describes			composition	composition			
	routes of							
1.3 Explain how	administering			Show route of	Show rate of			
to carry out	drugs and			medication	medication			
Bandaging	biological							
and splinting				Show splint	Show splint			
	Describe			application	application			
	types of							
	Bandag							
	application			Demonstrate	Show giving			
	and splints			giving	medication in			
				medication	different			
				e.g.	routes of			
	Explain how			Intravenous,	administration			
	and when			Intramuscular,	, .			
1.4 Explain	various			Intrape	Show how to			
procedure	medications			ъ.	recognize			
for giving	can be given			Demonstrate	problem,			
medications	Describ C.			how to	minimize			
1.5 Evaloia first	Describe first			recognize	injury and			
1.5 Explain first aid to	aid, recognize			problem minimize	take action			
and to animals and	injury and take action				Show how to			
how it is	take action			injury and take action	manage the wound and			
				take action	wound and			
administered								

1.6 Discuss	Explain how severe the	Demonstrate how to	other emergencies	
procedures necessary for successful	patient condition and describe treatment	manage the wound and other emergencies	emergencies	
treatment, for wound and other emergencies				

PROGRAMME: 1	ND VETERINARY LABORATORY	TECHNOLOGY							
COURSE TITLE:	Veterinary Laboratory Procedure	es (Glass fabrication, care	e and maintenance of equipmen	t)					
COURSE CODE:	VLT 215								
DURATION	Lecture:- 2 hours Tutorial:- Hours Practical:- 2 hours Total:-4								
CREDIT UNITS:	4	1							
GOAL: This cour	rse is designed to acquaint students w	ith knowledge of glass fab	prication and maintenance						
GENERAL OBJE	CCTIVES: On completion of the cou	rse, the student should be	able to:						
1.0. Know the the	principles of vacuum production								

- 2.0. Know common types of vacuum
- 3.0. Know different types of glasses used as laboratory waves
- 4.0. Know the construction of simple glass ware
- 5.0. Know the care and maintenance of glass blowing equipment

COURSE TITLE: Veterinary Laboration and maintenance		Glass	Course	Code: VLT 215	Co	ntact
fabrication, care and maintenance COURSE SPECIFICATION: The		urs			Practic	al Co
General Objective: 1.0: Know the p	orinciples of vacuum	production				
Specific Learning Objectives:	Teachers Activities	Learning Resou	rces	Specific Learning Objectives:	Teachers Activities	Lea
Vacuum Production 1.1. Classifying vacuum pressure gauges eg low medium, high, etc	Lecture Show gauges to students	Vacuum Pressure gauge		Operate a simple vacuum system	Get students involved in the operation of vacuum system	Vac pur
1.2. List the units in vacuum	Lecture					
1.3. List the various components of a simple vacuum set-up	Lecture					
General Objective: 2.0: Know communication Specific Learning Objectives:	Teachers Activities	Learning Resou	rces	Specific Learning Objectives:	Teachers Activities	Lea Res
Types of Vacuum 2.1. List common types of vacuum pumps: rotary and diffusion pumps	Lecture	Rotary pump Diffusion pump)	Operate pump in 2.2	Ensure each student has access to and operate a pump	Rot pur Dif pur
2.2. Describe the application of each of the pumps in 2.1 above	Emphasize areas of application of pumps					
General Objective: 3.0: Know comm	non types of veguum					
Specific Learning Objectives:	Teachers Activities	Learning Resou	rces	Specific Learning Objectives:	Teachers Activities	Lea Res
Types of glasses used as lab						
wares 3.1. List the types of glasses suitable for laboratory glass wares eg. Borosilicate, soda lime, silica glass, etc	Lecture and demonstration	Soda glass, Borosilicate Silica Glass		Identify types of glasses by chemical and physical	Lecture practical demonstratio n with soda and borosilicate	
3.2. State properties in glasses in 3.1 above eg. Transparency and durability etc	Teacher brings samples to class to show students			methods	with rods	
General Objective 4.0: Know the co					1	
Specific Learning Objectives:	Teachers Activities	Learning Resou	rces	Specific Learning Objectives:	Teachers Activities	Lea Res

Construction of simple glass wares 4.1. Identify various tools and equipment used in glass blowing workshops 4.2. Describe and apply glass cutting techniques 4.3. Describe and apply various methods of glass manipulation eg simple point pulling	Lecturer and teacher parades a number of these tools Teacher ask students to tabulate tools, draw and indicate uses	Glass cutting knife, calliper gauges, analing machine, lathe machine, glass cutting and calibrating machine, etc	Join two glass tube Blow bulbs at the end and in the middle of the tubes	Guide students in glass fabrication Teacher demonstration and construction of glass ware	
--	--	--	---	---	--

General Objective: 5.0: Know the c	are and maintenance	e of glass blowing equipment			
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Lea Res
Care and maintenance of glass blowing equipment 5.1. Identify various methods of taking care of equipment eg lubrication, servicing for valves, fittings etc	Lecturer demonstrate with some lubricants	Lubricants, lathe machine, laset glass cutter	Lubricate a machine, dismount, service and couple laset glass cutter	Guide students in glass fabrication Teacher demonstratio	Gui stud lub son mad
5.2. Describe the maintenance of glass blowing equipment	Check electrical connections and gas fittings	Lathe machine Anaeling furnace	Check for optimal performance of lathe machine	n and construction of glass ware	Sup stude che the fitti

COURSE TITLE	E: VETERINARY COM	MUNICATION					
COURSE CODE	COURSE CODE: VLT 216						
DURATION	Lecture:- 1hrs	Tutorial:- hrs	Practical:- 2hrs	Total:-			
15hrs/Semester							
CDEDIT UNITS	. 1	•	•				

CREDIT UNITS: 1

GOAL: This course is designed to enable students develop the ability to communicate on the field

GENERAL OBJECTIVES: On completion of the course, the student should be able to:

- 1.0 Know the concept of communication and communication skills
- 2.0 Know and apply the theories of communication
- 3.0 Understand the importance of communication in an a organizational set up
- 4.0 Understand the process of communication
- 5.0 Know communication flow in order to communicate effectively with others
- 6.0 Understand interpersonal and inter-group relationship for effective communication
- 7.0 Know the barriers to effective communication
- 8.0 Know how to write technical reports

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY							
COURSE TITLE: V	VETERINARY	Cou	ırse	e Code: VTL	Contact H	Iours:15hrs	/Semester
COMMUNICATION			216				
COURSE SPECIFIC	retical content			Pract	tical Conte	nt:	
General Objective 1	.0:Know the con	cept of commu	nica	ation and com	munication skill	S	
Specific Learning	Teachers	Learning	5	Specific	Teachers	Learnin	Evaluatio
Objectives:	Activities	Resources I		Learning Objectives:	Activities	g	n
				y		Resourc	
						es	
1.1 Define:	Defines	Text books,	I	Demonstrate	Examine	As	Quiz,
- Communica	Communicati	Journals, e-	a	ability to	critically all	shown	Test,
tion	on	library,/Intern	3	give	the variables	before	Assignme
- Communica		et,	a	acceptable	necessary for		nt,
tion skills		Posters/Chart	C	definition of			

1.2 Outline brief		s, Audio Visuals (AV)	communicati on	communicati on	Examinati on
history of communication			Exhibit listening speaking, reading, and writing skills	Role Play	
1.3 List					
communication skills i.e. listening, speaking, reading and writing	Gives a brief outline of history of communicati on Lists: Skills of communicati on Functions of communicati on	As in 1.1 above	Make oral presentations	Give culture- specific history of communicati on compare with modern technology	

General Objective 2	.0:Know and appl	y the theories	of communication	1		
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
2.1 Define theories of communication 2.2 List communication theories related to communication 2.3 Explain each theory listed in 2.2 above 2.4 Explain the purpose for use of theories in communication	Gives definition of communication theory Relates uses of theories to communication process in VLT		Apply communication theory Apply communication theory to VLT			
General Objective 3.0		1				
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning	Teachers Activities	Learning Resources	Evaluation
Objectives.	Activities	Resources	Objectives:	Activities	Resources	
3.1 Explain the						
importance of						

communication in an organization 3.2 Explain the importance of effective communication	
organization 3.2 Explain the importance of effective	
3.2 Explain the importance of effective	
importance of effective	
importance of effective	
effective	
effective	
communication	
in VLT	
2.2 Symbols	
3.3 Explain	
- register	
- factors	
influencing	
register	
3.4 Identify register	
peculiar to	
VLT	
General Objective 4.0:Understand the process of communication	
	Evaluation
Objectives: Activities Resources Learning Activities Resources	
Objectives:	
4.1 Analyze in	
detail the communication	
process	
process	
4.2 Describe	
different methods	
of	
communication	
- Oral	
- Written - Non-Verbal	
- Non-Verbai - Cultural	
- Modern	
technology	
4.3 Explain the	
roles of the	
speaker, message	
encoding,	
audience	
audience decoding	
audience decoding General Objective 5.0: Know communication flow in order to communicate effectively with other	
audience decoding General Objective 5.0: Know communication flow in order to communicate effectively with other Specific Learning Teachers Learning Specific Teachers Learning F	rs E valuation
audience decoding General Objective 5.0: Know communication flow in order to communicate effectively with other	

5.1 Describe the directions of				
of				
communication flow from				
internal to external				
5.2 Explain effective				
Flow				

General Objective 6.0	:Understand in	nterpersonal and	inter-group rela	tionship for e	effective comm	nunication
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
6.1 Explain:						
 Interpersonal 						
- Intra-personal						
- Inter-group						
Relationship						
6.2 Differentiate						
between the						
various						
terminologies						
listed above						
General Objective 7.0						
Specific Learning	Teachers	Learning	Specific	Teachers	Learning	Evaluation
Objectives:	Activities	Resources	Learning Objectives:	Activities	Resources	
7.1 Explain barriers to communication						
7.2 Identify the						
barriers to						
effective						
communication						
7.3 State the impact						
of barriers						
to communication						
e.g.						
- Break down						
- Tension						
- War						
7.4 Explain how						
barriers can be						
eliminated to						
foster						
communication						

General Objective 8.0:Know how to write technical reports						
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
8.1 Explain technical reports						
8.2 List types of Technical - Letters (business) - Memo - Invitation - Sanctions - Communication - Query - Promotion - Appointment - Dismissal, etc.						

PROGRAMME:	ND VETERINARY LA	BORATORY TECHN	OLOGY					
COURSE TITLE	: RURAL SOCIOLOGY	Y						
COURSE CODE:	COURSE CODE: VLT 217							
DURATION	Lecture:- 1 Hours	Tutorial:-	Practical:-	Total:-				
1Hr/Wk(15Hrs/Sem)								
CDEDIT UNITS.	1 CU							

GOAL: This course is designed toenable the students understand the behavior of rural people and rural society

GENERAL OBJECTIVES: On completion of the course, the student should be able to:

- 1.0 Understand common terms used in sociology
- 2.0 Understand concept of rural sociology
- 3.0 Know the rural settlement patterns and village organizations
- 4.0 Understand social change in the society
- 5.0 Know the rural social institution in Nigeria
- 6.0 Know communication patterns in a rural area

PROGRAMME: ND	VETERINARY	LABORATOR	Y TECHNOLO	GY			
			ourse Code: VLT 217	Contact H	Iours:1Hrs/W	(k(15Hrs/Sem)	
COURSE SPECIFIC	ATION: Theor	oretical content Practical Content:					
General Objective: 1.0: Understand common terms used in sociology							
Specific Learning	Teachers	Learning	Specific	Specific Teachers Learning Evaluat			
Objectives:	Activities	Resources	Learning Objectives:	Activities	Resources		
1.1 Define the	Explain the	Textbooks				Assignments	
following:	content in	Internet					
a. Sociology	1.1	Journals				Tests	
b. Rural Sociology		Projectors				Quizzes	

 c. Agricultural 						Examination
Laboratory						
d. Rural						
e. Culture						
f. Norms						
g. Values						
h. Beliefs						
i. Roles						
j. Status						
k. Power General Objective: 2	. 0: Understand o	oncent of rura	l sociology			
_	Teachers			Teachers	Lagunina	Evaluation
Specific Learning Objectives:	Activities	Learning Resources	Specific Learning Objectives:	Activities	Learning Resources	Evaluation
2.1 Explain the	Explain the	Textbooks				Assignments
following:	content in	Internet				
a. Change agent	2.1	Journals				Tests
b. Cosmopolites c. Ethnocentrism	Discuss the	Projectors				Oviesa
d. Xenocentrism	historical					Quizzes
e. Institution	development					Examination
f. Incest-Taboo	development					Zammaton
2.2 Explain the						
historical						
development of rural						
sociology in Nigeria	0.77			L		
General Objective: 3.			1		,	
Specific Learning	Teachers	Learning	Specific	Teachers	Learning	Evaluation
Objectives:	Activities	Resources	Learning Objectives:	Activities	Resources	
			j			Assignments
3.1 Explain:	Explain the	Textbooks				
3.1 Explain: a. Settlement pattern	Explain the contents in	Textbooks Internet				
_	-					Tests
a. Settlement pattern b. Village	contents in	Internet				Tests
a. Settlement pattern	contents in	Internet Journals				
a. Settlement pattern b. Village organization in	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the	contents in	Internet Journals				Tests
a. Settlement pattern b. Village organization in Nigeria	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas 3.3 State factors that influence settlement	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas 3.3 State factors that influence settlement patterns	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas 3.3 State factors that influence settlement	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas 3.3 State factors that influence settlement patterns 3.4 Explain the merit	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas 3.3 State factors that influence settlement patterns 3.4 Explain the merit and demerit of the various settlement	contents in	Internet Journals				Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas 3.3 State factors that influence settlement patterns 3.4 Explain the merit and demerit of the	contents in $3.1 - 3.4$	Internet Journals Projectors	the society			Tests Quizzes
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas 3.3 State factors that influence settlement patterns 3.4 Explain the merit and demerit of the various settlement patterns. General Objective: 4	contents in $3.1 - 3.4$	Internet Journals Projectors ocial change in		Teachers	Learning	Tests Quizzes Examination
a. Settlement pattern b. Village organization in Nigeria 3.2 List the settlement patterns in rural areas 3.3 State factors that influence settlement patterns 3.4 Explain the merit and demerit of the various settlement patterns.	contents in $3.1 - 3.4$	Internet Journals Projectors	the society Specific Learning	Teachers Activities	Learning Resources	Tests Quizzes

4.1 Define social	Explain the	Textbooks				Assignments
change in a rural	contents in	Internet				
area	4.1 - 4.3	Journals				Tests
4.2 Identify the		Projectors				Quizzes
characteristics of						Quizzes
rural change						Examination
4.3 Explain factors						
that influence						
rural change						
patterns						
General Objective: 5	.0 Know the rui	al social institu	ition in Nigeria			
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
5.1 Explain the	Explain the	Textbooks				Assignments
following terms:	contents in	Internet				_
a. Marriage and	5.1 - 5.3	Journals				Tests
the family		Projectors				Quizzes
b. Rural school						Quizzes
system						Examination
c. Religious						
institution						
d. Rural						
economic						
institution						
5.2 List types of						
rural social						
institutions in						
Nigeria						
- Marriage						
- Religious						
Institutions						
- School						
- Political						
Institutions						
- Traditional						
Institutions						
- Economic						
Institutions 5.3 Describe the						
factors that affect						
social change in the						
rural area:-						
- Type of						
marriage						
Educational						
- Educational Status						
- Family						
type/size						
type/size			1		1	

Economic power i.e.						
size of farm						
General Objective: 6	.0 Know commi	unication patte	rns in a rural area	a		
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
6.1 Define Communication 6.2 Explain patterns/types of communication in a rural set-up 6.3 State basic principles of Curriculum relevant to Veterinary medicine	Explain the contents in 6.1 – 6.3	Textbooks Internet Journals Projectors				Assignments Tests Quizzes Examination

KAMME: NA	ATTONAL DIP	LOMA IN '	VETERINARY	LABORATORY	TECHNOLOGY

RSE TITTLE: Research Methodology

RSE CODE: VLT 218

ATION Lecture: 2hrs Tutorial: Practical: - 0 Total: (30hrs/sem)

S: 2

This course is designed to enable the students carry out independent research projects.

GENERAL OBJECTIVES: On completion of this course, the student should be able to:

now what a research is and how to carry it out

nderstand basic principles of probability

nderstand the use and importance of some measures of central tendency and measures of dispersion in summarizing data

now the methodology applied in research

now the various methods of data collection

derstand the presentation of research report.

PROGRAMME: NATIONAL DIPLOMA IN VETERINARY LABORATORY TECHNOLOGY

COURSE:Research Methodology Course Code:VLT 218 Contact Hours:30hrs/sem

COURSE SPECIFICATION: Theoretical ContentPractical Content

General Objective:

1:Know what is a research and how to carry it out

Specific Learning	Teachers	Learnin	Specific	Teache	Learnin	Evaluati
Objectives	Activities	g	Learnin	rs	g	on
		Resourc	g	Activiti	Resourc	
		es	Objecti	es	es	
			ves			
1.1 Define the following:	i. Explain	Textboo				Assignme
• Research,	hypothesis and	ks,				nt
 Scientific method, 	its	laptop,				
• Theory,	characteristics.	overhea				Seminar
 Hypotheses. 	ii. Explain	d				Presentati
1.2 Describe various types	validation and	projector				on
of research e.g.	its	S,				
 Descriptive, 	problem in	journals,				
 Exploratory 	research.	writing				
• Casual,	iii. Distinguish	material				
Experimental and	among	S.				
Comparative etc.	specific,					
1.3 Explain the stages	general null					
involved in research:	and					
 Formulation of a 	alternate					
research problem	hypothesis. iv. Describe					
Literature Review	the					
 Hypothesis 	relationship					
 Justification of the 	between					
research	hypothesis and					
Data collection	problem					
and analysis	statement.					
 Findings etc. 	Statement.					
i manigo etc.	v. Explain the					
	relevance of					
	literature					
	review in					
	research.					
	vi. Explain the					
	sources of					
	literature.					
	vii. Describe					
	the					
	organisation					

and			
referer	ncing of		
literatu	ıre.		
viii. G	ive		
assign	ment		

General Objective:2: Understand basic principles of probability.

Specific Learning	Teachers	Learnin	Specific	Teache	Learnin	Evaluati
Objectives	Activities	g	Learnin	rs	g	on
		Resourc	g	Activiti	Resourc	
		es	Objecti	es	es	
			ves			
2.1 Define random	i. Explain	Textboo	-			Quiz
variable	variables, their	ks				Test
2.2 Explain the concept of	types and	-				Assignme
randomness	relevance.	Journals				nt
2.3 Define probability	ii. Explain					
2.4 State the laws of	consideration					
probability	in the choice					
2.5 Define conditional	of variables.					
probability	iii. Explain					
2.6 Solve simple problems	control					
involving the laws of	problems of					
probability as they relate	variables					
to their disciplines.						

General 3: Understand the use and importance of some measures of central tendency and

Objective: measures of dispersion in summarizing data

Specific Learning	Teachers	Learnin	Specific	Teache	Learnin	Evaluati
Objectives	Activities	g	Learnin	rs	g	on
		Resourc	g	Activiti	Resourc	
		es	Objecti	es	es	
			ves			
3.1 Explain the following:	Explain with	Textboo				Quiz
 arithmetic mean, 	examples the	ks				Test
 geometric mean, 	measures of	-				Assignme
 median mode and 	central	Journals				nt
 Harmonic mean. 	tendency and	Projecto				
3.2 Compute the measures	the measures	r				
in 3.1 above:	of dispersion					
(a) ungrouped data						
(b) grouped data						
3.3 Explain the uses of						
geometric and harmonic						
mean						
3.4 Calculate the						
following in a given set of						
data:						
 quartiles 						
 deciles 						
 percentiles 						

	I	1	ı	1	
3.5 List the merits and					
demerits of the above					
measures of central					
tendency					
3.6 Define measures of					
dispersion					
5.7 State the importance					
of the measures of					
dispersion					
3.8 Calculate the					
following					
 mean deviation 					
Semi Inter quartile					
range					
Variance and					
 standard deviation 					
3.9 Calculate the measures					
of the dispersion listed in					
3.8 given in a sets of data					

General Objective: 4: Know the methodology applied in research

Specific Learning	Teachers	Learnin	Specific	Teache	Learnin	Evaluati
Objectives	Activities	g Resourc	Learnin	rs Activiti	g Resourc	on
		es	g Objecti	es	es	
		CS	ves	CS	CS	
4.1 Explain sampling methods 4.2 Discuss the two major types of sampling: • probability sampling and • Non-probability 4.3 Explain the uses of sampling in research. 4.4 State the advantages and disadvantages of the two major types of sampling.	iExplain each of the method in probability sampling such as random, • system atic, • stratifie d and • clusters ii Explain Non-probability sampling such as: • acciden tal, • quotas and • Purposi ve.					Quiz Test Assignme nt

	iii. Explain population, sample and representativen ess. iv.Describe types ofsampling methods. v. Explain need for			
Camaral				

General Objective:

5: Know the various methods of data collection

Objective.	ı	1	T	1	1	T
Specific Learning	Teachers	Learnin	Specific	Teache	Learnin	Evaluati
Objectives	Activities	g	Learnin	rs	g	on
		Resourc	g	Activiti	Resourc	
		es	Objecti	es	es	
			ves			
5.1Define data 5.2 State the types of data:	Explain the following research instruments: 1. Questionnaire 2. Observation 3. Interview	Textbooks, laptop, overhead projectors, journals,				Quiz Test Assignme nt
 questionnaire, interview 5.4 Describe how to design questionnaires for research. 5.5 Explain how to administer questionnaires and interviews. 5.6 Design sample questionnaires for 	4. Ratings, etc. ii. Describe pitfalls of each instrument in above.	writing material s.				
research. 5.7 Administer the questionnaires in 5.6 above.						

General Objective:

6: Understand the presentation of research report

Specific Learning Objectives	Teachers Activities	Learnin g Resourc es	Specific Learnin g Objecti ves	Teache rs Activiti es	Learnin g Resourc es	Evaluati on
6.1 Explain the general format of a research report, e.g.	Explain research report and	Textboo ks, laptop,				Assignme nt

Preliminaries:	its contents.	overhea		Seminar
preface	ii. Conduct test	d		Presentati
foreword	Guidance for	projector		on
 Approval page 	students	s,		
• Title page	using	journals,		
 Dedication 	examples of	writing		
 Acknowledgement 	good reports.	material		
 Table of contents 	Reference to	S.		
 List of illustrations 	exercisesto			
e.g. maps/pictures,	review relevant			
diagrams, figures	literature etc.			
and tables	Identify the			
Main part	contents of			
 Introduction 	research			
 Aims and 	report.			
Objectives	1. Introduction			
 Limitations and 	2. Literature			
Delineations	Review			
Main text	3. Research			
 Conclusions and 	Method			
Recommendations	4. Data			
• The Appendix	Presentation			
• Glossary	and Analysis			
 Bibliography/Refe 	5. Summary,			
rences	Conclusion			
6.2 Explain the use of footnotes references and	and Recommendati			
bibliography				
6.3 Outline the use of	on 6. References			
abbreviations in citation	o. References			
		l .		

ND II

SECOND SEMESTER

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY									
COURSE TITLE: RECORD KEEPING AND DATA MANAGEMENT									
COURSE CODE	COURSE CODE: VLT 221								
DURATION	Lecture:- 1 hr	Tutorial:- hrs	Practical:- Ohrs	Total:-					
				15hrs/Semester					
CREDIT UNITS	S:								
GOAL: This co	urse is designed to give	background to record	s and its importance						
GENERAL OBJECTIVES: On completion of the course, the student should be able to:									
1.0 Know the principles of data collection and record keeping and data management									

2.0 Know how to use appropriate data storage devices available

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY								
COURSE TITLE:	RECORD KEEF	PING	C	ourse Code:	Contact Hours: 15hrs/Semester			
AND DATA MANA	GEMENT			VLT 221				
COURSE SPECIFI	CATION: The	oretical			Pr	actical Cont	ent:	
content								
General Objective 1								
Specific Learning	Teachers	Learning	-	Specific	Teachers	Learning	Evaluation	
Objectives:	Activities	Resource	es	Learning Objectives:	Activities	Resources		
1.1 Explain data	Define data	Text bool	S				Assignment	
collection	and data	Journals					Tests,	
	collection	Internet					Quizzes	
1.2 Describe	P 1 1 1						Examination	
importance of	Explain the							
record keeping	importance of record							
1.3 Explain data	keeping							
management	Recping							
	Explain data							
	management							
	_							
General Objective 2				_	vices availab			
Specific Learning	Teachers	Learning	_	Specific	Teachers	Learning	Evaluation	
Objectives:	Activities	Resource	es	Learning	Activities	Resources		
0.1.7.1	P 1 .			Objectives:				
2.1 List appropriate	Explain							
data storage devices available	appropriate							
uevices available	data storage							

2.2 Explain the following techniques: - Systemic collection - Recording; storage Retrieval, management and reporting of relevant information in veterinary domain 2.3 Explain the following terms: P records Billing, estimates	devices available Describe use of appropriate devices for data collection and processing				
--	---	--	--	--	--

PROGRAMME: NATIONAL DIPLOMA IN VETERINARY LABORATORY TECHNOLOGY							
COURSE TITLE: Seminar Presentation							
COURSE CODE: VLT 222							
DURATION:	Lecture: -	Tutorial: -	Practical: - 1hr	Total: 1hrs			
				(15hrs/semester)			
CREDIT UNITS: 1 CU							
GOAL: This course is designed to expose the students to practical skills for report writing and							
Presentation							
GENERAL OBJECTIVE: On completion of the course, the student should be able to:							
1.0 Understand concepts used in seminars and seminar writing.							
2.0 Know the stages involve	ed in seminar pre	eparation and w	riting.				
3.0 Demonstrate presentation of seminar paper.							

PROGRAMME: NATIONAL DIPLOMA IN VETERINARY LABORATORY TECHNOLOGY

COURSE: Seminar Presentation Course Code: VLT 222 Contact Hours: 1hrs

(15hrs/semester)

COURSE SPECIFICATION: Theoretical Content: Practical Content:

General Objective 1.0: Understand concepts used in seminars/seminar writing

Specific Learning	Teachers Activities	Learning Resource	Specific Learning	Teachers Activities	Learning Resources	Evaluation
Objectives	Activities	s s	Objective s	Activities	Resources	
1.1 Define: - Seminar - Symposi m - Worksho - Conference - Summit - Colloqui m. 1.2 Describe the processes of conducting each type listed in 1.1 above. 1.3 Identify ways of choosing themes/topics for seminars	p c	Lectures notes Pictures Charts Books Journals Seminar papers				Assignment , Test, Examinatio n

General Objective 2.0:Know the stages involved in seminar preparation and writing

Specific Learning Objectives	Specific Learning Objectives	Specific Learning Objective s	Specific Learning Objective s	Specific Learning Objectives	Specific Learning Objective s	Specific Learning Objectives
2.1 Explain the process of Choosing a seminar topic 2.2 Describe how to gather materials for seminar paper including data management.	Lecture, brainstormin g	Lectures Pictures Charts Books Journals	2.1 Identify the target audience Choose a seminar topics	Guide the students to identify the target audience, choose a topic and prepare a	Journals, books, Sample seminar papers	Assignment , report writing

2.3 Describe		Prepare a	seminar	
styles in seminar		seminar	paper	
paper preparation		paper		
and presentation.		r ·r		
2.4 State				
how to identify				
target for seminar				
presentation.				
2.5 Describe				
the use of peer				
review in seminar				
paper				

General Objective 3.0:Demonstrate presentation of seminar paper

Specific Learning	Specific Learning	Specific Learning	Specific Learning	Specific Learning	Specific Learning	Specific Learning
3.1 Describe how to prepare for	Objectives Lecture,	Objective s Lectures	Objective s 2.2 Write	Objectives Guide and	Objective s Seminar	Objectives Paper
seminar presentation (check list of material needed, method of presentation etc). 3.2 Describe how to capture your audience. 3.3 Describe styles of seminar presentation 3.4 Explain ways of convincing your audience 3.5 Describe ways of summarizing your points	brainstormin g, facilitation, demonstratio n	Pictures Charts Books Journals	a Seminar paper. Present a seminar paper. Conduct evaluatio n of seminar paper presented	coordinate students' presentatio n. Evaluate and score papers presented	papers, moderator s, PAS,	presentatio

PRO	PROGRAMME: NATIONAL DIPLOMA IN VETERINARY LABORATORY TECHNOLOGY									
COU	COURSE TITLE: Project									
COU	COURSE CODE:VLT 223									
DUR	RATION:	Lecture: 0	Tutorial: 0	Practical: 3	Total:45hrs/sem					
CRE	DIT UNITS:	3								
GOA	L:This projec	et provides the student wi	th an opportunit	y to investigate and re	port on a specific					
aspec	ct of microfina	nce and enterprise develo	opment							
GEN	IERAL OBJE	CTIVE: On completion	of the course, the	ne student should be ab	ole to:					
1.0	Review curre	ent literature pertaining to	a specific issue	or problem						
2.0										
3.0	O Gather and analyse information to identify potential solutions and reach a conclusion									
4.0	Produce a final report on the project									

PROGRAMME: NATIONAL DIPLOMA IN VETERINARY LABORATORY TECHNOLOGY						
COURSE: Proje	ect	COURSE	CODE:	CONTACT	HOURS: 45h	rs/sem
		VLT 223	T			
COURSE SPEC					Pra	ctical
Theoretical Con			Contents			
General Objecti						
Specific	Teachers	Learning	Specific	Teachers	Learning	Evaluation
Learning	Activities	Resources	Learning	Activities	Resources	
Objectives			Objective			
1.1. Choose,	Guide	List of	Students	Support	List of	Final
under	students in	available	should	students in	available	Project
guidance, an	selecting a	topics	select a	planning	topics	
appropriate	suitable	Notes	suitable	their	Notes	
topic for	topic	Internet	topic and	activities	Internet	
investigation	Provide	Textbooks	produce a	Support	Textbooks	
1.2. Review	guidance in	Internet	plan of	students in	Internet	
current	finding		action	their		
literature to	sources		setting out	research.		
relevant			key targets	Provide		
tochoose of			to meet	guidance in		
topic			over the	finding		
			semester Review	appropriate		
			current	sources		
			literature to			
			relevant to			
			choose of			
			topic			
General Objecti	ive 2 0. Assess t	he extent of t		blem in agricu	l Itural	
practice	14C 2.U.M33C33 [iic catelli 01 t	ine issue of pro	olem m agneu	itutai	
Specific	Teachers	Learning	Specific	Teachers	Learning	Evaluation
Learning	Activities	Resources	Learning	Activities	Resources	
Objectives			Objectives:			
2.1. Investigate	Support	Textbooks	Students	Guide	Textbooks	Final
the issue or	students in	Internet	should	students in	Internet	Project
problem in	their	Notes	identify	determining	Notes	
practice	investigation		required	the most		
			information	appropriate		
			and	data		
			appropriate	collection		
			methods of	strategy for		

			data	the specific		
			collection	issue		
General Object	tive 3.0: Gather	and analyse i	information to	identify potent	tial solutions a	nd reach a
conclusion		•		7 1		
Specific	Teachers	Learning	Specific	Teachers	Learning	Evaluation
Learning	Activities	Resources	Learning	Activities	Resources	
Objectives			Objectives:			
3.1. Collect	Support	Textbooks	Analyse the	Supervise	Textbooks	Final
information	students in	Notes	data	students in	Notes	Project
according to	data	Data	collected in	data	Data sheets	
agreed	collection	sheets	order to	analysis	Statistical	
methods.			reach a		packages	
			conclusion			
General Object	tive4.0:Produce	a final report	on the project			
Specific	Teachers	Learning	Specific	Teachers	Learning	Evaluation
Learning	Activities	Resources	Learning	Activities	Resources	
Objectives			Objectives:			
4.1. Produce	Guide	Internet	Students	Guide	Internet	
final written	students in	Notes	should	students in	Notes	
report on	the	Textbooks	produce a	the	Textbooks	
project	production		final report	production		
	of their		on their	of their		
	written		project	written		
	report			report		

PROGRAMME:	PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY									
COURSE TITLE:										
COURSE CODE: VLT										
DURATION	TION Lecture:- Hours Tutorial:- Practical:- Total:-									
					Hr/Wk(Hrs/Sem)					
CREDIT UNITS:	CU									
GOAL: This cour	rse is design	ed to								
GENERAL OBJE	CCTIVES:	On completion	of the course, the stud	ent should be able to:						

PROGRAMME: ND VETERINARY LABORATORY TECHNOLOGY								
COURSE TIT	TLE:		Cour	rse Code: VLT	Contac	t Hours:Hrs/W	k(Hrs/Sem)	
COURSE SPI content	ECIFICATION:	Theoretic	cal		P	ractical Conte	nt:	
General Obje	ctive: 1.0:							
Specific	Teachers	Learnin	ıg	Specific	Teachers	Learning	Evaluation	
Learning Objectives:	Activities	Resource		Learning Objectives:	Activities	Resources		
							Assignments	
							Tests	
							Quizzes	
							Examination	

General Obje	ctive: 2.0:					
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
Objectives.			Objectives.			Assignments
						Tests
						Quizzes
						Examination
General Obje	ctive: 3.0		<u>.</u>			<u> </u>
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
-						Assignments
						Tests
						Quizzes
						Examination
General Obje	ctive: 4.0				•	•
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
			3			Assignments
						Tests
						Quizzes
						Examination
General Obje						
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
y			y			Assignments
						Tests
						Quizzes
						Examination

General Obje	ctive: 6.0	•				•
Specific Learning Objectives:	Teachers Activities	Learning Resources	Specific Learning Objectives:	Teachers Activities	Learning Resources	Evaluation
			v			Assignments
						Tests
						Quizzes
						Examination

VETERINARY LABORATORY TECHNOLOGY PROGRAMME

LIST OF EQUIPMENT IN VETERINARY HOSPITAL

S/NO	DESCRIPTION OF ITEM	QUANTITY REQUIERED
1.	RECEPTION:	
	Reception chairs	8
	Table and chair	1 set
	File cabinet	1
		T
2.	CLINIC (SMALL AND LARGE ANIMAL)	3
	Digital weighing scale	2
	Deep Freezer	4
	Suction machine	3
	Stethoscope	9
	Endo- tracheal tube	5
	Elizebathal collar	5 sets
	Hoof trimmer	4 sets
	Orphthalmoscope	4
	Stomach tube	1
	Emasculator	8
	Dog catcher	5
	Vaginal speculum	2
	Small animal examination table	2
	Burdizzo	2
	Probang	1
	Drip stand	1
	Medication cabinet	2
	Autoclave	2
	Steam sterilizer	2
	Electric dehorner	1
	Gas dehorner	1
	Otoscope	3
	Obstetric box	1
	Knapsack sprayer	4
	Clinical thermometer	20
3	Surgical unit	
	Anesthetic machine	3
	Theatre lamp	1
	Thermo-cautery machine	2
	Small animal surgical box	2
	Large animal surgical box	1
	Post mortem box	2
4	Diagnostic laboratories	
	Haemo-analyzer	1
	Electrolyte Analyzer	1
	Chemical Analyzer	1
	Genotype machine	1

	Deep Freezer	1
	Refrigerator	1
	Incubator	4
	Binocular Microscope	2
	Refractometer	3
	Micro-haematocrit	2
	PH meter	4
	Haematospin	1
5	Radiology Laboratory	
	X-Ray machine	1
	X-Ray film dryer	1
	X-Ray film illuminator	2
	X-Ray film hanger	10
	X-Ray cabinet	1
	Dark room	1
	Endoscope cold light source	1
	Gastro scope	1
	Stabilizer	1
6	Pharmacy	
	Drug cabinet	2
	Refrigerator	1

LIST OF EQUIPMENTS IN PATHOLOGY LABORATORY

NECROPSY LABORATORY

S/No	Description	Quantity
1	Room roofed and well protected	1
2	Sustainable source of water	1
3	Working bench(stainless steel table)	2
4	Concrete floor well sloped for easy drainage and cleaning	1

5	Post mortem kits(knives, scissors, thumb forceps, bone cutter etc)	1
6	Protective clothing, masks, gloves.	6
7	Functional deep freezers	1
8	Source of light(bright)	4point
9	Glass slides and cover slips	10packs
10	Stains (Romanowsky)	4 liters
11	Forms and record books	50 forms,record book 1
12	Incinerator	1
13	Board for taking pictures	1
14	Camera	1
15	White board for writing interaction	1
16	Measuring cylinders (500ml,1000ml,2000ml)	3each
17	Butcher board	1
18	Stools (furniture) 10pieces	10pieces
19	Mechanized bone cutter (electric powered)	1

HISTOPATHOLOGY LABORATORY

S/N	Name of Equipment	Quantity
1	Fume cabinet	1
2	Automatic tissue processor	1
3	Embedding Centre	1
4	Microtome	1
5	Paratrimmer	1
6	Microtome blades (low grade)	1
7	Laboratory Microwave	1
8	Floating-out bath	1

9	Hot plate	1
10	Laboratory oven	1
11	Wax dispenser	1
12	Automatic strainer	1
13	Dissection kits	1
14	Dissection platform	1
15	Centrifuge (table top)	1
16	Refrigerator	1
17	Gas cylinder/burner/tripod stand	1 each
18	Microscope (optical binocular)	1
19	Microscope(camera-fixed)	1
20	Staining jars (set)	1
21	Autoclave	1
22	Hot air oven	1
23	Immunostiner	1
24	Projector	1

CLINICAL CHEMISTRY LABORATORY

S/N	Name of Equipment	Quantity
1	Auto chemistry analyzer	1
2	Auto haemoanalyser	1
3	Table top centrifuge	1
4	Microphaematocrit centrifuge	1
5	Spectrophotometer	1
6	Incubator	1
7	Hot air oven	1
8	Weighing balance	1
9	Water bath	1
10	Micropipette	10 packs
11	Autoclave	1

12	Test tubes (glass, plastic)	100 pieces
13	Test tube rack	1
14	Measuring cylinder	3
15	Refrigerator	1
16	Deep freezer	1
17	Microscope(optical binocular)	1
18	Microscope(camera-fixed)	2
19	Staining jars(set)	1
20	Pipette tips	5packs

LIST OF EQUIPMENT IN MICROBIOLGY LABORATORY

S/NO	ITEM/DESCRIPTION	QTY REQUIRED
A.	BACTERIOLOGICAL EQUIPMENTS	10
1	Microscope (Binocular)	2
2	Incubator	1
3	Hot Air oven	1
4	Water distiller	2
5	Autoclave	1
6	Weighing balance (electronic)	1
7	Hot plate	1
8	Colony counter	1
9	Refrigerator	1
10	Deep freezer	1
11	Anaerobic jar	1

B.	VIROLOGY EQUIPMENTS	1
1	Egg Candler	1
2	Co2 incubator	1
3	Well washer	1
4	Incubator shaker	1
5	Vice & clamp	1
6	Refrigerated centrifuge	1
7	ELISA Reader	1
8	Biosafety cabinet (CLASS II)	1
9	Trough	1
10	Tiles	1
11	Revco (-8°c)	1
12	Liquid Nitrogen Jar	1
13	Centrifuge	1

PARASTOLOGY LABORATORY

S/NO	ITEM/DESCRIPTION	QUANTITY REQUIRED
1	Microhaematocrit Centrifuge & Reader	1
2	Refrigerator	1
3	Bucket centrifuge	2
4	Hot plate	1
5	Microscope (Binocular)	10
6	CO2 Incubator	1
7	Dissecting microscope	1
8	Hot Air Oven	1
9	Incubator shaker	1
10	PH Meter	1
11	Water bath	1

TEAM LIST

ND/HND VETERINARY LABORATORY TECHNOLOGY

1	
•	Prof. S. O. Shoyinka
	University of Nigeria, Nsukka
2	Dr. John Garba
	Veterinary Council of Nigeria (VCN), Abuja
3	Prof. J. B. Adeyanju
	University of Ilorin
4	Prof. G. H. Sharubutu
	Fed. Coll. Of Animal Health and Production Technology, Vom
5	Dr. C. Bot
	Fed. Coll. Of Animal Health and Production Technology, Vom
6	Prof. M. D. Salihu
	College of Agriculture, Lafia
7	Prof. G. M. Agaie
	Usman Danfodiyo University, Sokoto
	NBTE Staff
8	Helen Oduntan
9	Dr.Rufai Ibrahim
10	Mrs. Kubura D. Jauro
11	Nsan Nsan Tiku